

	PATENT NO.	KIND	DATE	APPLICATION NO.	AGE
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11	WD 1059549	A1	10001115	WD 1000-USA 1000	1000-814
	W1:	AE, AF, AG, AH, AI, AJ, AK, AL, AM, AN, AO, AP, AQ, AR, AS, AT, AU, AV, AW, AX, AY, AZ, BA, BB, BC, BD, BE, BF, BG, BH, BI, BJ, BK, BL, BM, BN, BO, BP, BQ, BR, BS, BT, BU, BV, BW, BX, BY, BZ, CA, CB, CC, CD, CE, CF, CG, CH, CI, CJ, CK, CL, CM, CN, CO, CP, CQ, CR, CS, CT, CU, CV, CW, CX, CY, CZ, DA, DB, DC, DD, DE, DF, DG, DH, DI, DJ, DK, DL, DM, DN, DO, DP, DQ, DR, DS, DT, DU, DV, DW, DX, DY, DZ, EA, EB, EC, ED, EE, EF, EG, EH, EI, EJ, EK, EL, EM, EN, EO, EP, EQ, ER, ES, ET, EU, EV, EW, EX, EY, EZ, FA, FB, FC, FD, FE, FF, FG, FH, FI, FJ, FK, FL, FM, FN, FO, FP, FQ, FR, FS, FT, FU, FV, FW, FX, FY, FZ, GA, GB, GC, GD, GE, GF, GH, GI, GJ, GK, GL, GM, GN, GO, GP, GQ, GR, GS, GT, GU, GV, GW, GX, GY, GZ, HA, HB, HC, HD, HE, HF, HG, HH, HI, HJ, HK, HL, HM, HN, HO, HP, HQ, HR, HS, HT, HU, HV, HW, HX, HY, HZ, IA, IB, IC, ID, IE, IF, IG, IH, II, IJ, IK, IL, IM, IN, IO, IP, IQ, IR, IS, IT, IU, IV, IW, IX, IY, IZ, JA, JB, JC, JD, JE, JF, JG, JH, JI, JJ, JK, JL, JM, JN, JO, JP, JQ, JR, JS, JT, JU, JV, JW, JX, JY, JZ, KA, KB, KC, KD, KE, KF, KG, KH, KI, KJ, KK, KL, KM, KN, KO, KP, KQ, KR, KS, KT, KU, KV, KW, KX, KY, KZ, LA, LB, LC, LD, LE, LF, LG, LH, LI, LJ, LK, LL, LM, LN, LO, LP, LQ, LR, LS, LT, LU, LV, LW, LX, LY, LZ, MA, MB, MC, MD, ME, MF, MG, MH, MI, MJ, MK, ML, MM, MN, MO, MP, MQ, MR, MS, MT, MU, MV, MW, MX, MY, MZ, NA, NB, NC, ND, NE, NF, NG, NH, NI, NJ, NK, NL, NM, NO, NP, NQ, NR, NS, NT, NU, NV, NW, NX, NY, NZ, OA, OB, OC, OD, OE, OF, OG, OH, OI, OJ, OK, OL, OM, ON, OO, OP, OQ, OR, OS, OT, OU, OV, OW, OX, OY, OZ, PA, PB, PC, PD, PE, PF, PG, PH, PI, PJ, PK, PL, PM, PN, PO, PP, PQ, PR, PS, PT, PU, PV, PW, PX, PY, PZ, QA, QB, QC, QD, QE, QF, QG, QH, QI, QJ, QK, QL, QM, QN, QO, QP, QQ, QR, QS, QT, QU, QV, QW, QX, QY, QZ, RA, RB, RC, RD, RE, RF, RG, RH, RI, RJ, RK, RL, RM, RN, RO, RP, RQ, RR, RS, RT, RU, RV, RW, RX, RY, RZ, SA, SB, SC, SD, SE, SF, SG, SH, SI, SJ, SK, SL, SM, SN, SO, SP, SQ, SR, SS, ST, SU, SV, SW, SX, SY, SZ, TA, TB, TC, TD, TE, TF, TG, TH, TI, TJ, TK, TL, TM, TN, TO, TP, TQ, TR, TS, TU, TV, TW, TX, TY, TZ, UA, UB, UC, UD, UE, UF, UG, UH, UI, UJ, UK, UL, UM, UN, UO, UP, UQ, UR, US, UT, UU, UV, UW, UX, UY, UZ, VA, VB, VC, VD, VE, VF, VG, VH, VI, VJ, VK, VL, VM, VN, VO, VP, VQ, VR, VS, VT, VU, VV, VW, VX, VY, VZ, WA, WB, WC, WD, WE, WF, WG, WH, WI, WJ, WK, WL, WM, WN, WO, WP, WQ, WR, WS, WT, WU, WV, WW, WX, WY, WZ, XA, XB, XC, XD, XE, XF, XG, XH, XI, XJ, XK, XL, XM, XN, XO, XP, XQ, XR, XS, XT, XU, XV, XW, XX, XY, XZ, YA, YB, YC, YD, YE, YF, YG, YH, YI, YJ, YK, YL, YM, YN, YO, YP, YQ, YR, YS, YT, YU, YV, YW, YX, YY, YZ, ZA, ZB, ZC, ZD, ZE, ZF, ZG, ZH, ZI, ZJ, ZK, ZL, ZM, ZN, ZO, ZP, ZQ, ZR, ZS, ZT, ZU, ZV, ZW, ZX, ZY, ZZ			
	AW:	GH, GI, GJ, GK, GL, GM, GN, GO, GP, GQ, GR, GS, GT, GU, GV, GW, GX, GY, GZ, HA, HB, HC, HD, HE, HF, HG, HH, HI, HJ, HK, HL, HM, HN, HO, HP, HQ, HR, HS, HT, HU, HV, HW, HX, HY, HZ, IA, IB, IC, ID, IE, IF, IG, IH, II, IJ, IK, IL, IM, IN, IO, IP, IQ, IR, IS, IT, IU, IV, IW, IX, IY, IZ, JA, JB, JC, JD, JE, JF, JG, JH, JI, JJ, JK, JL, JM, JN, JO, JP, JQ, JR, JS, JT, JU, JV, JW, JX, JY, JZ, KA, KB, KC, KD, KE, KF, KG, KH, KI, KJ, KK, KL, KM, KN, KO, KP, KQ, KR, KS, KT, KU, KV, KW, KX, KY, KZ, LA, LB, LC, LD, LE, LF, LG, LH, LI, LJ, LK, LL, LM, LN, LO, LP, LQ, LR, LS, LT, LU, LV, LW, LX, LY, LZ, MA, MB, MC, MD, ME, MF, MG, MH, MI, MJ, MK, ML, MM, MN, MO, MP, MQ, MR, MS, MT, MU, MV, MW, MX, MY, MZ, NA, NB, NC, ND, NE, NF, NG, NH, NI, NJ, NK, NL, NM, NO, NP, NQ, NR, NS, NT, NU, NV, NW, NX, NY, NZ, OA, OB, OC, OD, OE, OF, OG, OH, OI, OJ, OK, OL, OM, ON, OO, OP, OQ, OR, OS, OT, OU, OV, OW, OX, OY, OZ, PA, PB, PC, PD, PE, PF, PG, PH, PI, PJ, PK, PL, PM, PN, PO, PP, PQ, PR, PS, PT, PU, PV, PW, PX, PY, PZ, QA, QB, QC, QD, QE, QF, QG, QH, QI, QJ, QK, QL, QM, QN, QO, QP, QQ, QR, QS, QT, QU, QV, QW, QX, QY, QZ, RA, RB, RC, RD, RE, RF, RG, RH, RI, RJ, RK, RL, RM, RN, RO, RP, RQ, RR, RS, RT, RU, RV, RW, RX, RY, RZ, SA, SB, SC, SD, SE, SF, SG, SH, SI, SJ, SK, SL, SM, SN, SO, SP, SQ, SR, SS, ST, SU, SV, SW, SX, SY, SZ, TA, TB, TC, TD, TE, TF, TG, TH, TI, TJ, TK, TL, TM, TN, TO, TP, TQ, TR, TS, TU, TV, TW, TX, TY, TZ, UA, UB, UC, UD, UE, UF, UG, UH, UI, UJ, UK, UL, UM, UN, UO, UP, UQ, UR, US, UT, UU, UV, UW, UX, UY, UZ, VA, VB, VC, VD, VE, VF, VG, VH, VI, VJ, VK, VL, VM, VN, VO, VP, VQ, VR, VS, VT, VU, VV, VW, VX, VY, VZ, WA, WB, WC, WD, WE, WF, WG, WH, WI, WJ, WK, WL, WM, WN, WO, WP, WQ, WR, WS, WT, WU, WV, WW, WX, WY, WZ, XA, XB, XC, XD, XE, XF, XG, XH, XI, XJ, XK, XL, XM, XN, XO, XP, XQ, XR, XS, XT, XU, XV, XW, XX, XY, XZ, YA, YB, YC, YD, YE, YF, YG, YH, YI, YJ, YK, YL, YM, YN, YO, YP, YQ, YR, YS, YT, YU, YV, YW, YX, YY, YZ, ZA, ZB, ZC, ZD, ZE, ZF, ZG, ZH, ZI, ZJ, ZK, ZL, ZM, ZN, ZO, ZP, ZQ, ZR, ZS, ZT, ZU, ZV, ZW, ZX, ZY, ZZ			

The present invention relates to sustained-release formulations using biodegradable **alginate** delayed gels or particles and leptin. Leptin (1.0 mg/mL; 10 mM Tris HCl, pH 8.8; pH adjusted from 9.0 to 8.6 with 1M NaOH) and 6% Et **alginate** (15 mol%, 10 mM Tris HCl, pH 8.6) were cooled on an ice bath. Leptin (0.5 mL) was added to the 6% Et **ester alginate** (0.19 mL) and the mixt. stirred on an ice bath for 10-15 min; the final pH was 8.6-8.8. To this mixt. was added a suspension of 1M CaCO₃ (16 μmol/L) and the resulting dispersion was mixed well. To this suspension was dropwise added, without stirring, a soln. of 0.1M ZnCl₂ (100 μmol/L); water was then added to bring the vol. to 1 mL. Then a soln. of 1.0M 6-glucancharotone (50 μmol/L) was thoroughly stirred into this mixt. The final mixt. (50 mg/mL leptin, 6% Et **alginate**; 0.1 mL) was cast on the inside of an Eppendorf tube and left overnight at 4 degree. to gel. After overnight storage, the **in vitro** release was conducted in 10 mM histidine buffer, pH 7.4. The gel cast with 15 mol% degree of **esterification** had minimal burst and fairly constant leptin release showing 60% released in 6 days. The gel cast with 10 mol% degree of **esterification** had minimal burst and fairly constant leptin release showing 55% released in 6 days.

1993, 17, 19, 21, 23, 25, 27, 29, 31, 33, 35, 37, 39, 41, 43, 45, 47, 49, 51, 53, 55, 57, 59, 61, 63, 65, 67, 69, 71, 73, 75, 77, 79, 81, 83, 85, 87, 89, 91, 93, 95, 97, 99, 101, 103, 105, 107, 109, 111, 113, 115, 117, 119, 121, 123, 125, 127, 129, 131, 133, 135, 137, 139, 141, 143, 145, 147, 149, 151, 153, 155, 157, 159, 161, 163, 165, 167, 169, 171, 173, 175, 177, 179, 181, 183, 185, 187, 189, 191, 193, 195, 197, 199, 201, 203, 205, 207, 209, 211, 213, 215, 217, 219, 221, 223, 225, 227, 229, 231, 233, 235, 237, 239, 241, 243, 245, 247, 249, 251, 253, 255, 257, 259, 261, 263, 265, 267, 269, 271, 273, 275, 277, 279, 281, 283, 285, 287, 289, 291, 293, 295, 297, 299, 301, 303, 305, 307, 309, 311, 313, 315, 317, 319, 321, 323, 325, 327, 329, 331, 333, 335, 337, 339, 341, 343, 345, 347, 349, 351, 353, 355, 357, 359, 361, 363, 365, 367, 369, 371, 373, 375, 377, 379, 381, 383, 385, 387, 389, 391, 393, 395, 397, 399, 401, 403, 405, 407, 409, 411, 413, 415, 417, 419, 421, 423, 425, 427, 429, 431, 433, 435, 437, 439, 441, 443, 445, 447, 449, 451, 453, 455, 457, 459, 461, 463, 465, 467, 469, 471, 473, 475, 477, 479, 481, 483, 485, 487, 489, 491, 493, 495, 497, 499, 501, 503, 505, 507, 509, 511, 513, 515, 517, 519, 521, 523, 525, 527, 529, 531, 533, 535, 537, 539, 541, 543, 545, 547, 549, 551, 553, 555, 557, 559, 561, 563, 565, 567, 569, 571, 573, 575, 577, 579, 581, 583, 585, 587, 589, 591, 593, 595, 597, 599, 601, 603, 605, 607, 609, 611, 613, 615, 617, 619, 621, 623, 625, 627, 629, 631, 633, 635, 637, 639, 641, 643, 645, 647, 649, 651, 653, 655, 657, 659, 661, 663, 665, 667, 669, 671, 673, 675, 677, 679, 681, 683, 685, 687, 689, 691, 693, 695, 697, 699, 701, 703, 705, 707, 709, 711, 713, 715, 717, 719, 721, 723, 725, 727, 729, 731, 733, 735, 737, 739, 741, 743, 745, 747, 749, 751, 753, 755, 757, 759, 761, 763, 765, 767, 769, 771, 773, 775, 777, 779, 781, 783, 785, 787, 789, 791, 793, 795, 797, 799, 801, 803, 805, 807, 809, 811, 813, 815, 817, 819, 821, 823, 825, 827, 829, 831, 833, 835, 837, 839, 841, 843, 845, 847, 849, 851, 853, 855, 857, 859, 861, 863, 865, 867, 869, 871, 873, 875, 877, 879, 881, 883, 885, 887, 889, 891, 893, 895, 897, 899, 901, 903, 905, 907, 909, 911, 913, 915, 917, 919, 921, 923, 925, 927, 929, 931, 933, 935, 937, 939, 941, 943, 945, 947, 949, 951, 953, 955, 957, 959, 961, 963, 965, 967, 969, 971, 973, 975, 977, 979, 981, 983, 985, 987, 989, 991, 993, 995, 997, 999, 1001, 1003, 1005, 1007, 1009, 1011, 1013, 1015, 1017, 1019, 1021, 1023, 1025, 1027, 1029, 1031, 1033, 1035, 1037, 1039, 1041, 1043, 1045, 1047, 1049, 1051, 1053, 1055, 1057, 1059, 1061, 1063, 1065, 1067, 1069, 1071, 1073, 1075, 1077, 1079, 1081, 1083, 1085, 1087, 1089, 1091, 1093, 1095, 1097, 1099, 1101, 1103, 1105, 1107, 1109, 1111, 1113, 1115, 1117, 1119, 1121, 1123, 1125, 1127, 1129, 1131, 1133, 1135, 1137, 1139, 1141, 1143, 1145, 1147, 1149, 1151, 1153, 1155, 1157, 1159, 1161, 1163, 1165, 1167, 1169, 1171, 1173, 1175, 1177, 1179, 1181, 1183, 1185, 1187, 1189, 1191, 1193, 1195, 1197, 1199, 1201, 1203, 1205, 1207, 1209, 1211, 1213, 1215, 1217, 1219, 1221, 1223, 1225, 1227, 1229, 1231, 1233, 1235, 1237, 1239, 1241, 1243, 1245, 1247, 1249, 1251, 1253, 1255, 1257, 1259, 1261, 1263, 1265, 1267, 1269, 1271, 1273, 1275, 1277, 1279, 1281, 1283, 1285, 1287, 1289, 1291, 1293, 1295, 1297, 1299, 1301, 1303, 1305, 1307, 1309, 1311, 1313, 1315, 1317, 1319, 1321, 1323, 1325, 1327, 1329, 1331, 1333, 1335, 1337, 1339, 1341, 1343, 1345, 1347, 1349, 1351, 1353, 1355, 1357, 1359, 1361, 1363, 1365, 1367, 1369, 1371, 1373, 1375, 1377, 1379, 1381, 1383, 1385, 1387, 1389, 1391, 1393, 1395, 1397, 1399, 1401, 1403, 1405, 1407, 1409, 1411, 1413, 1415, 1417, 1419, 1421, 1423, 1425, 1427, 1429, 1431, 1433, 1435, 1437, 1439, 1441, 1443, 1445, 1447, 1449, 1451, 1453, 1455, 1457, 1459, 1461, 1463, 1465, 1467, 1469, 1471, 1473, 1475, 1477, 1479, 1481, 1483, 1485, 1487, 1489, 1491, 1493, 1495, 1497, 1499, 1501, 1503, 1505, 1507, 1509, 1511, 1513, 1515, 1517, 1519, 1521, 1523, 1525, 1527, 1529, 1531, 1533, 1535, 1537, 1539, 1541, 1543, 1545, 1547, 1549, 1551, 1553, 1555, 15

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14. ANSWER 2 OF 57 HCAELUS COPYRIGHT 1 ACC
 15. 100111708 HCAELUS
 16. 1011116-4
 17. Microbicial and sanitizing soap compositions
 18. Lipes, John A.
 19. USA
 20. U.S., 1, pp., Cont. of U. S. Ser. No. 530,681, spaced 11.
 CODEN: USXAM
 21. Patent
 22. English
 23. WT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
US 5041474	A	1993-02-04	US 1997-01361	1997-01-04
US 1995-05361		1995-10-13		

24. The invention relates to microbicial and sanitizing soap compns. that incorporate agents with tuberculocidal properties in ready-to-use form that has gel properties or thixotropic properties and to soap compns. suitable for diln. in or with water or non-aq. solvent to produce gel-like or thixotropic solns. or dispersions ranging from free flowing to solidified forms. The ready-to-use compns. and the soln. compns. are applied for purposes of personal or animal hygiene or sanitizing on hair, hands and skin or other body parts, or are applied on inanimate surfaces and objects that need to be sanitized. For example, soap compn. contained Na C14-16 .alpha.-olefin sulfonates (40 % w/w), lactic acid (88 % w/w), xanthan gum 0.5, Aloe vera powder 0.1, lemon oil 0.1, and water b.s. to 100 %.

RE. WT 12

RE

01. Ambice; US 4545979 1985 HCAELUS
 (3) Anon; DE 3229097 1993 HCAELUS
 (4) Anon; GB 2216419 1994 HCAELUS
 (5) Brokken; US 4945110 1990 HCAELUS
 (7) Curtis; US 4213961 1980 HCAELUS
 ALL CITATIONS AVAILABLE IN THE RE FORMAT

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	PATENT NO.	CLASS	DATE	APPLICATION NO.	DATE
11	JP 1-16101	A1	199-12-14	JP 1997-1331	1997-04-14

[illegible]

	PATENT NO.	KIND	DATE	APPLICATION NO.	A.E.
FI	FR 1275667	A1	1996-01-09	FR 1996-13683	19-11-96
	FR 1275667	B1	1996-11-14		
	WO 97/01446	A1	1997-01-09		
DE: DE, SE, NO					

[illegible]

AP The title samples contain 1.00 g. 1.0* fixing polymer, 1.00 g. gelatin agent and has viscosity (40) 1044.4-1.61 cps (P) $\times V = 1.00 \times 1.00 = 1.00$ where P = total % concn. of fixing polymer and P = 1.00 g. 1.0* for P = 1.0* the viscosity is 1.0445-1.11 cps (P) $\times V = 1.0445-1.11$ cps. The gels provide rapid styling and hold as well as good wetting properties such as shine and a natural feel. Thus, a styling gel was prepd. from vinyl acetate-vinyl p-tert-butylbenzoate-acrylonitrile fixing polymer 2-amino-2-methyl-1-propanol salt (fixing polymer) and hydroxypropyl guar gum (Jamar HF60) in water. The viscosity of the gel was 100 cps. The gel, applied to hair with hands, dried rapidly, and imparted good hold and feel. The hair also exhibited good shine and a natural feel. Hair styled with gels based on the same components, but having a viscosity outside the range defined, was wettened by the gel and the hair style was flattened.

[illegible][illegible]

	PARENT N.O.	FIND	DATE	APPLICATION N.O.	AGE
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11	10-11-17	A	1-15-51	10-100-101	1-1-51
AB	The disinfectants, which are water-soluble and active, are made with suitable salts, comprise fatty acids, esters, and alcohols, glycols, natural polysaccharides and in their series, fatty acids, esters , and lower alcohols and disinfectants, in which the amount of the lower alcohols is 1-5 wt.%. A pick-irrigation fluid was prepared from benzalkonium chloride 0.5, benzalkonium chloride 0.5, xanthan gum 0.5, isopropyl adipate 0.3, triethanolamine 1.0, EtOH 10.0, and H ₂ O 56.4 g.				

1994-1995

1. COMPANY: BROADWAY BROADCASTING CO. INC.
2. ADDRESS: 100 BROADWAY
3. CITY: NEW YORK
4. The company is presently called BROADWAY BROADCASTING CO. INC.
5. Delivery: 100 BROADWAY
6. Contact: Evan D. Matson, Terry L. Yellowhair, Inc.
7. Inter: Pharmaceutical Corp., USA
8. US, 4. EE. Mail-in-part of U.S. Ser. No. 7, 8, 9.
9. IDENT: USXXAM
10. Patent
11. Revision
12. Note

IDENT	NO.	POINT	DATE	AFFILIATION	NO.	AGE
10	US 574471	A	1994-431	US 1994-4444	19-41120	
	US 574472	A	1994-431	US 1994-4444	19-41120	
	W 574473	A1	1994-711	WO 1994-4444	19-41120	
	W 574474	CA, JE				
	RW: AT, BE, CH, DE, DK, ES, FR, GB, GR, IE, IT, LI, NL, SE					
	US 574475	TL	1994-811	JE 1994-8111	19-41120	
	AT 574476	E	1994-611	AT 1994-8111	19-41120	
	ES 574477	TS	1994-711	ES 1994-8111	19-41120	
	US 574478	A	1994-711	US 1994-7111	19-41120	
	W 574479	A1	1994-101	WO 1994-4444	19-41120	
	W 574480	CA, JE				
	RW: AT, BE, CH, DE, DK, ES, FR, GB, GR, IE, IT, LI, NL, SE					
	AT 574481	A1	1994-711	AT 1994-7111	19-41120	
	AT 574482	EL	1994-811			
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	EE 574484	A1	1994-811	EE 1994-8111	19-41120	
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	US 574485	A	1994-1120	US 1994-7633	19-41120	
	US 574486	A	1994-1203	US 1994-7633	19-41120	
	US 574487	A	1994-1203	US 1994-7633	19-41120	
	US 574488	A	1994-811	US 1994-1611	19-41120	
	US 574489	A	1994-811	US 1994-1611	19-41120	
	US 574490	A	1994-811	US 1994-1611	19-41120	
	US 574491	A	1994-811	US 1994-1611	19-41120	
	US 574492	A	1994-811	US 1994-1611	19-41120	
	US 574493	A	1994-811	US 1994-1611	19-41120	
	US 574494	A1	1994-1203	WO 1994-4444	19-41120	
	W 574495	CA, JE				
	RW: AT, BE, CH, DE, DK, ES, FR, GB, GR, IE, IT, LI, NL, SE					
	US 574496	A	1994-811	US 1994-3073	19-41120	
	US 574497	AA	1994-811	CA 1994-1177	19-41120	
	JE 574498	TL	1994-811	JE 1994-8111	19-41120	
	US 574499	A	1994-1103	US 1994-4646	19-41120	
	US 574500	A	1994-411	US 1994-1911	19-41120	
	US 574501	A	1994-114	US 1994-8687	19-41120	
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	US 574503	A	1994-116	US 1994-7517	19-41120	
	AT 574504	EL	1994-116	AT 1994-8687	19-41120	
	AT 574505	A1	1994-811	AT 1994-8687	19-41120	
	AT 574506	A1	1994-811	AT 1994-8687	19-41120	
	AT 574507	A1	1994-811	AT 1994-8687	19-41120	
	AT 574508	A1	1994-811	AT 1994-8687	19-41120	
	AT 574509	A1	1994-811	AT 1994-8687	19-41120	
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	US 574511	1994-1203				
	US 574512	1994-611				
	US 574513	1994-611				
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10-14-64-1096 10-11-67
10-14-64-1097 10-11-67
10-14-64-1098 10-11-67
10-14-64-1099 10-11-67
10-14-64-1100 10-11-67

44 Gas and gasless precast filled microspheres, and also provide novel
topical and s.c. delivery vehicles for various active ingredients,
including drugs and cosmetics. Gas and gasless precast filled
microspheres were prepd. from dipalmitoylphosphatidyl choline.

SEARCHED BY JUDITH HANLEY 10-14-64

10-14-64

144-144-11

144 ANSWER 11: 1967: HAWAIIAN PATENT CO. A.W.
 145 1974: HAWAIIAN PATENT CO.
 146 1974: HAWAIIAN PATENT CO.
 147 Manufacture of stable emulsion containing 10% concentration
 and 10% concentration
 148 Matsubara, Shunji; Hattori, Tetsuo
 149 Asahi Chemical Ind., Japan
 150 Japan Patent Office, Tokyo, 4 pp.
 151 JPN. PAT. NO. 200,000
 152 Japan
 153 Japan
 154 JPN. PAT. NO. 200,000

154 JPN. PAT. NO. 200,000

PATENT NO.	POINT DATE	AFFILIATION NO.	A.W.
155 JPN. PAT. NO. 200,000	1967.6.10	1965-11-11	1965-11-11

156 Title: emulsion emulsions are manufd. by mixing and emulsifying oily phase
 contg. cationic surfactants and aq. phases contg. oil sol. materials.

Stearyltrimethylammonium chloride 0.6 g was dissolved in an oily
 phase contg. distearyl alc. 1.0, glycerin pyr. 1.0, lauric ester 1.0,
 glyceryl monostearate 1.0, polyoxyethylene monostearate 1.0, and ethylene
 glycol distearate 1.0. Then, the oily phase was emulsified with an aq.
 phase contg. poly aspartic acid Na 4.0, glycerin 1.0, sodium lauryl sulfate 1.4,
 and H₂O 165.0 and stirred with 1.0 g methylpolyethyl alcohol to give a rinse,
 which was stable at 15 degree for 1 wk without sep.

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[illegible]

PATENT NO.		PIND	DATE	APPLICATION NO.		DATE
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11	WO 97/1629	AL	1997-116	WO 1998-01164	1998-016	
	WI: AL, AM, AT, AU, AD, BF, BJ, BR, BY, CA, CH, C, CI, CR, CY, DE, EG, FI, FR, GB, GR, HU, IL, IN, JP, KE, KG, KH, KR, KZ, LA, LB, LG, LI, LU, LV, MC, MD, ME, MG, MP, MX, MY, NZ, NL, NO, PE, PL, PT, RO, RU, SE, SG, SI, SK, SL, SM, SN, ST, SV, SW, SZ, TH, TJ, TR, TT, UA, UG, US, UZ, VC, VE, VN, YU, ZA, ZW					
	FI: BE, BG, BR, CA, CH, CL, CO, AT, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IT, JP, KR, NL, NO, PT, SE, SG, SI, SK, SL, SM, SN, ST, SV, SW, TH, TJ, TR, TT, UA, UG, US, UZ, VC, VE, VN, YU, ZA, ZW					
	AL 1997-116		1997-116	AL 1998-01164	1998-016	
	FI 1997-116		1997-116	FI 1998-01164	1998-016	
	FI: BE, BG, BR, CA, CH, CL, CO, AT, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IT, JP, KR, NL, NO, PT, SE, SG, SI, SK, SL, SM, SN, ST, SV, SW, TH, TJ, TR, TT, UA, UG, US, UZ, VC, VE, VN, YU, ZA, ZW					
1998-016	1998-016		1998-016			
	WO 1998-01164		1998-016			

A cellulolytic enzyme prepn. comprising a cellulase with reduced stability is prepd., e.g., by increasing the mol. wt. or apparent size of the cellulase protein mol. or by insolubilizing or immobilizing the cellulase. The cellulase component may be immobilized by incorporation into a gel, by the formation of stable or temporary aggregates with a pol. mol. mass, by rapid immobilization of cellulase protein on insol. supports, by rapid autoimmobilization of the cellulase protein, or by adsorption on an insol. or sol. carrier. The carrier is preferably a cell base-contg. carrier of fibrous, microcryst., or amorphous structure, or more preferably a sol. or insol. polymer, e.g., a **polysaccharide** capable of interaction with the enzyme via a cellulase binding domain. The insol. support is a solid support, e.g., a sol. polyacrylonitrile copolymer. For example, Humicola insolens 48-kDa cellulase (1.6 mg) may be autoimmobilized on 100 g/L Avicel (microcryst. cellulose) by incubation in sodium phosphate buffer (0.05M, pH 8.5) at 20 degrees C for 30 min, repeated centrifugation at 4000 rpm for 15 min and 5 degree, draining the moist sediment, and milling. About 50% of the total cellulase is autoimmobilized by this procedure, and the immobilized cellulase retains full activity as "free" cellulase. The cellulase prepn. has a much lesser effect on influence on the durability or aging behavior of the cellulose substrate than corresponding unmodified cellulases, and at least having as good an effect on the look or feel, when used for treatment of cellulose fabrics or textiles. The cellulase prepn. may be used for domestic or industrial laundering or fabric softening as an ingredient of a detergent prepn., for bio-polishing, or for stone-wash or denim fabric treatments, or for other dyed fabric or garments.

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1. POLYMER IN FORM OF HYDROGEL OR HYDROGEL-LIKE GEL
 2. IN FORM OF HYDROGEL
 3. IN FORM OF HYDROGEL
 4. A polysaccharide hydrogel material, its preparation, its use
 in medicine, surgery, dentistry and for the preparation of dental
 products
 5. Bellini, Lucio; Callera, Lucio
 6. Bellini Advanced Polymers S.p.A., Italy; Bellini, Lucio; Callera,
 Lucio

7. Int. Appl., 1966
 8. INT. APPL.

9. INT. APPL.

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$$a = \frac{1}{2} \left(\frac{1}{\sqrt{2}} + \frac{1}{\sqrt{2}} \right) = \frac{1}{\sqrt{2}} \quad \text{and} \quad b = \frac{1}{2} \left(\frac{1}{\sqrt{2}} - \frac{1}{\sqrt{2}} \right) = 0$$

14. ANSWER: 14-1F, HAWAIIAN NIGHTSHADE, LILI
 15. 1408-1778, HAWAIIAN
 16. 1408-1778
 17. Description: A perennial plant with a woody stem and a dense, upright, branched habit. The leaves are alternate, elliptical, and have a smooth margin. The flowers are small and tubular, and the fruit is a small, round, red berry.
 18. Name: 1408-1778, HAWAIIAN
 19. Locality: Hawaii, Eastern Region, Puna District, Puna District, HI, 1408-1778, USA
 20. Date: 1408-1778, 1408-1778, 1408-1778
 21. Number: 1408-1778, 1408-1778
 22. Name: 1408-1778, 1408-1778
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 99. Name: 1408-1778, 1408-1778
 100. Name: 1408-1778, 1408-1778

PATENT NO.	KIND	DATE	APPLICATION NO.	CLASS
NO. 41,868	A1	1996-01-01	NO. 1994-06114	11-41-6
NO. 42,000, 42,001, 42,002	A1	1996-01-01	NO. 1994-06114	11-41-6
NO. 42,003, 42,004, 42,005, 42,006, 42,007, 42,008, 42,009, 42,010, 42,011, 42,012, 42,013, 42,014, 42,015, 42,016, 42,017, 42,018, 42,019, 42,020, 42,021, 42,022, 42,023, 42,024, 42,025, 42,026, 42,027, 42,028, 42,029, 42,030, 42,031, 42,032, 42,033, 42,034, 42,035, 42,036, 42,037, 42,038, 42,039, 42,040, 42,041, 42,042, 42,043, 42,044, 42,045, 42,046, 42,047, 42,048, 42,049, 42,050, 42,051, 42,052, 42,053, 42,054, 42,055, 42,056, 42,057, 42,058, 42,059, 42,060, 42,061, 42,062, 42,063, 42,064, 42,065, 42,066, 42,067, 42,068, 42,069, 42,070, 42,071, 42,072, 42,073, 42,074, 42,075, 42,076, 42,077, 42,078, 42,079, 42,080, 42,081, 42,082, 42,083, 42,084, 42,085, 42,086, 42,087, 42,088, 42,089, 42,090, 42,091, 42,092, 42,093, 42,094, 42,095, 42,096, 42,097, 42,098, 42,099, 42,100, 42,101, 42,102, 42,103, 42,104, 42,105, 42,106, 42,107, 42,108, 42,109, 42,110, 42,111, 42,112, 42,113, 42,114, 42,115, 42,116, 42,117, 42,118, 42,119, 42,120, 42,121, 42,122, 42,123, 42,124, 42,125, 42,126, 42,127, 42,128, 42,129, 42,130, 42,131, 42,132, 42,133, 42,134, 42,135, 42,136, 42,137, 42,138, 42,139, 42,140, 42,141, 42,142, 42,143, 42,144, 42,145, 42,146, 42,147, 42,148, 42,149, 42,150, 42,151, 42,152, 42,153, 42,154, 42,155, 42,156, 42,157, 42,158, 42,159, 42,160, 42,161, 42,162, 42,163, 42,164, 42,165, 42,166, 42,167, 42,168, 42,169, 42,170, 42,171, 42,172, 42,173, 42,174, 42,175, 42,176, 42,177, 42,178, 42,179, 42,180, 42,181, 42,182, 42,183, 42,184, 42,185, 42,186, 42,187, 42,188, 42,189, 42,190, 42,191, 42,192, 42,193, 42,194, 42,195, 42,196, 42,197, 42,198, 42,199, 42,200, 42,201, 42,202, 42,203, 42,204, 42,205, 42,206, 42,207, 42,208, 42,209, 42,210, 42,211, 42,212, 42,213, 42,214, 42,215, 42,216, 42,217, 42,218, 42,219, 42,220, 42,221, 42,222, 42,223, 42,224, 42,225, 42,226, 42,227, 42,228, 42,229, 42,230, 42,231, 42,232, 42,233, 42,234, 42,235, 42,236, 42,237, 42,238, 42,239, 42,240, 42,241, 42,242, 42,243, 42,244, 42,245, 42,246, 42,247, 42,248, 42,249, 42,250, 42,251, 42,252, 42,253, 42,254, 42,255, 42,256, 42,257, 42,258, 42,259, 42,260, 42,261, 42,262, 42,263, 42,264, 42,265, 42,266, 42,267, 42,268, 42,269, 42,270, 42,271, 42,272, 42,273, 42,274, 42,275, 42,276, 42,277, 42,278, 42,279, 42,280, 42,281, 42,282, 42,283, 42,284, 42,285, 42,286, 42,287, 42,288, 42,289, 42,290, 42,291, 42,292, 42,293, 42,294, 42,295, 42,296, 42,297, 42,298, 42,299, 42,300, 42,301, 42,302, 42,303, 42,304, 42,305, 42,306, 42,307, 42,308, 42,309, 42,310, 42,311, 42,312, 42,313, 42,314, 42,315, 42,316, 42,317, 42,318, 42,319, 42,320, 42,321, 42,322, 42,323, 42,324, 42,325, 42,326, 42,327, 42,328, 42,329, 42,330, 42,331, 42,332, 42,333, 42,334, 42,335, 42,336, 42,337, 42,338, 42,339, 42,340, 42,341, 42,342, 42,343, 42,344, 42,345, 42,346, 42,347, 42,348, 42,349, 42,350, 42,351, 42,352, 42,353, 42,354, 42,355, 42,356, 42,357, 42,358, 42,359, 42,360, 42,361, 42,362, 42,363, 42,364, 42,365, 42,366, 42,367, 42,368, 42,369, 42,370, 42,371, 42,372, 42,373, 42,374, 42,375, 42,376, 42,377, 42,378, 42,379, 42,380, 42,381, 42,382, 42,383, 42,384, 42,385, 42,386, 42,387, 42,388, 42,389, 42,390, 42,391, 42,392, 42,393, 42,394, 42,395, 42,396, 42,397, 42,398, 42,399, 42,400, 42,401, 42,402, 42,403, 42,404, 42,405, 42,406, 42,407, 42,408, 42,409, 42,410, 42,411, 42,412, 42,413, 42,414, 42,415, 42,416, 42,417, 42,418, 42,419, 42,420, 42,421, 42,422, 42,423, 42,424, 42,425, 42,426, 42,427, 42,428, 42,429, 42,430, 42,431, 42,432, 42,433, 42,434, 42,435, 42,436, 42,437, 42,438, 42,439, 42,440, 42,441, 42,442, 42,443, 42,444, 42,445, 42,446, 42,447, 42,448, 42,449, 42,450, 42,451, 42,452, 42,453, 42,454, 42,455, 42,456, 42,457, 42,458, 42,459, 42,460, 42,461, 42,462, 42,463, 42,464, 42,465, 42,466, 42,467, 42,468, 42,469, 42,470, 42,471, 42,472, 42,473, 42,474, 42,475, 42,476, 42,477, 42,478, 42,479, 42,480, 42,481, 42,482, 42,483, 42,484, 4				

A protein stabilizer additive comprises two or more of the group of the formula: $(\text{HOCH}_2)_n\text{-O-R}$, wherein R is: C1-4 alkyl; substituted C1-6 alkyl; HMD; NR₂R' wherein R and R' may be independent; H; C1-6 alkyl sulfonate; C1-6 hydroxyalkyl sulfonate; C1-6 alkyl $\text{SO}_3^-\text{CH}_2\text{CH}_2\text{OH}$; C1-6 alkyl; C1-6 hydroxyalkyl; C1-6 alkyl **carboxylate**; a polyelectrolyte; a buffer; and one or more additional components for example polyvalent metal salts. An example of the $(\text{HOCH}_2)_n\text{-O-R}$ group is 1,1,1'-tris(hydroxymethyl)ethane. An example of the polyelectrolyte is alginate acid. The stabilization of protein enzymes such as lactate oxidase and aldolase are described.

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11	84 01041	A	1984-01-11	84 1004-4	1984-10-11
	84 01042	A	1984-01-11		
	84 01043	A	1984-01-11		
	84 01044	A	1984-01-11		
	84 01045	A	1984-01-11		
	84 01046	A	1984-01-11		
	84 01047	A	1984-01-11		
	84 01048	A	1984-01-11		
	84 01049	A	1984-01-11		
	84 01050	A	1984-01-11		
	84 01051	A	1984-01-11		
	84 01052	A	1984-01-11		
	84 01053	A	1984-01-11		
	84 01054	A	1984-01-11		
	84 01055	A	1984-01-11		
	84 01056	A	1984-01-11		
	84 01057	A	1984-01-11		
	84 01058	A	1984-01-11		
	84 01059	A	1984-01-11		
	84 01060	A	1984-01-11		
	84 01061	A	1984-01-11		
	84 01062	A	1984-01-11		
	84 01063	A	1984-01-11		
	84 01064	A	1984-01-11		
	84 01065	A	1984-01-11		
	84 01066	A	1984-01-11		
	84 01067	A	1984-01-11		
	84 01068	A	1984-01-11		
	84 01069	A	1984-01-11		
	84 01070	A	1984-01-11		
	84 01071	A	1984-01-11		
	84 01072	A	1984-01-11		
	84 01073	A	1984-01-11		
	84 01074	A	1984-01-11		
	84 01075	A	1984-01-11		
	84 01076	A	1984-01-11		
	84 01077	A	1984-01-11		
	84 01078	A	1984-01-11		
	84 01079	A	1984-01-11		
	84 01080	A	1984-01-11		
	84 01081	A	1984-01-11		
	84 01082	A	1984-01-11		
	84 01083	A	1984-01-11		
	84 01084	A	1984-01-11		
	84 01085	A	1984-01-11		
	84 01086	A	1984-01-11		
	84 01087	A	1984-01-11		
	84 01088	A	1984-01-11		
	84 01089	A	1984-01-11		
	84 01090	A	1984-01-11		
	84 01091	A	1984-01-11		
	84 01092	A	1984-01-11		
	84 01093	A	1984-01-11		
	84 01094	A	1984-01-11		
	84 01095	A	1984-01-11		
	84 01096	A	1984-01-11		
	84 01097	A	1984-01-11		
	84 01098	A	1984-01-11		
	84 01099	A	1984-01-11		
	84 01100	A	1984-01-11		

AB Insulin-dependent diabetes mellitus is treated in a rat by prolonged administration of peptide "3" of pancreas-like peptide 1, insulinotropin, His^1 and related peptides, esp. in combination with a polymer matrix, in a water-insoluble oil suspension, in a complex with a polymer matrix, in a complex with a basic polypeptide, a phenolic resin, in a liposome delivery system, or after conjugation to compounds, e.g. in an rpm. or other material formation (e.g. high shear extrusion of salts to prolong the release of the peptide). Thus, a solution of 1 mg insulinotropin/ml phosphate-buffered saline (PBS) was mixed with an equal vol. of a soln. of 0.6 mg protamine and 4.4 mg Eudragit L (PBS) to produce an oil suspension.

[illegible][illegible]

11. $\int_0^1 (1-x^2)^{1/2} dx = \frac{\pi}{8}$ (containing water and ice) (1000) (1000) (1000)

IN *State, v. Taylor; People, v. McLaughlin; People, v. ...*

[illegible]

See also: *See also: "The Great Depression," 1929-1933*

...the

[illegible]2.3.3. *Carbone*

1998, 1999, 2000, 2001, 2002, 2003, 2004, 2005, 2006, 2007, 2008, 2009, 2010, 2011, 2012, 2013, 2014, 2015, 2016, 2017, 2018, 2019, 2020, 2021, 2022, 2023, 2024, 2025, 2026, 2027, 2028, 2029, 2030, 2031, 2032, 2033, 2034, 2035, 2036, 2037, 2038, 2039, 2040, 2041, 2042, 2043, 2044, 2045, 2046, 2047, 2048, 2049, 2050, 2051, 2052, 2053, 2054, 2055, 2056, 2057, 2058, 2059, 2060, 2061, 2062, 2063, 2064, 2065, 2066, 2067, 2068, 2069, 2070, 2071, 2072, 2073, 2074, 2075, 2076, 2077, 2078, 2079, 2080, 2081, 2082, 2083, 2084, 2085, 2086, 2087, 2088, 2089, 2090, 2091, 2092, 2093, 2094, 2095, 2096, 2097, 2098, 2099, 2100, 2101, 2102, 2103, 2104, 2105, 2106, 2107, 2108, 2109, 2110, 2111, 2112, 2113, 2114, 2115, 2116, 2117, 2118, 2119, 2120, 2121, 2122, 2123, 2124, 2125, 2126, 2127, 2128, 2129, 2130, 2131, 2132, 2133, 2134, 2135, 2136, 2137, 2138, 2139, 2140, 2141, 2142, 2143, 2144, 2145, 2146, 2147, 2148, 2149, 2150, 2151, 2152, 2153, 2154, 2155, 2156, 2157, 2158, 2159, 2160, 2161, 2162, 2163, 2164, 2165, 2166, 2167, 2168, 2169, 2170, 2171, 2172, 2173, 2174, 2175, 2176, 2177, 2178, 2179, 2180, 2181, 2182, 2183, 2184, 2185, 2186, 2187, 2188, 2189, 2190, 2191, 2192, 2193, 2194, 2195, 2196, 2197, 2198, 2199, 2200, 2201, 2202, 2203, 2204, 2205, 2206, 2207, 2208, 2209, 2210, 2211, 2212, 2213, 2214, 2215, 2216, 2217, 2218, 2219, 2220, 2221, 2222, 2223, 2224, 2225, 2226, 2227, 2228, 2229, 2230, 2231, 2232, 2233, 2234, 2235, 2236, 2237, 2238, 2239, 2240, 2241, 2242, 2243, 2244, 2245, 2246, 2247, 2248, 2249, 2250, 2251, 2252, 2253, 2254, 2255, 2256, 2257, 2258, 2259, 2260, 2261, 2262, 2263, 2264, 2265, 2266, 2267, 2268, 2269, 2270, 2271, 2272, 2273, 2274, 2275, 2276, 2277, 2278, 2279, 2280, 2281, 2282, 2283, 2284, 2285, 2286, 2287, 2288, 2289, 2290, 2291, 2292, 2293, 2294, 2295, 2296, 2297, 2298, 2299, 2300, 2301, 2302, 2303, 2304, 2305, 2306, 2307, 2308, 2309, 2310, 2311, 2312, 2313, 2314, 2315, 2316, 2317, 2318, 2319, 2320, 2321, 2322, 2323, 2324, 2325, 2326, 2327, 2328, 2329, 2330, 2331, 2332, 2333, 2334, 2335, 2336, 2337, 2338, 2339, 2340, 2341, 2342, 2343, 2344, 2345, 2346, 2347, 2348, 2349, 2350, 2351, 2352, 2353, 2354, 2355, 2356, 2357, 2358, 2359, 2360, 2361, 2362, 2363, 2364, 2365, 2366, 2367, 2368, 2369, 2370, 2371, 2372, 2373, 2374, 2375, 2376, 2377, 2378, 2379, 2380, 2381, 2382, 2383, 2384, 2385, 2386, 2387, 2388, 2389, 2390, 2391, 2392, 2393, 2394, 2395, 2396, 2397, 2398, 2399, 2400, 2401, 2402, 2403, 2404, 2405, 2406, 2407, 2408, 2409, 2410, 2411, 2412, 2413, 2414, 2415, 2416, 2417, 2418, 2419, 2420, 2421, 2422, 2423, 2424, 2425, 2426, 2427, 2428, 2429, 2430, 2431, 2432, 2433, 2434, 2435, 2436, 2437, 2438, 2439, 2440, 2441, 2442, 2443, 2444, 2445, 2446, 2447, 2448, 2449, 2450, 2451, 2452, 2453, 2454, 2455, 2456, 2457, 2458, 2459, 2460, 2461, 2462, 2463, 2464, 2465, 2466, 2467, 2468, 2469, 2470, 2471, 2472, 2473, 2474, 2475, 2476, 2477, 2478, 2479, 2480, 2481, 2482, 2483, 2484, 2485, 2486, 2487, 2488, 2489, 2490, 2491, 2492, 2493, 2494, 2495, 2496, 2497, 2498, 2499, 2500, 2501, 2502, 2503, 2504, 2505, 2506, 2507, 2508, 2509, 2510, 2511, 2512, 2513, 2514, 2515, 2516, 2517, 2518, 2519, 2520, 2521, 2522, 2523, 2524, 2525, 2526, 2527, 2528, 2529, 2530, 2531, 2532, 2533, 2534, 2535, 2536, 2537, 2538, 2539, 2540, 2541, 2542, 2543, 2544, 2545, 2546, 2547, 2548, 2549, 2550, 2551, 2552, 2553, 2554, 2555, 2556, 2557, 2558, 2559, 2560, 2561, 2562, 2563, 2564, 2565, 2566, 2567, 2568, 2569, 2570, 2571, 2572, 2573, 2574, 2575, 2576, 2577, 2578, 2579, 2580, 2581, 2582, 2583, 2584, 2585, 2586, 2587, 2588, 2589, 2590, 2591, 2592, 2593, 2594, 2595, 2596, 2597, 2598, 2599, 2600, 2601, 2602, 2603, 2604, 2605, 2606, 2607, 2608, 2609, 2610, 2611, 2612, 2613, 2614, 2615, 2616, 2617, 2618, 2619, 2620, 2621, 2622, 2623, 2624, 2625, 2626, 2627, 2628, 2629, 2630, 2631, 2632, 2633, 2634, 2635, 2636, 2637, 2638, 2639, 2640, 2641, 2642, 2643, 2644, 2645, 2646, 2647, 2648, 2649, 2650, 2651, 2652, 2653, 2654, 2655, 2656, 2657, 2658, 2659, 2660, 2661, 2662, 2663, 2664, 2665, 2666, 2667, 2668, 2669, 2670, 2671, 2672, 2673, 2674, 2675, 2676, 2677, 2678, 2679, 26

[illegible]

1. *Journal of the American Medical Association*, 1994; 271: 1039-1044.

Water-insoluble particles having poly-ionic type surface are used for immunoassay to reduce non-specific interaction. The surface of polyelectrolyte can be selected from anionic (carboxyl, sulfonate, phosphate, etc. functional group-contg. polyelectrolyte or cationic (quaternary ammonium, etc. functional group-contg. polyelectrolyte). In example, different combination of anionic and cationic polyelectrolyte were used as carrier for immunoassay of α -fetoprotein or α -glucosid-antitrypsin.

1. *Journal of the American Medical Association*, 1997; 277: 1033-1037.

AB: Biomaterials are disclosed which are comprised of: woven fabric, nonwoven fabric, and nonstretchable nonwoven fabric materials. The nonwoven fabric materials are comprised of threads imbedded in a matrix; both matrix and threads can be comprised of hyaluronic acid esters, singly or in combination with esters of alginic acid and other polymers. The fabric can be used for treating skin patches, surgery, etc. Preparation of a variety of hyaluronic acid esters is described, as is manufacture of the fabric. A nonwoven fabric of hyaluronic acid rennet ester was impregnated with vancomycin.

	PATENT NO.	WIND	DATE	APPLICATION NO.	DATE
11	W 100-410	A	1904-41	W 100-411	1904-414
	W 100-411	A	1904-41	W 100-412	1904-414
	W 100-412	A	1904-41	W 100-413	1904-414
	W 100-413	A	1904-41	W 100-414	1904-414
	W 100-414	A	1904-41	W 100-415	1904-414
	W 100-415	A	1904-41	W 100-416	1904-414
	W 100-416	A	1904-41	W 100-417	1904-414
	W 100-417	A	1904-41	W 100-418	1904-414
	W 100-418	A	1904-41	W 100-419	1904-414
	W 100-419	A	1904-41	W 100-420	1904-414

AB **Uronic esters** of acidic polysaccharides, such as hyaluronic acid, **alginate** acid, and CM cellulose, are effective against inhibitors and gastrointestinal agents. **Alginic acid** **uronic ester**. It was orally administered to rats for mucoprotein-inhibiting, gastrointestinal activity of 1.4 mg/kg body-weight and its efficacy was greater than that of sulfalate. 4.4 mg/kg to mix with water before use comprised granules comp. 1.4 mg/kg, 0.1 mg/kg and Na CMC 451, colloidal silica 10, talc 30, aspartame 20, flavor 10, and sucrose to 150 mg.

$$\frac{a}{b} = \frac{c}{d} \iff \frac{a}{c} = \frac{b}{d} \iff \frac{a}{b} = \frac{d}{c} \iff \frac{c}{d} = \frac{a}{b}$$
[illegible][illegible]

20 The title salts, water-sol. and lacking side effects, e.g., gastrointestinal and cell cytotoxicity, are (quaternary) ammonio-beta-hydroxyalkyl ethers of chitosan and new anionic, acceptable water-sol. polyanions.. In-vivo assays for tumors are required.

1000-144-11

1. ANSWER TO P. 4. RECAPITULATES DISCLOSURE OF P. 4.

2. 10-10-14-1 RECAPITULATES

3. 10-10-14-1

4. **Polysaccharide** gel formed in place5. **Barro, James R.**

6. Minnesota Mining and Manufacturing Co., USA

7. Expt. Pat. Appl., 6 pp.

8. CIPEN: EPXN10W

9. Patent

10. English

11. WT *

PATENT NO.	KIND	DATE	APPLICATION NO.	A. E.
EP 100-134	AL	1969-1-1	EP 100-134	10-119
EP 100-134	AL	1969-1-1		
EP 100-134	AL	1969-1-1		
AL: DE, FR, GB, NL				
US 3,678,686	A	1969-1-1	US 100-134	10-119
CA 1,067,111	AA	1969-1-1	CA 100-134	10-119
JP 1,140,184	AL	1969-1-1	JP 100-134	10-119

12. US 100-134 10-119

13. Framed and nonframed title gels are prepd. by mixing of COLH group-bearing **polysaccharides** contg. suspended di- or trivalent metal salts with aq. acids forming H₂O-sol. products with these salts, and metal ions which complex with COLH groups. Thus, a suspension of solid Na₂CO₃ in 0.6 g 4.5% aq. Na **alginate** is mixed with 0.5 g 0.5% AVOH in a double-barrel syringe assembly, and in-tube mixing tip to give a slightly framed, homogeneous gel approx. 1 min.

- | PATENT NO. | FIND | DATE | APPLICATION NO. | AGE |
|---------------|------|------------|-----------------|----------|
| BE 10-4-11-40 | AI | 1944-11-10 | BE 10-4-11-40 | 10-11-40 |
| BE 10-4-11-41 | AI | 1944-11-10 | BE 10-4-11-41 | 10-11-41 |
| BE 10-4-11-42 | AI | 1944-11-10 | BE 10-4-11-42 | 10-11-42 |
| BE 10-4-11-43 | AI | 1944-11-10 | BE 10-4-11-43 | 10-11-43 |
| BE 10-4-11-44 | AI | 1944-11-10 | BE 10-4-11-44 | 10-11-44 |
| BE 10-4-11-45 | AI | 1944-11-10 | BE 10-4-11-45 | 10-11-45 |
| BE 10-4-11-46 | AI | 1944-11-10 | BE 10-4-11-46 | 10-11-46 |
| BE 10-4-11-47 | AI | 1944-11-10 | BE 10-4-11-47 | 10-11-47 |
| BE 10-4-11-48 | AI | 1944-11-10 | BE 10-4-11-48 | 10-11-48 |
| BE 10-4-11-49 | AI | 1944-11-10 | BE 10-4-11-49 | 10-11-49 |
| BE 10-4-11-50 | AI | 1944-11-10 | BE 10-4-11-50 | 10-11-50 |
| BE 10-4-11-51 | AI | 1944-11-10 | BE 10-4-11-51 | 10-11-51 |
| BE 10-4-11-52 | AI | 1944-11-10 | BE 10-4-11-52 | 10-11-52 |
| BE 10-4-11-53 | AI | 1944-11-10 | BE 10-4-11-53 | 10-11-53 |
| BE 10-4-11-54 | AI | 1944-11-10 | BE 10-4-11-54 | 10-11-54 |
| BE 10-4-11-55 | AI | 1944-11-10 | BE 10-4-11-55 | 10-11-55 |
| BE 10-4-11-56 | AI | 1944-11-10 | BE 10-4-11-56 | 10-11-56 |
| BE 10-4-11-57 | AI | 1944-11-10 | BE 10-4-11-57 | 10-11-57 |
| BE 10-4-11-58 | AI | 1944-11-10 | BE 10-4-11-58 | 10-11-58 |
| BE 10-4-11-59 | AI | 1944-11-10 | BE 10-4-11-59 | 10-11-59 |
| BE 10-4-11-60 | AI | 1944-11-10 | BE 10-4-11-60 | 10-11-60 |
| BE 10-4-11-61 | AI | 1944-11-10 | BE 10-4-11-61 | 10-11-61 |
| BE 10-4-11-62 | AI | 1944-11-10 | BE 10-4-11-62 | 10-11-62 |
| BE 10-4-11-63 | AI | 1944-11-10 | BE 10-4-11-63 | 10-11-63 |
| BE 10-4-11-64 | AI | 1944-11-10 | BE 10-4-11-64 | 10-11-64 |
| BE 10-4-11-65 | AI | 1944-11-10 | BE 10-4-11-65 | 10-11-65 |
| BE 10-4-11-66 | AI | 1944-11-10 | BE 10-4-11-66 | 10-11-66 |
| BE 10-4-11-67 | AI | 1944-11-10 | BE 10-4-11-67 | 10-11-67 |
| BE 10-4-11-68 | AI | 1944-11-10 | BE 10-4-11-68 | 10-11-68 |
| BE 10-4-11-69 | AI | 1944-11-10 | BE 10-4-11-69 | 10-11-69 |
| BE 10-4-11-70 | AI | 1944-11-10 | BE 10-4-11-70 | 10-11-70 |
| BE 10-4-11-71 | AI | 1944-11-10 | BE 10-4-11-71 | 10-11-71 |
| BE 10-4-11-72 | AI | 1944-11-10 | BE 10-4-11-72 | 10-11-72 |
| BE 10-4-11-73 | AI | 1944-11-10 | BE 10-4-11-73 | 10-11-73 |
| BE 10-4-11-74 | AI | 1944-11-10 | BE 10-4-11-74 | 10-11-74 |
| BE 10-4-11-75 | AI | 1944-11-10 | BE 10-4-11-75 | 10-11-75 |
| BE 10-4-11-76 | AI | 1944-11-10 | BE 10-4-11-76 | 10-11-76 |
| BE 10-4-11-77 | AI | 1944-11-10 | BE 10-4-11-77 | 10-11-77 |
| BE 10-4-11-78 | AI | 1944-11-10 | BE 10-4-11-78 | 10-11-78 |
| BE 10-4-11-79 | AI | 1944-11-10 | BE 10-4-11-79 | 10-11-79 |
| BE 10-4-11-80 | AI | 1944-11-10 | BE 10-4-11-80 | 10-11-80 |
| BE 10-4-11-81 | AI | 1944-11-10 | BE 10-4-11-81 | 10-11-81 |
| BE 10-4-11-82 | AI | 1944-11-10 | BE 10-4-11-82 | 10-11-82 |
| BE 10-4-11-83 | AI | 1944-11-10 | BE 10-4-11-83 | 10-11-83 |
| BE 10-4-11-84 | AI | 1944-11-10 | BE 10-4-11-84 | 10-11-84 |
| BE 10-4-11-85 | AI | 1944-11-10 | BE 10-4-11-85 | 10-11-85 |
| BE 10-4-11-86 | AI | 1944-11-10 | BE 10-4-11-86 | 10-11-86 |
| BE 10-4-11-87 | AI | 1944-11-10 | BE 10-4-11-87 | 10-11-87 |
| BE 10-4-11-88 | AI | 1944-11-10 | BE 10-4-11-88 | 10-11-88 |
| BE 10-4-11-89 | AI | 1944-11-10 | BE 10-4-11-89 | 10-11-89 |
| BE 10-4-11-90 | AI | 1944- | | |

1841 AT 1000-1850 1000-1014
 Ak A maleic anhydride (II) polymer is **esterified** with 1 g I and 1 g I
 alc. II and to a **polysaccharide** to convert 1.5 g I to 1 g
 hydroxy groups to **ester** groups, the remainder being the free
 free acid, salt, and/or amide groups. The **esterification**
 product is useful as an absorbent for water, blood, urine, etc. A
 polymer was prep'd from 1.5 g I and 1 g Me vi d 1 alc. and
esterified 1.5 g alc. with 5 g II. Mow 1-1-1 in the
 presence of NaOH, giving an absorbent which had 17% acid groups
esterified and absorbed 434 vol H₂O and 42 vol Me vi d 1 alc. with.

10-14-1954

1. KOLLEGE 4-19 1954 (10-14-1954) 10-14-1954

2. KOLLEGE 4-19 1954 (10-14-1954) 10-14-1954

3. KOLLEGE 4-19 1954 (10-14-1954) 10-14-1954

4. KOLLEGE 4-19 1954 (10-14-1954) 10-14-1954

5. KOLLEGE 4-19 1954 (10-14-1954) 10-14-1954

6. KOLLEGE 4-19 1954 (10-14-1954) 10-14-1954

7. KOLLEGE 4-19 1954 (10-14-1954) 10-14-1954

8. KOLLEGE 4-19 1954 (10-14-1954) 10-14-1954

9. KOLLEGE 4-19 1954 (10-14-1954) 10-14-1954

10. KOLLEGE 4-19 1954 (10-14-1954) 10-14-1954

11. KOLLEGE 4-19 1954 (10-14-1954) 10-14-1954

12. KOLLEGE 4-19 1954 (10-14-1954) 10-14-1954

PATENT NO.	PINT DATE	APPLICATION NO.	APP. DATE
10-14-1954	10-14-1954	10-14-1954	10-14-1954
10-14-1954	10-14-1954	10-14-1954	10-14-1954

13. KOLLEGE 4-19 1954 (10-14-1954) 10-14-1954

AB Capsule walls are formed by pptn. of anionic and cationic polyelectrolytes at their interface. The anionic polyelectrolytes are formed from **carboxylate** and/or **polysaccharide** and/or synthetic polymers, and the cationic polyelectrolytes include **ammonium** surfactants and/or dyes. The microcapsules are used for eqn. processes in preparative and anal. chem., pharm., and in pharmacy, medicine, and agricem. and food industries. Thus, 0.1 g Na cellulose sulfate (D-5-11-5) with a degree of substitution of 1.4 was dissolved in 0.5 l H₂O, and the soln. was pressed through a 0.5-mm inner diam. capillary and dropped from a height of 30 cm into a stirred bath contg. 1% aq. methylene blue (61-5-4). After 10 min capsules formed were decanted and washed with H₂O. The deep-blue capsules had a diam. of 4-5 mm.

122: 11, 12, 13, 14-15, 16, 17, 38-41

121: ANSWER 6 OF 45 USPATFULL

AN 1984:14-15 USPATFULL

TI Aqueous viscoelastic surfactant solutions for the treatment of hair and skin
IN Bader, Dieter, Herten, Germany, Federal Republic of
IA Bader Aktiengesellschaft, Marl, Germany, Federal Republic of Germany

121: 1984:14-15

121: 1984:14-15

121: 1984:14-15

121: Continuation of Ser. No. US 1978-441666, filed July 1, 1978, now abandoned

121: 1984:14-15

121: 1984:14-15

EXAM: Primary Examiner: Gupta, Yashendra; Assistant Examiner: Price, John R.

121: 1984:14-15

121: 1984:14-15

121: 1984:14-15

121: 1984:14-15

121: 1984:14-15

121: 1984:14-15

AB Aqueous viscoelastic surfactant solutions for the treatment of hair and skin which contain:

A. from 4 to 25% by weight of an anionic surfactant;

B. from 0.1 to 1% by weight of a cationic surfactant;

C. from 0.1 to 1% by weight of a nonionic surfactant;

D. from 0.1 to 6% by weight of an electrolyte;

E. from 0.1 to 5% by weight of a water-soluble polymer; and

F. from 0.1 to 5% by weight of a further constituent in which the sum of the amounts of A, B, and C is at least 1% by weight and the sum of the amounts of D, E, and F is between 1 and 10% by weight, in each case based on the total weight of the aqueous solution, and having a shear modulus, G' , between 5 and 50 Pa at temperatures between 20 and 40 degrees C. and a pH of from 4 to 8, and in which the conditions for the identity of the storage modulus, G' , and the loss modulus, G'' , are in the angular frequency range: ω from 1 and 6 rad.s⁻¹ exhibit optimum flow behavior for the intended applications.

121: 1984:14-15

121: ANSWER 11 OF 45 USPATFULL

AN 1984:14-15 USPATFULL

TI Preparation of lactams from **aliphatic** α,β -unsaturated nitriles

IN Di Cosimo, Robert, Rockland, DE, United States

121: 1984:14-15

121: 1984:14-15

121: 1984:14-15

121: 1984:14-15

121: 1984:14-15

121: 1984:14-15

121: 1984:14-15

121: 1984:14-15

EXAM: Primary Examiner: Lilling, Herbert J.

121: 1984:14-15

121: 1984:14-15

121: 1984:14-15

121: 1984:14-15

121: 1984:14-15

AB A process for the preparation of five-membered lactams prepared from **aliphatic** α,β -unsaturated nitriles, β,γ -unsaturated nitriles, and α,β,γ -unsaturated nitriles, in the presence of **aliphatic** α,β -unsaturated nitriles.

1 ANSWER 1. F 47 USPATFULL
AN 1991:12783 USPATFULL
TI Non-woven fabric material comprising auto-crosslinking electronic acid
derivatives
IN 1. Ligatti, Franco, Via Sersutini 13, 20138 Lecco, Italy
2. Ligatti, Gianfranco, Via Sersutini 38, 20138 Lecco, Italy
3. Ligatti, Aurelio, Viale Ippolito, 20, 20138 Lecco, Italy
FI US 6-24385 199-12
AI US 1990-4-74 7 1995-6 7 1995-6
ALL Continuation-in-part of Ser. No. US 1991-00700, filed on 18 Dec 1990,
now patented, Pat. No. US 5500616
FRA1 IT 1991-ED229 19911219
DT 1991
EXAM Primary Examiner: Wetman, Edward J.
LSE1 Birch, Stewart, Holasch & Birch, LLP
CLA1 Number of Claims: 1
ECL1 Exemplary Claim: 1
DRA1 1 Drawing Figure(s) ; 1 Drawing Page(s)
LSE1 1991

AB Biomaterials are disclosed comprised of biodegradable, absorbable, and bioabsorbable non-woven fabric materials for **oral surgery** for the guided regeneration of tissues. The non-woven fabric materials constitute threads embedded in a matrix, wherein the matrix and the threads constitute self-crosslinked hydrogels.

07. ANSWER 14 OF 45 USEPATEFULL
 AN 1998:6-38 USEPATEFULL
 TI Coated papers
 IN Malhotra, Shadi L., Mississauga, Canada
 EE Xerox Corporation, Stamford, CT, United States (U.S.) Corporation,
 FI US 57,9076 1998-11-11
 AI 02/1996-656-14 1996 6 1 14
 DT Utility
 EPCAT Priority Examiner: Schwartz, Pamela R.
 LISA Palocz, E. C.
 NAME Name of Owner: 01
 EPCAT Exemplary Name: 1
 LISA No Drawings
 LISA 01

[illegible]

[illegible]

1. *Journal of the American Medical Association*, 1990; 263: 1025-1028.

[illegible]

NO INDEXING IS AVAILABLE FOR THIS PATENT.

AB Inter- and/or intramolecular cross-linked esters of acid polysaccharides are disclosed in which a part or all of the carboxylic groups are esterified with hydroxyl groups of the same molecule and/or of different molecules of the acid polysaccharide. These inter- and/or cross-linked esters of polysaccharide acids are useful in the field of absorbable plastic materials, to reinforce sanitary and **surgical** articles, in the dental and pharmaceutical fields, in the food industry and in many other industrial fields.

THE UNIVERSITY OF CHICAGO PRESS

01 ANSWER 16 OF 48 USFATFULL
AN 1985 - USFATFULL
TI Composite membranes for the guided regeneration of nerves
IN Orziatti, Franco; Davis, Italy
Gallegani, Lanfranco; Forte di Brenta, Italy
Rome; Anelli, Rome, Italy
EA M.I.S.A.T. Italian Ministry for Universities and Scientific and
Technological Research, Rome, Italy non-U.S. Patent No.
FI 2,742,000-1985-11
AL 2,742,000-1985-11
PARI 2,742,000-1985-11
IT 2,742,000-1985-11
EXCH Primary Examiner: Weisman, Edward C.
1985 Birch, Stewart, Kohnen & Birch, LLP
NUM Number of Claims: 27
EPI Exemplary Claim: 1
PPN Drawing Figure no.; 3 Drawing Pages
INT INT

ACKNOWLEDGMENTS

AB Bi materials are also used comprised of nitrocellulose, polypropylene, and nitrocellulose composite membranes for use in surgery for the limited regeneration of tissues. The composite materials are comprised of fibers embedded in a matrix, where the matrix and the fibers are both comprised of nitrocellulose. The fibers are used primarily for suturing, and the matrix is used primarily for limited regeneration of tissues.

Abstract: **alginate** and other polymers.

NO INDEXING IS AVAILABLE FOR THIS PATENT.

1. ANSWER 17 OF 48. USPATFULL
 2. 36487-1 USPATFULL
 3. In regard to the purification of hyaluronic acid, a fraction of pure hyaluronic acid is obtained by:
 4. F.lli, Angelo, Rome, Italy
 5. F.lli, Angelo, Rome, Italy
 6. F.lli S.p.A., Anagnino, Italy non-U.S. application
 7. US 3650141 1968.514
 8. US 1968-48947 1968.511
 9. Continuation of Ser. No. US 1968-48947, filed on 11/1/68, now abandoned
 10. IT 1961-8117 1961.419
 11. Utility
 12. Primary Examiner: F.lli, Angelo; Assistant Examiner: F.lli, Matteo
 13. F.lli, Angelo, Rome, Italy
 14. Number of Claims: 15
 15. Exemplary Claim: 1
 16. No Drawings
 17. INT. CL. 1961

NO INDEXING IS AVAILABLE FOR THIS PATENT.

AB A highly pure fraction of hyaluronic acid is disclosed which is non-inflammatory and avoids post-operative complications in ocular surgery. Also disclosed is a process for the preparation of hyaluronic acid characterized by converting hyaluronic acid into a corresponding potassium ammonium salt and, following purification procedures, reconverts the potassium ammonium salt into a pure salt of hyaluronic acid.

NO INDEXING IS AVAILABLE FOR THIS PATENT.

1. ANSWER 18 OF 48. USPATFULL
 2. 36487-1 USPATFULL
 3. Non-woven fabric material comprising hyaluronic acid derivatives
 4. F.lli, Angelo, Trent, Italy
 5. F.lli, Angelo, Trent, Italy
 6. F.lli, Angelo, Trent, Italy
 7. M.I.S.T. (Italian Ministry for Universities and Scientific and Technological Research), Rome, Italy non-U.S. application
 8. US 3650141 1968.514
 9. US 1968-48947 1968.511
 10. IT 1961-8117 1961.419
 11. Utility
 12. Primary Examiner: Webman, Edward J.
 13. F.lli, Angelo, Trent, Italy
 14. Number of Claims: 14
 15. Exemplary Claim: 1
 16. 1 Drawing Figure 1; 1 Drawing Page 1
 17. INT. CL. 1961

NO INDEXING IS AVAILABLE FOR THIS PATENT.

AB Biomaterials are disclosed comprised of biodegradable, biocompatible, and strong stable nonwoven fabric materials for use in **surgery** for the guided regeneration of tissues. The non-woven fabric materials are comprised of threads embedded in a matrix, wherein the matrix and the threads can be comprised of esters of hyaluronic acid, used singly or in combination, or esters of hyaluronic acid in combination with esters of **alginate** and other polymers.

NO INDEXING IS AVAILABLE FOR THIS PATENT.

1. ANSWER 19 OF 48. USPATFULL
 2. 36487-1 USPATFULL
 3. Biodegradable esters
 4. F.lli, Angelo, Trent, Italy
 5. F.lli, Angelo, Trent, Italy
 6. F.lli S.p.A., Anagnino, Italy non-U.S. application
 7. US 3650141 1968.514
 8. US 1968-48947 1968.511
 9. IT 1961-8117 1961.419
 10. Utility
 11. Primary Examiner: Webman, Edward J.
 12. F.lli, Angelo, Trent, Italy
 13. Number of Claims: 14
 14. Exemplary Claim: 1
 15. 1 Drawing Figure 1; 1 Drawing Page 1
 16. INT. CL. 1961

11. 10-44-41 10-4114
 12. 10-44-41 10-4114
 13. Division of Ser. No. US 10-44-41, filed 10-41, now patented,
 Pat. No. 2,811,114
 14. 10-44-41 10-4114
 15. Utility
 16. Primary Examiner: Adams, Charles A.
 17. Brown, Stewart, Polson & Birch
 18. Number of Claims: 1
 19. Exemplary Claim: 1
 20. No Drawings
 21. WI 10-41
 22. New polyacrylamide esters are disclosed, which are poly-esters of
 acrylic polyacrylamides chosen from the group consisting of
 acrylamide, dimethylacrylamide, diethylacrylamide and hexamethylenediacrylamide.
 These new esters and some esters of the type also known are useful as
 reagents, for the manufacture of pharmaceutical and cosmetic
 preparations, in the field of irrepressible plastic materials and,
 in detail, for the manufacture of medical, **surgical** and
 dentistry articles, as well as other industrial uses in the
 place of existing polyacrylamides now in common use.

23. ANSWER 1 OF 48. USPATFULL
 24. 10-44-41 USPATFULL
 25. New esters of **alginic** acid
 26. della Valle, Francesco, Padua, Italy
 27. Rome, Aurelio, Rome, Italy
 28. Fila, S.p.A., Arona Terme, Italy non-U.S. application
 29. 10-44-41 10-4114
 30. 10-44-41 10-4114
 31. Utility
 32. Division of Ser. No. US 10-41-41, filed 10-41, now patented,
 Pat. No. US 2,811,114 which is a division of Ser. No. 10-41-41, filed
 10-41, now abandoned
 33. IT 10-41-41 10-4114
 34. Utility
 35. Primary Examiner: Robinson, Douglas W.; Assistant Examiner: Lee, Howard
 36. Brown, Stewart, Polson & Birch
 37. Number of Claims: 1
 38. Exemplary Claim: 1
 39. No Drawings
 40. WI 10-41

41. INDEXING IS AVAILABLE FOR THIS PATENT.
 42. Partial esters of **alginic** acid and salts thereof are
 important bioplastic and pharmaceutical preparations useful in
 various fields including medical, **surgical**, cosmetic and
 foods.

43. INDEXING IS AVAILABLE FOR THIS PATENT.

44. ANSWER 11 OF 48. USPATFULL
 45. 10-44-41 USPATFULL
 46. Esters of **alginic** acid
 47. della Valle, Francesco, Padua, Italy
 48. Rome, Aurelio, Rome, Italy
 49. Fila, S.p.A., Arona Terme, Italy non-U.S. application
 50. 10-44-41 10-4114
 51. 10-44-41 10-4114
 52. Utility
 53. Division of Ser. No. US 10-41-41, filed 10-41, now
 patented
 54. IT 10-41-41 10-4114
 55. Utility
 56. Primary Examiner: Griffin, Paul W.
 57. Brown, Stewart, Polson & Birch
 58. Number of Claims: 1
 59. Exemplary Claim: 1
 60. No Drawings

INDEXING IS AVAILABLE FOR THIS PATENT.

ABSTRACT: This invention relates to **alginate** and which is used in

impregnated plastic and pharmaceutical applications. It is useful in various fields including medical, **surgical**, and dental fields.

INDEXING IS AVAILABLE FOR THIS PATENT.

ANSWER 14 OF 45 USEPATFULL

ABSTRACT: This invention relates to

fractionated polysaccharide derivatives with hypotensive activity

INVENTOR: Monti, Franco, Milan, Italy

PA: Etablissement Technico, Liechtenstein, Italy (U.S. corporation)

FI: US 5,595,561 1991 11 11

AI: US 1990-08,117 1991 11 11

ABSTRACT: Continuation of Ser. No. US 1988-140114, filed in U.S. Pat. Office, now

abandoned

INVENTOR: IT 1988-11711 1989-11-11

IT: Utility

EXAMINER: Primary Examiner: Griffin, Ronald W.; Assistant Examiner: White, Everett

ABSTRACT: Burkhardt, Wendel & Rossi

INVENTOR: Number of Claims: 7

ABSTRACT: Exemplary Claim: 1

ABSTRACT: No Drawings

INVENT 486

INDEXING IS AVAILABLE FOR THIS PATENT.

ABSTRACT: Optimized derivatives of natural polysaccharides having a polyglucoside structure with 50-5000 monomer units and one or more side chains bonded to the glucoside nucleus by a nitrogen or oxygen atom, an amide group, and side chains having one or more quaternary nitrogen atoms so that each monomer unit has a cation charge density exceeding two. The new compounds are particularly active as hypotensive and diuretic agents.

INDEXING IS AVAILABLE FOR THIS PATENT.

ANSWER 17 OF 45 USEPATFULL

ABSTRACT: This invention relates to

highly purified composition for nerve cell degeneration repairing or protective agent and process for preparing a highly purified composition contained in the composition

INVENTOR: Tanaka, Tatsuyoshi, Tokushima, Japan

ABSTRACT: Sakurai, Yoji, Tokushima, Japan

ABSTRACT: Okudaki, Hiroshi, Tokushima, Japan

ABSTRACT: Hasegawa, Takashi, Tokushima, Japan

ABSTRACT: Fukuyama, Yoshiyasu, Tokushima, Japan

PA: Otsuka Pharmaceutical Company, Ltd., Tokyo, Japan (U.S. corporation)

FI: US 5,053,549 1991 11 01

AI: US 1990-476913 1990 02 09 (7)

ABSTRACT: JP 1989-199415 1990 02 08

ABSTRACT: JP 1989-199415 1990 11 16

IT: Utility

EXAMINER: Primary Examiner: Gerstl, Robert

ABSTRACT: Shukrue, Mitch, Dinn Macpeak & Seas

INVENTOR: Number of Claims: 16

ABSTRACT: Exemplary Claim: 1

ABSTRACT: 4 Drawing Figure 3; 4 Drawing Page 3

INVENT 1740

INDEXING IS AVAILABLE FOR THIS PATENT.

ABSTRACT: There is disclosed a biphenyl derivative of the formula: **##STR1##** wherein R₁, R₂, R₃, R₄, R₅ and R₆ are as defined or its salt, a composition for nerve cell degeneration repairing or protective agent containing a biphenyl derivative of the formula: **##STR2##** wherein R₁, R₂, R₃ and R₄ are as defined, and a process for preparing a biphenyl derivative contained in the composition.

INDEXING IS AVAILABLE FOR THIS PATENT.

ANSWER 18 OF 45 USEPATFULL

ABSTRACT: This invention relates to

11 Takaoka, Yoshio, agent.
 12 Miyake, Hisashi, 1-114-18, Hoshinokanemate, Miyake-cho, Miyake-gun,
 Japan.
 13 Imai, Hisayuki, Miyake-cho, Japan.
 14 Miyake, Hisashi, Miyake-cho, Japan.
 15 Imai, Hisayuki, Miyake-cho, Japan.
 16 Araki, Shigeo, Higashi-Furuta, Akita, Japan.
 17 Miyake, Hisashi, Miyake-cho, Japan.
 18 US 486-848 1978-11-14
 19 US 10-1-8-18 1978-11-14
 20 US 10-1-173669 1978-11-14
 21 US 10-1-8846 1978-11-14
 22 Utility
 EXAM Primary Examiner: Shapiro, Lionel M.
 23 Attorney: K. P. P.
 24 Number of Claims: 15
 25 Exemplary Claim: 1
 26 Drawing Figure 1; 2 Drawing Page 3
 27 INT. WT 4419

28 INDEXING IS AVAILABLE FOR THIS PATENT.

AB The invention discloses that the tissues of earthworms contain
 fibrinolytically or thrombolytically active ingredients which can be
 extracted and purified by a suitable sequence of extraction and
 purification procedures into the individual active ingredients including
 six novel proteases named F-O-HM-48, F-I-1-HM-54, F-II-1-HM-15,
 F-III-1-HM-54, F-III-1-HM-17 and F-III-1-HM-20. The invention describes
 fractionation of the earthworm extract with an aqueous extractant gives
 five active fractions, the first four of which contain one of the
 first mentioned four proteases and the last of which contains the last
 mentioned two proteases. The disclosure includes a definition of the
 suitable purification methods for the proteases as well as the
 physico-chemical identification data thereof. Various thrombolytic
 medicament forms prepared with the novel proteases and the effective
 ingredient are described together with the results of the clinical tests
 carried out by the oral administration of the novel proteases.

29 INDEXING IS AVAILABLE FOR THIS PATENT.

30 ANSWER 30 OF 48 USPTAFULL
 31 4487476 USPTAFULL
 32 Stabilization of thickened aqueous fluids
 33 Sandell, Lionel S., Hagerstown, MD, United States
 34 E. I. du Pont de Nemours and Company, Wilmington, DE, United States
 35 (U.S. Corporation)
 36 US 4486617 1982-11-24
 37 US 1090-415681 1982-09-07 (6)
 38 INT 4419
 39 Continuation-in-part of Ser. No. US 1081-128728, filed on 16 Jan. 1981,
 40 now patented, Pat. No. US 4390482
 41 Utility
 42 Primary Examiner: Gwynn, Herbert B.
 43 Number of Claims: 15
 44 Exemplary Claim: 1
 45 No Drawings
 46 INT 4419

47 INDEXING IS AVAILABLE FOR THIS PATENT.

AB Thickeners, preferably galactomannans, in aqueous solutions or
 slurries, e.g., water gel explosives, oil well drilling fluids,
 and hydraulic fracturing fluids, are stabilized against thermal
 degradation by incorporating iodide and/or iodate ions into the solution
 or slurry. The preferred stabilizer is the iodide and/or iodate ions
 the fluids by dissolving an alkali metal or alkali earth metal iodide,
 or ammonium or alkyl-substituted ammonium iodide, in an aqueous phase.

48 INDEXING IS AVAILABLE FOR THIS PATENT.

49 ANSWER 4 OF 48 USPTAFULL
 50 4487476 USPTAFULL
 51 Microcapsules containing viable tissue cells
 52 Lin, Franklin, Richmond, VA, United States

1A Daniel Corporation, Needham Heights, MA, United States, U.S. Corporation.
 11 US 4,414,414, 1-1-1981
 12 US 4,414,414, 1-1-1981
 13 US 4,414,414, 1-1-1981
 14 US 4,414,414, 1-1-1981
 15 US 4,414,414, 1-1-1981
 16 US 4,414,414, 1-1-1981
 17 US 4,414,414, 1-1-1981
 18 US 4,414,414, 1-1-1981
 19 US 4,414,414, 1-1-1981
 20 US 4,414,414, 1-1-1981
 21 US 4,414,414, 1-1-1981
 22 US 4,414,414, 1-1-1981
 23 US 4,414,414, 1-1-1981
 24 US 4,414,414, 1-1-1981
 25 US 4,414,414, 1-1-1981
 26 US 4,414,414, 1-1-1981
 27 US 4,414,414, 1-1-1981
 28 US 4,414,414, 1-1-1981
 29 US 4,414,414, 1-1-1981
 30 US 4,414,414, 1-1-1981
 31 US 4,414,414, 1-1-1981
 32 US 4,414,414, 1-1-1981
 33 US 4,414,414, 1-1-1981
 34 US 4,414,414, 1-1-1981
 35 US 4,414,414, 1-1-1981
 36 US 4,414,414, 1-1-1981
 37 US 4,414,414, 1-1-1981
 38 US 4,414,414, 1-1-1981
 39 US 4,414,414, 1-1-1981
 40 US 4,414,414, 1-1-1981
 41 US 4,414,414, 1-1-1981
 42 US 4,414,414, 1-1-1981
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 65 US 4,414,414, 1-1-1981
 66 US 4,414,414, 1-1-1981
 67 US 4,414,414, 1-1-1981
 68 US 4,414,414, 1-1-1981
 69 US 4,414,414, 1-1-1981
 70 US 4,414,414, 1-1-1981
 71 US 4,414,414, 1-1-1981
 72 US 4,414,414, 1-1-1981
 73 US 4,414,414, 1-1-1981
 74 US 4,414,414, 1-1-1981
 75 US 4,414,414, 1-1-1981
 76 US 4,414,414, 1-1-1981
 77 US 4,414,414, 1-1-1981
 78 US 4,414,414, 1-1-1981
 79 US 4,414,414, 1-1-1981
 80 US 4,414,414, 1-1-1981
 81 US 4,414,414, 1-1-1981
 82 US 4,414,414, 1-1-1981
 83 US 4,414,414, 1-1-1981
 84 US 4,414,414, 1-1-1981
 85 US 4,414,414, 1-1-1981
 86 US 4,414,414, 1-1-1981
 87 US 4,414,414, 1-1-1981
 88 US 4,414,414, 1-1-1981
 89 US 4,414,414, 1-1-1981
 90 US 4,414,414, 1-1-1981
 91 US 4,414,414, 1-1-1981
 92 US 4,414,414, 1-1-1981
 93 US 4,414,414, 1-1-1981
 94 US 4,414,414, 1-1-1981
 95 US 4,414,414, 1-1-1981
 96 US 4,414,414, 1-1-1981
 97 US 4,414,414, 1-1-1981
 98 US 4,414,414, 1-1-1981
 99 US 4,414,414, 1-1-1981
 100 US 4,414,414, 1-1-1981

NO INDEXING IS AVAILABLE FOR THIS PATENT.

AB This invention is directed to a method of encapsulating cells or other cells are encapsulated within a spherical semipermeable membrane comprising a polysaccharide having acidic groups cross-linked with a polymer having a molecular weight greater than 10,000. The cells within the microcapsules are viable, healthy, physiologically active and capable of forming metabolites. The encapsulated cells are useful for implantation in a mammalian body to produce substances and effect a change in the characteristics of the cells in vivo tissue.

NO INDEXING IS AVAILABLE FOR THIS PATENT.

1A ANSWER 41 OF 45, USPATFULL
 11 US 4,414,414, 1-1-1981
 12 US 4,414,414, 1-1-1981
 13 US 4,414,414, 1-1-1981
 14 US 4,414,414, 1-1-1981
 15 US 4,414,414, 1-1-1981
 16 US 4,414,414, 1-1-1981
 17 US 4,414,414, 1-1-1981
 18 US 4,414,414, 1-1-1981
 19 US 4,414,414, 1-1-1981
 20 US 4,414,414, 1-1-1981
 21 US 4,414,414, 1-1-1981
 22 US 4,414,414, 1-1-1981
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 24 US 4,414,414, 1-1-1981
 25 US 4,414,414, 1-1-1981
 26 US 4,414,414, 1-1-1981
 27 US 4,414,414, 1-1-1981
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 39 US 4,414,414, 1-1-1981
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 95 US 4,414,414, 1-1-1981
 96 US 4,414,414, 1-1-1981
 97 US 4,414,414, 1-1-1981
 98 US 4,414,414, 1-1-1981
 99 US 4,414,414, 1-1-1981
 100 US 4,414,414, 1-1-1981

NO INDEXING IS AVAILABLE FOR THIS PATENT.

AB A core material such as living tissue, individual cells, hormones, enzymes or antibodies is encapsulated in a semipermeable membrane that is permeable to small molecules for contact with the core material but is impermeable to potentially deleterious large molecules. Encapsulation may be carried out by suspending the core material in a aqueous medium containing a water soluble gum that can be reversibly gelled, forming the suspension into droplets, contacting the droplets with a solution of multivalent cations to gel the droplets, forming discrete, shape-retaining, water insoluble temporary capsules and cross-linking a surface layer of the temporary capsules to produce a semipermeable membrane around the capsules. Optionally, the gel within the membrane may be reified by removing multivalent cations from the gel.

NO INDEXING IS AVAILABLE FOR THIS PATENT.

1A ANSWER 41 OF 45, USPATFULL
 11 US 4,414,414, 1-1-1981
 12 US 4,414,414, 1-1-1981
 13 US 4,414,414, 1-1-1981
 14 US 4,414,414, 1-1-1981
 15 US 4,414,414, 1-1-1981
 16 US 4,414,414, 1-1-1981
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1. *Chlorophyll a* (Chl *a*) and *Chlorophyll b* (Chl *b*) were determined using the method of Lichtenthaler and Whistler (1973). The total chlorophyll content was determined using the method of Lichtenthaler and Whistler (1973). The total chlorophyll content was determined using the method of Lichtenthaler and Whistler (1973).

[illegible][illegible]

1. *Journal of the American Medical Association*, 1990; 263: 1025-1028.

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ACKNOWLEDGMENTS

1. *Staphylococcus aureus* 2. *Staphylococcus epidermidis* 3. *Staphylococcus saprophyticus* 4. *Staphylococcus sciuri* 5. *Staphylococcus carnosus* 6. *Staphylococcus hyicus* 7. *Staphylococcus epidermidis* 8. *Staphylococcus aureus* 9. *Staphylococcus aureus* 10. *Staphylococcus aureus* 11. *Staphylococcus aureus* 12. *Staphylococcus aureus* 13. *Staphylococcus aureus* 14. *Staphylococcus aureus* 15. *Staphylococcus aureus* 16. *Staphylococcus aureus* 17. *Staphylococcus aureus* 18. *Staphylococcus aureus* 19. *Staphylococcus aureus* 20. *Staphylococcus aureus* 21. *Staphylococcus aureus* 22. *Staphylococcus aureus* 23. *Staphylococcus aureus* 24. *Staphylococcus aureus* 25. *Staphylococcus aureus* 26. *Staphylococcus aureus* 27. *Staphylococcus aureus* 28. *Staphylococcus aureus* 29. *Staphylococcus aureus* 30. *Staphylococcus aureus* 31. *Staphylococcus aureus* 32. *Staphylococcus aureus* 33. *Staphylococcus aureus* 34. *Staphylococcus aureus* 35. *Staphylococcus aureus* 36. *Staphylococcus aureus* 37. *Staphylococcus aureus* 38. *Staphylococcus aureus* 39. *Staphylococcus aureus* 40. *Staphylococcus aureus* 41. *Staphylococcus aureus* 42. *Staphylococcus aureus* 43. *Staphylococcus aureus* 44. *Staphylococcus aureus* 45. *Staphylococcus aureus* 46. *Staphylococcus aureus* 47. *Staphylococcus aureus* 48. *Staphylococcus aureus* 49. *Staphylococcus aureus* 50. *Staphylococcus aureus* 51. *Staphylococcus aureus* 52. *Staphylococcus aureus* 53. *Staphylococcus aureus* 54. *Staphylococcus aureus* 55. *Staphylococcus aureus* 56. *Staphylococcus aureus* 57. *Staphylococcus aureus* 58. *Staphylococcus aureus* 59. *Staphylococcus aureus* 60. *Staphylococcus aureus* 61. *Staphylococcus aureus* 62. *Staphylococcus aureus* 63. *Staphylococcus aureus* 64. *Staphylococcus aureus* 65. *Staphylococcus aureus* 66. *Staphylococcus aureus* 67. *Staphylococcus aureus* 68. *Staphylococcus aureus* 69. *Staphylococcus aureus* 70. *Staphylococcus aureus* 71. *Staphylococcus aureus* 72. *Staphylococcus aureus* 73. *Staphylococcus aureus* 74. *Staphylococcus aureus* 75. *Staphylococcus aureus* 76. *Staphylococcus aureus* 77. *Staphylococcus aureus* 78. *Staphylococcus aureus* 79. *Staphylococcus aureus* 80. *Staphylococcus aureus* 81. *Staphylococcus aureus* 82. *Staphylococcus aureus* 83. *Staphylococcus aureus* 84. *Staphylococcus aureus* 85. *Staphylococcus aureus* 86. *Staphylococcus aureus* 87. *Staphylococcus aureus* 88. *Staphylococcus aureus* 89. *Staphylococcus aureus* 90. *Staphylococcus aureus* 91. *Staphylococcus aureus* 92. *Staphylococcus aureus* 93. *Staphylococcus aureus* 94. *Staphylococcus aureus* 95. *Staphylococcus aureus* 96. *Staphylococcus aureus* 97. *Staphylococcus aureus* 98. *Staphylococcus aureus* 99. *Staphylococcus aureus* 100. *Staphylococcus aureus*

These inner cross-linked esters of polyols and diacids are useful in the field of biodegradable plastic materials, in manufacture of sutures and surgical articles, in the production of pharmaceutical fields, in the food industry and in many other industrial fields.

CONCLUSION: The use of the new inner esters, for example in the field of biodegradable plastic materials, to manufacture of dental and surgical articles, in the cosmetic and pharmaceutical fields, in the food industry and in many other industrial fields.

... as the natural ones of animal or vegetable origin, and their derivatives of the same, but above all, alginic acid, alginic acid, carboxymethylcellulose, carboxymethyl...
... as "carboxymethylcellulose" and carboxymethyl...
... the external partial esters of alginic acid, polysaccharide, ... of hyaluronic acid and **alginic acid**, may serve as ...
...
... The partial esters of carboxymethylcellulose, of carboxymethylamide and carboxymethylchitin which are used as ...

formation of activated esters are those, such as, carboxylic acid chlorides, carbodiimides, diisylchexylcarbodiimide, benzyl-3-(3'-phyl-carbodiimide, benzyl-ethyl-carbodiimide; ethoxyacetylene; Woodward's reagent (4-ethyl-5-phenylisoxazolium-3'-sulfonate), or halogen derivatives from aliphatic, cyclicaliphatic or aromatic hydrocarbons, such as heterocyclic compounds with halogen made available by the presence of the active group.

... sulfonides, such as tetramethylene sulfone, dioxolane or
polyethylene sulfones, such as tetramethylene sulfone, dioxolane and
lower alkyl dialkylsulfides of lower **aliphatic** nature.
The alkyl groups have a maximum of six carbon atoms, and the dimethyl
dialkyl formamide- or ... also be used, however, ... these need not
always be aprotic, such as alcohols, ethers, ketones, esters, such as
lower **aliphatic** dialkoxymethylcarbamides, such as
dimethoxyethane and especially **aliphatic** or hetero ...
alcohols and ketones with a low boiling point, such as lower
N-alkyl-pyrrolidones, such as N-methylpyrrolidone
N-methyl-pyrrolidone, hexamethylphosphoride, ...
... bases and especially tertiary ...
pyridine and its homologues, such as collidine, **aliphatic**
amine bases, such as triethylamine or N-methyl-diethyl ...

[illegible]

aliphatic

A number of the cyclic aliphatic : **aliphatic** cycl. is also
derived by dehydr. from acyclic hydroxy carbinols, may be formally
have a maximum of 14 carbon atoms, may be linear or branched and may contain
one or more substituents, such as those mentioned above for the
aliphatic alcohols. Of the alcohols derived from cyclohexane used
hydroxy carbinol hydrates, special mention should be made of those with a
molecular w.

Aliphatic-aryl aliphatic polyhydric alcohols, i.e., 1,2,3,4,5,6-hexahydro-1,2,3,4,5,6-hexakis(hydroxymethyl)-2,4,6-trimethyl-1,3,5-triazine, the esters of the present invention are diethyl, triethyl, butyl and octyl, 2,4,6-trimethyl-1,3,5-triazine.

Water-soluble polymers are considered to be non-flammable if they would not undergo an **aliphatic**-cyclic polymerization, if their linear or cyclic chains are interrupted, i.e., if there exists a break in the chain between the two chain-ends.

Example 1. The following are the alcohols which are able to esterify with acetic acid. The alcohols which are not able to esterify with acetic acid are generally lower **aliphatic** alcohols with a basic character. The alcohols are, especially saturated monovalent alcohols, the saturated, glycol alcohols, isopropanol, and, in general, alcohols.

esters of these cross-linked products, which are then deriving from therapeutically inactive alcohols, such as for example substituted or unsubstituted **aliphatic** alcohols, for example unsubstituted alcohols of this kind with a straight or branched chain, for example with between 1 and 10 carbon bonds, such as vinyl or allyl alcohols and their condensed derivatives, or polyols such as alcohols, such as glycerine. Also useful are **aliphatic** alcohols, for example those derived from cyclopentane or cyclohexane and their derivatives substituted by lower alkyl groups, for example alcohols with between 1 and 4 carbon atoms, especially by methyl groups. Particular interesting are also esters with cycloaliphatic and **aliphatic** -aryl aliphatic alcohols derived from terpenes, which can be mentioned as well as from therapeutically active alcohols, which are also useful in themselves.

Extremely important is the use of cross-linked products in hybrid use and for the manufacture of sanitary **surgical** items. The esters of these cross-linked products are preferably to be mentioned above for use in cosmetics.

50001 systemic effect thanks to translocation of lipid, for
example in suppositories. All these applications are available both in
human and veterinary **medicine**. In human **medicine** the
new medicaments are particularly suitable for parent use. The present
invention includes in particular any one of them in a tablet.

... in cross-linking are preferably free of substituents which are esterified with pharmacologically inactive alcohols, for example one of the lower aliphatic alcohols mentioned previously.

These two groups can, however, also contain people who are not with
us. We must have therefore a constant control.

4. The important field of the present invention is not limited by sanitary and **surgical** articles, by their manufacturing process, or by their use. These articles are for example similar to those already known.

Surgical and sanitary articles of special importance are those which can be obtained from appropriate substances: the cross-linked products in olefinic liquids which are capable of being cast into films, sheets and threads to be used in **surgery** as auxiliary or substitutive articles for the skin in cases of severe damage to this organ, such as burns, or as suture threads in **surgery**. The present invention includes particularly these uses and a process for preparing suitable articles consisting in a liquid olefinic monomer.

[illegible]

the 1990s, the number of people in the world who are under 15 years of age is expected to increase from 1.1 billion to 1.5 billion. The number of people aged 65 and over is expected to increase from 200 million to 400 million. The number of people aged 15 and over is expected to increase from 3.5 billion to 4.5 billion. The number of people aged 15 and over is expected to increase from 3.5 billion to 4.5 billion. The number of people aged 15 and over is expected to increase from 3.5 billion to 4.5 billion.

- which may be polymeric either in the solid, liquid, or gaseous state. **aliphatic** alcohols, for example, may be used in the preparation of articles with a high degree of resistance to water, which may be used in the preparation of linings for the treatment of **wounds** and in **surgery**. The use of such linings has the special advantage of being non-irritating to the skin and mucous membranes, as compared with other materials.
- 1870 In the preparation of the so-called sanitary and **surgical** articles, it is possible also to include or combine with stabilizing materials in order to improve their mechanical and chemical properties, such as:
- 1871 Another application in the fields of **medicine** and **surgery** of the cross-linked hyaluronic products is represented by the preparation of various solid inserts such as prostheses, discs, lenses, etc.
- 1872 In the skin, with its elimination of impurities and negative reactions. For this reason there is a constant demand for **medicine** and a suitable material. The cross-linked products of hyaluronic acid may be safely used to correct skin defects.
- 1873 Part of the applications in the fields of **medicine** and **surgery** of the new hyaluronic derivatives according to the present invention are preparations made of expanded material, especially in the form of sponges, for the treatment of **wounds** in various lesions.
- 1874 The above applications of the cross-linked products of a hyaluronic acid base represent the ideal solution for these sanitary and **surgical** articles which are intended to be introduced in the way or inserted into human or animal organisms in the form of the same articles, using other cross-linked polysaccharides according to the invention, such as those mentioned above and especially those with an **alginic** acid base. In the same way, too, the cross-linked products are broken down in the organism to give acid polysaccharides.
- 1875 In the cross-linked **alginic** acid products, special attention should be given to industrial and household uses of emulsiles and elementary articles and their uses. These, especially in the form of cross-linked partial salts, possibly further esterified with inert alcohols, such as especially lower **aliphatic** alcohols, for the preparation of soles, which can be widely used in the shoe industry, for the manufacture of ice-creams, etc., etc. Another property is their ability to emulsify and to stabilize emulsions. From this point of view, too, the **alginic** cross-linked products are important in the food industry, where they serve in the preparation of condiments and for the stabilization of many drinks such as beer and fruit juice, sauces and syrups. As emulsifiers, **alginic** cross-linked products can be used in the manufacture of polishes, anti-freeze, etc., lactics and as stabilizers in the ceramics and.
- 1876 Preparation of Cross-Linked **Alginic** Acid
- 1877 4.17 g of **alginic** acid tetrabutylammonium salt (from **alginic** acid obtained from Laminaria hyperborea) corresponding to 1.17 mEq of a monomeric unit are solubilized in 4 ml of DMSO.
- 1878 Preparation of Cross-Linked **Alginic** Acid
- 1879 4.17 g of **alginic** acid tetrabutylammonium salt (from **alginic** acid obtained from Aegophyllum medosum) corresponding to 1.17 mEq of a monomeric unit are solubilized in 4 ml of DMSO.
- 1880 Preparation of Cross-Linked **Alginic** Acid
- 1881 4.17 g of **alginic** acid tetrabutylammonium salt (from **alginic** acid obtained from Macrocystis pyrifera) corresponding to 1.17 mEq of a monomeric unit are solubilized in 4 ml of DMSO.
- 1882 Preparation of Cross-Linked **Alginic** Acid
- 1883 4.17 g of **alginic** acid tetrabutylammonium salt (from **alginic** acid obtained from Laminaria hyperborea) corresponding to 1.17 mEq of a monomeric unit are solubilized in 4 ml of DMSO.
- 1884 Preparation of Cross-Linked **Alginic** Acid
- 1885 4.17 g of **alginic** acid tetrabutylammonium salt (from **alginic** acid obtained from Macrocystis pyrifera) corresponding to 1.17 mEq of a monomeric unit are solubilized in 4 ml of DMSO.
- 1886 Preparation of Cross-Linked **Alginic** Acid
- 1887 1.8 g of carboxylic acid solubilized with sodium 4.17 g of **alginic**

- 4.17 g of **alginate** acid tetraethylammonium salt of **alginate** acid obtained from *Aerophyllum* spores, containing 10 mg of a monomeric unit are solubilized in 4 ml of DMSO.
- 181 Preparation of Cross-Linked **Alginic Acid**
- 182 4.17 g of **alginate** acid tetraethylammonium salt of **alginate** acid obtained from *Laminaria hyperborea* spores, containing 10 mg of a monomeric unit are solubilized in 4 ml of DMSO.
- 183 Preparation of the Partial Ethyl Ester of Cross-Linked **Alginic Acid**
- 184 4.17 g of **alginate** acid tetraethylammonium salt of **alginate** acid obtained from *Aerophyllum* spores, containing 10 mg of a monomeric unit are solubilized in 4 ml of DMSO.
- 185 Preparation of the Partial Ethyl Ester of Cross-Linked **Alginic Acid**
- 186 4.17 g of **alginate** acid tetraethylammonium salt of **alginate** acid obtained from *Laminaria hyperborea* spores, containing 10 mg of a monomeric unit are solubilized in 4 ml of DMSO.
- 187 Preparation of the Ethyl Ester of Cross-Linked **Alginic Acid**
- 188 4.17 g of **alginate** acid tetraethylammonium salt of **alginate** acid obtained from *Aerophyllum* spores, containing 10 mg of a monomeric unit are solubilized in 4 ml of DMSO.
- 189 The following preparations exemplify the medical uses according to the invention containing the **alginate** esters.
- 190 What is claimed is:
1. Cross-linked acidic polysaccharides according to claim 1, wherein said polysaccharides are selected from the group consisting of **alginate** acid, **alginate** acid, carboxymethylcellulose, and carboxymethylchitin.
2. Cross-linked acidic polysaccharides according to claim 6, wherein said alcohol is a member selected from the group consisting of **aliphatic**, **araliphatic**, **cycloaliphatic** and **heterocyclic** alcohols.
3. Cross-linked acidic polysaccharides according to claim 7, wherein said alcohols of the **aliphatic** series have a maximum of 34 carbon atoms and may be substituted by one or two functional groups selected from the group consisting of hydroxyl, amino, and alcohols of the **aliphatic** series may be interrupted in the carbon chain by heteroatoms selected from the group consisting of oxygen, sulfur and . . .
4. polysaccharides according to claim 7, wherein said alcohols of the **araliphatic** series have only one benzene residue and have an **aliphatic** chain with a maximum of 4 carbon atoms and the benzene residue may be substituted by between 1 and 3 methyl or hydroxy groups, or by halogen atoms, and wherein the **aliphatic** chain may be substituted by one or two functions selected from the group consisting of free amino groups, mono- or . . .
14. Cross-linked acidic polysaccharides according to claim 7, wherein said alcohols of the **cycloaliphatic** or **aliphatic** -**cycloaliphatic** series are mono- or polycyclic hydrocarbons with a maximum of 34 carbon atoms.
15. Cross-linked acidic polysaccharides according to claim 7, wherein said heterocyclic alcohols are mono- or polycyclic **cycloaliphatic** or **aliphatic** **cycloaliphatic** alcohols interrupted in the carbon chain or ring by one or more heteroatoms selected from the group consisting of . . .
16. A salt according to claim 17, wherein said salt is a **aliphatic**, **araliphatic**, **cycloaliphatic** or **heterocyclic** salt.
17. A salt according to claim 17, wherein said salt is a **aliphatic**, **araliphatic**, **cycloaliphatic** or **heterocyclic** salt.
18. A salt according to claim 17, wherein said salt is a **aliphatic**, **araliphatic**, **cycloaliphatic** or **heterocyclic** salt.
19. Cross-linked acidic polysaccharides or a salt thereof according to claim 1, wherein said polysaccharide is **alginate** acid.

- [illegible]

[illegible]

alginate

1. 7. 2000

alginate

46 Total and partial extracts of alginic acid containing poly-
merized alginate gelatin and glutaraldehyde polymerized cross-linked al-
ginate gels are available including medical, surgical, dental and
industrial uses.

1. **alginate** is a natural polysaccharide found in brown algae. It is a linear polymer of D-mannuronic and L-gulonic acid residues.

0000 The drug also provides interesting and valuable data on the
0001 pharmaceutical properties and may be used in other fields, from
0002 surgery to medicine. The invention
0003 also includes pharmaceutical preparations containing the active
0004 ingredient in the form of **alginic acid esters** and in the
0005 salt as described above, as well as medicaments containing
0006 the a) varying vehicle comprising a total of parts by weight of
0007 **alginic acid**.

1980) The invention also includes various uses of alginic acid for the at least mentioned indications, such as in the treatment of medicine, surgery, ophthalmology and otorhinolaryngology. The invention also includes alginic acid esters.

Alginic acid is a natural acidic polysaccharide which is found in all forms of so-called brown algae (Phaeophyta) and in some molecular weight varying . . . the algae, as well as the age of the plant itself. The main species of brown algae used to obtain **alginic acid** are, for example, *Macrocystis Pyrifera*, *Ulva lactuca*, *Laminaria hyperborea*, *Laminaria flexilis*, *Laminaria digitata*, *Desmoullella radicum*, *Fucus serratus*.

FROM **Alginic acid** is found in these algae as an extensive constituent of the cell walls in the form of a salt. Some of its alkaline salts, of these especially sodium salt, its nature is also known as "algin". These salts are normally extracted in aqueous conditions with a sodium carbonate solution and it is possible to obtain **alginic acid** directly from this extract by precipitation with an acid, for example a mineral acid such as hydrochloric acid. Another first making an insoluble sodium salt by adding a soluble sodium salt, such as chloride, and then washing this salt, **alginic acid** is obtained again by treatment with an acid.

XXXXX Alginate acid or alkaline alginates may, however,
also be obtained microbially, for instance by fermentation with
Pseudomonas aeruginosa or mutants of *Pseudomonas aeruginosa*, *Pseudomonas*
fluorescens, ...

NOTE: The metal salts of **alginic acid**, especially the sodium and alkaline earth metal salts, have interesting chemical and physical properties and are therefore widely used in industry. Thus, for example, the solutions of alkaline or alkaline earth **alginates** are extremely suitable, due to their viscosity, and their adjustability by temperature and pH, for the preparation of gels with various types of tastes and puddings. Another property which is widely exploited in the field of alimentation is the ability of **alginates** to retain water, and for this reason they are used for example in the preservation of many types of frozen foods. A third property of **alginates** is their power to emulsify and to stabilize emulsions; for this reason too these salts are important in the food industry.

XXXX The ability of **alginate** solutions to form films and gels has been exploited in the paper industry, in making various labels, in textile printing and dyeing, and in the preparation of dentistry, medical and **surgical** articles. **Alginates** are used as emulsifiers in the preparation of polishes, and as foams, latices and as stabilizers in the storage and distribution of emulsions.

Alginic acid and its salts have also been used in the pharmaceutical, medical, **surgical** and cosmetic fields. For example for the preparation of dressings for burns and sanitary and **surgical** articles. For example the British official description of "Alginate dressing" describes "artificial skin" consisting of layers of the salt, from which the water is removed.

alginate :

On addition of water to alginate, alginate "gel" forms. The gel is formed in part by the water filling the voids in the gel. This leaves the polymer chains in contact with each other. **alginate**, which is found in the layer of denture, is a naturally occurring polymer. It is a linear polymer of D-glucuronic acid, which is a sugar and is found in the denture. It is a linear polymer of D-glucuronic acid, which is a sugar and is found in the denture.

In this **alginate** has been used for the production of the film for use in the pharmaceutical industry. Films of the polylactide-N-vinyl pyrrolidone copolymer, Surfactant pattern N 17, and the polylactide film in the form of a beaded and spherical beads were prepared.

wounds, **zinc** is a common **alginate** filler. **Alginate** is also used as a hemostatic agent in the treatment of bleeding profusely from the walls of hemorrhoids. **Alginate** is used in the treatment of ulcers, fistulas, and in the treatment of abscesses. **Alginate**, **zinc** and **alginate** are also used as fillers for pills, and **alginate** is also used for its binding properties.

Also used in industry in many of the above mentioned cases are the **alginic acid esters** or salts of such esters, more particularly ethylene glycol and propylene glycol esters. The latter is used for example in U.S. Pat. 2,331,013 by Academic Press, Inc. 1949, pp. 44-443. The above mentioned esters have been claimed by U.S. Pat. 2,331,013 as **alginic acid**, or its salt or partial salt, with ethylene glycol or propylene glycol respectively. This preparation method is also claimed in U.S. Patents in the preparation of the above mentioned **alginic acid** esters and esters of divalent alcohols by reaction of an **aliphatic** hydrazine epoxide, possibly substituted or interrupted by hetero atoms in the carbon atom chain (see for example U.S. Pat. 2,331,013).

The **alginic acid esters** obtainable by the action of the above mentioned epoxides on the free acid or its salts, in particular, of glycol esters with a low molecular weight, and a very low degree in the case of glycol esters with **long chains**. It has not been possible to obtain a series of glycol esters with **long chains**. It has not been possible to prepare total esters by this method.

Most valent alcohols, esters, both **aliphatic** and **aromatic**, have
 also been mentioned in literature, where all of them are **derivatives of**
alginic acid obtained by reaction of **alginic acid**.

alginate acid is a polymeric polymer of **alginic acid** in an ethereal solution of dioximethane. (Deitschman, *Chem. physiol. diätet. Chemie*, Vol. 134, p. 121, 1953; A. E. Steiner, *Ind. Eng. Chem. Anal. Ed.*, 1941, 13, 114; U.S. Patent No. 767,132. It seems however that the products obtained by reaction with dioximethane are not really **alginic acid** esters but rather methyl esters of an **alginic acid** partially etherified to the hydroxy alcohol group, as described for example in Example 4 of the above mentioned U.S. Pat. No. 1. One methyl ester has also been obtained by reaction of **alginic acid** with a soluble salt of **alginic acid** in an organic solvent with low solubility in water, but in the presence of water. (U.S. Pat. No. 2,746,706. The product obtained, referred to as **ethyl alginic acid** or **methyl alginate**, is not to be considered a simple ester, since it is known that sugar hydroxyls are easily etherified with

Also mentioned in literature are **alginic acid esters** of monovalent alcohols, with no indication however of their preparation method and no description of their chemical and physical mixed products, as in the case of methyl products. See for example Ref. [6], pp. 1, 10, 14 in which a typical **alginate** is mentioned without indication of its alginic acid preparation method, and also a mixture of 1, 1, 1, 1-tetra-*n*-butyl ester.

alginic acid esters: only three esters of alginic acid are known, and are therefore, only the partial esters of alginic acid.

The main object of the invention is therefore the use of **alginate** esters, such as those already mentioned, and the use of these esters for their intended use.

The present invention concerns new polyamides of the type I and II, previously **alginic** acid esters and particularly of the type I.

The invention also includes the use of alginate alginate

... medicine

and surgery was also included, therefore, new information on the relationship between algal and bacterial flora in the oral cavity, in particular, denture, and surgery, is needed, and the relationship between the oral flora and the periodontal disease, and the relationship between the oral flora and the periodontal disease. The new study is intended to be a review of the relationship between algal and bacterial flora in the oral cavity. In the periodontal disease, the relationship between the oral flora and the periodontal disease is discussed.

The present invention includes a simple and very efficient procedure for the preparation of **alginic esters**, based on the treatment of potassium ammonium salts of **alginic acid** with a functional aliphatic ester in ethanol, preferably glycolic, ethyl glycol, such as in dimethyloxalide, making a large number of new **alginic esters** available, especially those esters of non valent alcohols, such as dimethylol, dimethylol ether, and esters of aromatic alcohols, allyl, etc. The procedure may be used also for the preparation of esters derived from substituted alcohols, in particular from **ω**-hydroxy esters of aliphatic alcohols, obtaining in this way a large number of **alginic esters with aliphatic groups**, as described in the literature, and mainly new **alginic esters of aliphatic alcohols**.

The new alginic esters may be used in various sectors of the food, cosmetic, pharmaceutical, sanitary, surgical and domestic industries, where metal alginates or the esters of aliphatic aliphatic alcohols of the type of glycerol and other esters of alginic acid are already used, for example in the food and cosmetic industries. Therefore, part of the invention is represented by the use of the new esters, and the corresponding articles and industrial products, such as cosmetic, sanitary, surgical and pharmaceutical articles, or food products and the like, especially emulsifying agents, emulsion stabilizers, and thickening agents and possibly related.

With the discovery of the new **alginic** esters and, in view of the present invention, a new use for **alginic** esters in the oral has also come to light, that is for the new esters and for already known. This new use is for substances, especially those with a local, oral or dental action, but also those for parenteral administration. The use of known **alginic** esters of kivalent alcohols was limited to the function of emulsifying agents, emulsion stabilizers, suspending agents and grasihty related uses. No use in the pharmaceutical, sanitary, medical, **surgical** or prosthetic fields was envisaged for these esters. The present invention therefore concerns the new above mentioned use and respective products, especially the pharmaceutical preparations containing an **alginic** ester as vehicle for the active substances.

substance as their alcohol component. Of the armamentarium of preparations of the present invention, therefore, particularly interesting are those containing an **alginic ester** derived from a therapeutically active alcohol, such as those defined hereafter, that is, esters comprised of a **alginic acid** esterified with the alcohol moiety of a therapeutically active compound.

000000 The invention also includes partial **alginic ester** compositions containing alginic bases. In the following description, the term "alginic ester" does not exclude this, the terms "**alginic acid esters**" and "

alginate esters" shall be taken to mean both the esters themselves and their active mentioned salts. In part (c), in the above mentioned pharmaceutical preparations, the terms "active" and "biologically active" substances may be replaced, apart from any other biologically active or inactive **alginate esters**, also by pharmaceutically active and/or substances used to salify pharmaceutically active carboxyl groups of partial **alginic esters**.

The use of the above mentioned alkaline alginates in various forms is of industry, pharmaceutical, surgery and all in the food industry, presents some disadvantages in solid conditions, because of the insolubility of alginic acid with low solubility which may separate in the presence of calcium ions, some insoluble products, alginate hydrogels, and in this sense, alginates are suitable for use in liquid conditions, for example, in food, in medicine, in dentistry.

stabilized.

50000 In addition, in the case of certain alcohols, the **alginic** ester is stabilized by the addition of a small amount of a stabilizing agent, in the case of which the stabilizing agent is a substance which is soluble in water and which is capable of forming a complex with the **alginic** ester, such as a complex of a stabilizer, for example, a complex of a stabilizer. The **alginic** ester is a water-soluble substance, and its action is limited. This is due to the fact that the **alginic** ester is a water-soluble substance, and its action is limited.

50001 In the case of certain alcohols, the **alginic** ester is stabilized by the addition of a small amount of a stabilizing agent, in the case of which the stabilizing agent is a substance which is soluble in water and which is capable of forming a complex with the **alginic** ester, such as a complex of a stabilizer, for example, a complex of a stabilizer. The **alginic** ester is a water-soluble substance, and its action is limited. This is due to the fact that the **alginic** ester is a water-soluble substance, and its action is limited.

50002 The low level of toxicity of the esters of higher valent alcohols of **alginic** acid according to the present invention, which is exploited mainly in the pharmaceutical, cosmetic and sanitary-surgical fields where the new **alginic** esters may be used as biodegradable plastic materials with various functions as the case may be. Thus, for example, the **alginic** ester may be used as additives for the wide range of polymeric materials for sanitary and surgical articles, such as polyurethanes, polyesters, polyolefins, polyamides, polysiloxanes, vinyl and acrylic polymers, with the effect of rendering these materials biodegradable. In this case the addition of an **alginic** ester is carried out for example by coating the surface of these materials or by dispersing them in the same or.

50003 In the cosmetic and pharmaceutical fields, the **alginic** esters of the invention may be used for the preparation of ointments, creams and other types of medicaments for topical use. For example, such as emulsion creams, where they act as stabilizers and emulsifiers having a greater degree of stability than alkaline **alginates**, especially with regard to higher temperatures, and a lower degree of toxicity compared to glycol esters. In pharmaceuticals they may be used as the **alginic** ester serves as vehicle and is associated mechanically, physically mixed with the active substance; or the **alginic** ester (partial) is salted with the active substance; and

50004 or the **alginic** ester is esterified with an alcohol, and represents the active substance.

50005 In the case of variations 12 and 13, it is possible to vary and combine the alcohol residues in the **alginic** ester, or the basic component in the salts, and it is possible to use esters of a mixed character, in.

50006 In the industrial sectors, such as in the food, paper, textile and printing industries, and in the preparation of sanitary, medical and surgical articles, detergents, household articles, etc., are represented by those esters in which the properties of the **alginic** component are the properties to be exploited. The esters derive from alcohols of the **aliphatic**, aromatic, alicyclic, cycloaliphatic or heterocyclic series, and have no toxic or pharmacological action, such as for example the saturated alcohols of the **aliphatic** series or simple alcohols of the **aliphatic** series. Examples of these alcohols are contained in the following.

50007 In therapy is represented by the esters in which the pharmacological properties of the alcohol component are dominant, that is, **alginic** acid esters with pharmacologically active alcohols, such as steroidal alcohols, such as those of the corticosteroid type. These esters possess a wide range of action, but with a wide range of action. Even as compared to already known esters of steroidal alcohols, the **alginic** esters ensure a more balanced, constant and regular pharmacological action and generally a more marked retard effect of the active.

50008 Another group of **alginic** acid esters according to the present invention, and representing a particularly important and useful aspect of the same, is that in which the mixed character of the esters is

previous groups. That is, esters in which part of the carboxylic acid is esterified with a particular alcohol and another part with a particular, usually different, alcohol, or the activity of . . .

SUMM Most of the esters of **alginic acid**, in contrast to the acids, present a certain degree of solubility in organic solvents. This solubility depends on the . . . esterified carboxylic groups and on the type of alkyl group bound to the carboxyl. For example, a total ester of **alginic acid** thus obtained presents at room temperature good solubility, for example, in diethyl ether. The total esters which are all new. . .

SUMM . . . in saline and have the particular desired . . . Such articles may be prepared for example by dissolving an ester of **alginic acid** in an organic solvent, giving the extremely . . . solution the form of the desired article and lastly by extracting with an organic solvent with another solvent which can be mixed with the first, but in which the **alginic acid ester** is insoluble, for example an alcohol or water.

SUMM The invention includes the industrial use of the new **alginic esters** in all the aforementioned sectors, especially in the alimentary, cosmetic, pharmaceutical and medical fields, in the manufacture of household and industrial articles, especially for the manufacture of sanitary and **surgical** articles.

SUMM The invention includes also the use of **alginic esters** in general, that is the new ones and those already mentioned in literature, for the new applications described here, for example in use as vehicles for pharmacologically active substances, either in the form of **alginic esters** with therapeutically active alcohols, or as **alginic esters** of inert alcohols to mix with therapeutically active substances, or with therapeutically active bases as well as the pharmaceutical medicaments or preparations resulting from this use of **alginic esters**. In all cases the free carboxy groups are possibly salified with inactive but therapeutically acceptable bases.

SUMM The main object of the present invention is therefore represented by the total or partial esters of **alginic acid** with alcohols of the **aliphatic**, **araliphatic**, **cycloaliphatic** or **heterocyclic** series and by the salts of such partial esters with inorganic or organic bases, with the exception of the partial esters of bivalent **aliphatic** alcohols.

SUMM A second object of the invention is represented by a new procedure for the preparation of **alginic esters** characterised by the treatment of a quaternary ammonium salt of **alginic acid** with an etherifying agent in an aprotic solvent, and, if desired, by the salification of the free carboxy groups with inorganic or organic bases, and the partial **alginic esters** thus obtained.

SUMM A third object of the invention is represented by the use of the new **alginic esters** and their salts, in substitution of the metal **alginates** or of the **alginates** of **aliphatic** bivalent alcohols, in the respective industrial sectors or in their applications in the cosmetic, pharmaceutical or sanitary-**surgical** fields, and by the respective products or industrial articles.

SUMM A fourth object of the invention is represented by the use of **alginic esters** as vehicles for pharmacologically active substances and by pharmaceutical preparations or medicaments containing:

SUMM 1. a carrying vehicle containing a total or partial ester of **alginic acid** or salts of such partial esters with inorganic or organic bases, or pharmaceutical preparations or medicaments containing an **alginic ester** possibly salified with inorganic or organic bases, in which at least one ester group or a salified carboxy group. . .

SUMM Alcohols of the **aliphatic** series for use as esterifying components of the carboxy groups of **alginic acid** according to the present invention are, for example, those with a maximum of 34 carbon atoms, which may be . . .

SUMM In the above groups containing hydrocarbon radicals there are preferably lower **aliphatic** radicals, such as hetero atoms, oxygen, sulphur, nitrogen and sulfur. Preference is given to alcohols substituted with one or two . . .

SUMM Of the higher saturated **aliphatic** alcohols, those with a maximum of

- special mention are for example **ethyl alcohol** and **isopropyl alcohol**, but especially important for the purposes of the present invention are the **aliphatic alcohols** in the sense that the term **aliphatic** is used to designate those with only the benzene residue and in which the **aliphatic** chain has a maximum of 4 carbon atoms, in which one or more benzene residue may be substituted by between 1 and 3 methyl or isopropyl groups and/or halogen atoms, especially chlorine, bromine or iodine and in which the **aliphatic** chain may be substituted by one or more functional groups selected from the group consisting of free amino groups, etc.
- SUM 1 The alcohols of the cycloaliphatic or **aliphatic** cycloaliphatic series may derive from mono or polycyclic hydrocarbons and may have a maximum of 34 carbon atoms. Of these:
- SUM 2 Polycyclic **aliphatic** cycloaliphatic alcohols for example containing the esters of the present invention are the **steroids**, **choleic acids** and **steroids**, such as:
- SUM 3 The total and partial esters of **alginic acid** and of the invention have the following general formula: $R_1(R_2)_n$ wherein R_1 and R_2 are each independently hydrogen or an **aliphatic** moiety selected from the group consisting of **aliphatic**, **cycloaliphatic**, **aliphatic**, **cycloaliphatic** and heterocyclic radicals and pharmaceutically acceptable salts thereof with the proviso that said partial ester is not a **steroid**.
- SUM 4 As discussed above, in some cases **alginic acid** esters in which the ester groups derive from one or more hydroxy compounds with therapeutic action, may be of use with the same or similar activity as that of the esterifying component. In particular, it is possible to have **alginic** esters deriving in the whole or in part from an anti-inflammatory steroid, such as one of those mentioned above, and in the:
- SUM 5 The degree of esterification of **alginic acid** with the above mentioned alcohols depends first and foremost on the special properties desired for the various fields of use. To such extent, for example the skin. Usually, a high degree of esterification to the point of total esterification of **alginic acid** increases its lipophilic character and therefore decreases its solubility in water. For a use in therapy of the new, it is of utmost importance to control the degree of esterification in order to ensure, despite good or increased lipophilia compared to metal **alginates**, a sufficient degree of hydrophilicity, for example a solubility of 10 mg/ml. Naturally it is necessary to consider also the:
- SUM 6 As has been said previously, esterification of the hydroxy groups of **alginic acid** may play several roles, to be exploited in various fields, for example in **medicine**, using the esters as therapeutic agents or in **surgery** using them as plastic articles. For use in therapy we have already said that esterification of an alcohol can in itself be considered therapeutically active, such as anti-inflammatory corticosteroids for example, with **alginic acid** as a means of improving therapeutic efficacy.
- SUM 7 With regard to similar therapeutically active alcohols **alginic acid** acts therefore as a particularly efficient vehicle which is perfectly compatible with the biological environment. Many of these pharmacologically active esters with therapeutically active alcohols it is possible to esterify part or all of the remaining hydroxy groups of the **alginic** component with pharmacologically inert alcohols, such as for example saturated lower **aliphatic alcohol**, for example **ethyl** or **isopropyl alcohols**.
- SUM 8 It is possible to obtain drugs with a "retard" action with **alginic** esters with therapeutically active alcohols, preferably mixed also with therapeutically active bases.
- SUM 9 For cosmetic purposes it is preferable to use total or partial esters of **alginic acid** with pharmacologically inert alcohols, for example saturated or unsaturated **aliphatic alcohols**, for example substituted alcohols of this type with straight or branched chains, for example between 1 and 4 carbon atoms, especially:
- SUM 10 For example alcohols with between 1 and 4 carbon atoms, especially for methyl groups. Particularly interesting are alcohols with cycloaliphatic and **aliphaticcycloaliphatic** alcohols derived from terpenes, such as those mentioned above and other therapeutically active alcohols, which can otherwise be used in the:
- SUM 11 The alcohols to be used preferably for the manufacture of sanitary and

surgical articles are essentially the same as those mentioned above for therapeutic use.

- SUMM Thus, for example, for the manufacture of surgical -surgical articles it is preferable to use total or partial esters with a high rate of esterification, for example between 70% and 90%.
- SUMM are those partial esters in which at least one and at the most 5% of all the carboxy groups of **alginic** acid are esterified, and especially those with a percentage of between 1 and 5%, are used preferably for alimentary,
- SUMM Of particular interest are the salts with organic bases, especially acidized bases and, therefore, **aliphatic**, **aromatic**, **cycloaliphatic** or **heterocyclic** amines. These and other salts may derive from therapeutically acceptable amines or from non-therapeutically active amines, or from amines with a therapeutic action of the first type, preferred are **aliphatic** amines, for example **ethyl**, **diethyl** and **triethyl**amines with alkyl groups with a maximum of 6 carbon atoms or arylalkylamines with
- SUMM Examples of specific useful drugs are all those mentioned above having acidized basic groups regarding the **alginic** acid with therapeutically active alcohols or those mentioned first in this text, for example the various antibiotics.
- SUMM the partial esters with the aforesaid therapeutically active bases and the use of such salts represents a particular case of **alginic** esters functioning as a vehicle, obtainable by the simple addition to the active substance of partial or total esters of
- SUMM The vehicling action of **alginic** esters therefore presents possibilities for new medicaments wherein the active substance is an **alginic** ester as described above or one of its salts.
- SUMM These medicaments are a further object of the invention. The **alginic** esters for use in these medicaments are all those in which the esterifying alcohol is itself not pharmacologically active, for example a simple **aliphatic** alcohol, as described above. Included in the invention however are medicaments of this type in which the ester is also
- SUMM In such medicaments, if partial esters of **alginic** acid are used, the possible salification of the remaining carboxy groups is carried out preferably with therapeutically neutral basic substances corresponding association of substances having a basic character, such as for example antibiotics containing amino groups, and if partial esters of **alginic** acid are used with remaining free carboxy groups, a salt is formed between the free carboxy groups of **alginic** acid and these basic substances. The basic substance may or may not be expensive, thus having basic salts. The new medicaments therefore include in particular the partial esters of **alginic** acid partially salified with pharmacologically active substances of basic character, as described above. The nonesterified carboxy groups are also
- SUMM The use of **alginic** esters as a vehicle is particularly useful in ophthalmology, where it is possible to observe a particular compatibility of the
- SUMM By using the esters of the present invention these difficulties can be overcome. The presence of the **alginic** ester as vehicle in ophthalmic drugs allows for the formulation of excellent preparations with no concentration gradient of the active
- SUMM systemic effect thanks to transcutaneous absorption, for example in suppositories. All these applications are possible both in human and veterinary **medicine**. In human **medicine** the new medicaments are particularly suitable for use in ophthalmology. The present invention therefore also includes in particular any use
- SUMM catalogued with respect to their use in the various fields of therapy, beginning with the distinction between human and veterinary **medicine** and then specifying the various sectors of application with respect to the organs or tissues to be treated, for example anti-inflammatory, vasoconstrictor, antibacterial, and anal drip. In the field of ophthalmology, the indications are in particular for example: mytic, anti-inflammatory, wound healing and antiproliferative effects.
- SUMM several antibiotics with one or more other actions, or one or more antibiotics with a hydrating or a pyrolytic wound healer or an anti-allergic agent etc. For example, it is possible to use

the following are variations of pharmaceutical compositions containing phenyl-phenyl-dexamethasone.

SUMMI associations of various active substances for use limited to these fields, but in all the above cases

medicine it is possible to use associations similar to those already in use for the pharmaceutical preparation of the art.

SUMMI the use-referred to above wherein the of a basic character, the salts formed with a partial **alginic acid** may be of various types. That is, all of the may be classified as follows

SUMMI According to a main aspect of the invention however, the medicaments containing the **alginic acid** ester or its salts are used with therapeutically active or inactive substances as alone excepting possibly an aqueous all types of medicaments described here and also mixtures of such medicaments, as well as possibly mixtures of the new **alginic acid** esters with the

alginic acid or mixtures of their salts, for example salts.

SUMMI hydrocortisone, prednisolone, flurimetholone, prednisolone and possibly their esters, for example phosphoric acid esters; steroid anti-inflammatories, for example indomethacin, naproxen, ibuprofen, flurbiprofen; wound healers such as epidermal growth factor (EGF); local anesthetics, such as Benzocaine, propylamine and possibly their salts; cholinergic agonists such

SUMMI substances are used, such as those referred to above, the salts of the basic active substances and the partial ester of **alginic acid** may be mixed salts of one or more of such active substances or possibly mixed salts of this type with the mentioned metals or bases. For example it is possible to prepare salts of a partial ester of **alginic acid** with a pharmaceutically inactive alcohol, for example a lower alkyl, and with a certain percentage of the acid groups one of the mentioned metals. It is possible to mix this type of mixed salt with free

alginic acid or its fractions or their metal salts, as indicated above for the medicaments constituted by salts of only

SUMMI to determine by analogy which medicaments according to the present invention are to be used in the other fields of **medicine** mentioned above, such as otolaryngology or dentistry or in internal **medicine**. For example, in endocrinology, it is possible to use preparations absorbed intradermally or through the skin, for example by rectal

SUMMI METHODS OF PREPARATION FOR THE **ALGINIC ESTERS**

SUMMI According to the chemically new and original process of the present invention, the **alginic acid** esters may be prepared to advantage starting with quaternary ammonium salts of **alginic acid** with an etherifying agent in a preferably aprotic organic solvent, such as dialkylsulfoxides, dialkylcarboxamides, and in particular lower alkyl dialkylsulfoxides, above all dimethylsulfoxide, and lower alkyl dialkylamides of lower **aliphatic** acids, such as dimethyl or diethyl formamide or dimethyl or diethylacetamide. It is possible, however, to use other solvents which are not always aprotic, such as alcohols, ethers, ketones, esters, especially **aliphatic** or heterocyclic alcohols and ketones with a low boiling point, such as hexafluoroisopropanol and trifluoroethanol. The reaction is brought about

SUMMI The preferred esterification process, therefore, comprises reacting, in an organic solvent, a quaternary ammonium salt of **alginic acid** with a stoichiometric quantity of a compound of the formula

SUMMI wherein A is selected from the group consisting of **aliphatic**, **araliphatic**, **cycloaliphatic**, **aliphatic-cycloaliphatic** and heterocyclic radicals and X is a halogen atom, and the stoichiometric quantity of A-X is determined by the

SUMMI As starting quaternary **ammonium** salts it is preferred to use lower **ammonium** tetraalkylates, the **alkyl** groups having preferably between 1 and 6 carbon atoms. Most often, the **alginate** of tetrabutylammonium is used. These

quaternary **ammonium** salts can be prepared by reacting a metal salt of **alginic acid**, preferably one of those mentioned above, especially sodium or potassium salt, in aqueous solution with a sulfonic resin sulfated with the quaternary **ammonium** salt.

The tetraalkylammonium alginates derived from **alkyls**, especially **alkyls** with between 1 and 4 carbon atoms, are new and find another aspect of the present invention. Unexpectedly, these salts proved to be soluble in many organic solvents, and esterification of **alginic acid** and salts. The present new procedure is therefore made possible easily and gives abundant yields. Only by using this procedure, it is possible to exactly determine the number of carboxy groups of **alginic acid** to be esterified.

THE PRESENT INVENTION provides a new and simple method in reacting a preparation of sodium salt of **alginic acid**, suspended in a suitable solution such as dimethylsulfoxide, with an alkylating agent in the presence of a catalyzing agent, such as tetraabutylammonium iodide. The new procedure made it possible to obtain, as already stated, the total esters of **alginic acid** and also substituted alcohols such as glycols, which were previously unobtainable.

TO PREPARE new esters according to the present invention, it is possible to use **alginic acids** of any origin, such as from algae or the acids extracted from the above mentioned natural animal materials. The preparation of these acids is described in literature. It is preferable to use purified **alginic acids**.

PREPARED the Tetraabutylammonium Salt of **Alginic Acid**:

10 g (28.9 m. Eq.) of a sodium salt of **alginic acid**, corresponding to 2 g. of dry compound, are solubilized in 400 ml of distilled water. The solution is then:

PREPARED the (Partial) Ethyl Ester of **Alginic Acid**--10% of the Carboxy Groups Esterified--20% of the Carboxy Groups Salified:

10 g (28.9 m. Eq.) of the tetraabutylammonium salt of **alginic acid** (prepared from **alginic acid** obtained from *Leishia hyperborea*) are solubilized in 400 ml of DMSO at 25°C. (m. Eq. 11.39 m. Eq.).

PREPARED the (Partial) Ethyl Ester of **Alginic Acid**--30% of the Carboxy Groups Esterified--70% of the Carboxy Groups Salified:

10 g (28.9 m. Eq.) of the tetraabutylammonium salt of **alginic acid** (prepared from **alginic acid** obtained from *Alginium nodosum*) are solubilized in 400 ml of DMSO at 25°C. (m. Eq. 11.15 m. Eq.).

PREPARED the (Partial) Ethyl Ester of **Alginic Acid**--50% of the Carboxy Groups Esterified--50% of the Carboxy Groups Salified:

10 g (28.9 m. Eq.) of the tetraabutylammonium salt of **alginic acid** (prepared from **alginic acid** obtained from *Marine algae* *gyriferia*) are solubilized in 400 ml of DMSO at 25°C. (m. Eq. 11.12 m. Eq.).

PREPARED the (Partial) Ethyl Ester of **Alginic Acid**--70% of the Carboxy Groups Esterified--30% of the Carboxy Groups Salified:

10 g (28.9 m. Eq.) of the tetraabutylammonium salt of **alginic acid** (prepared from **alginic acid** obtained from *Leishia hyperborea*) are solubilized in 400 ml of DMSO at 25°C. (m. Eq. 11.64 g (11.67 m. Eq.).

PREPARED the (Partial) Ethyl Ester of **Alginic Acid**--90% of the Carboxy Groups Esterified--10% of the Carboxy Groups Salified:

10 g (28.9 m. Eq.) of the tetraabutylammonium salt of **alginic acid** (prepared from **alginic acid** obtained from *Marine algae* *gyriferia*) are solubilized in 400 ml of DMSO at 25°C. (m. Eq. 11.18 m. Eq.).

PREPARED the (Partial) Isopropyl Ester of **Alginic Acid**--90% of the Carboxy Groups Esterified--10% of the Carboxy Groups Salified:

10 g (28.9 m. Eq.) of the tetraabutylammonium salt of **alginic acid** (prepared from **alginic acid** obtained from *Alginium nodosum*) are solubilized in 400 ml of DMSO at 25°C. (m. Eq. 11.15 m. Eq.).

PREPARED the (Partial) Isopropyl Ester of **Alginic Acid**--70% of the Carboxy Groups Esterified--30% of the Carboxy Groups Salified:

10 g (28.9 m. Eq.) of the tetraabutylammonium salt of **alginic acid** (prepared from **alginic acid** obtained from *Leishia hyperborea*) are solubilized in 400 ml of DMSO at 25°C. (m. Eq. 11.64 g (11.67 m. Eq.).

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- 1870 10 g (28.3 m. Eq.) of the tetrabutylammonium salt of **alginate** acid prepared from **alginate** acid obtained from *Laminaria* are solubilized in 400 ml of DMSO at 50°C. (D. 4.45 g (16 m. Eq.).
- 1871 Preparation of the Partial Benzyl Ester of **Alginate** Acid--4% of the Carboxylic Groups Esterified--75% of the Carboxylic Groups Salified
- 1872 10 g (28.3 m. Eq.) of the tetrabutylammonium salt of **alginate** acid prepared from **alginate** acid obtained from *Laminaria* are solubilized in 400 ml of DMSO at 50°C. (D. 4.45 g (16 m. Eq.).
- 1873 Preparation of the Partial Benzyl Ester of **Alginate** Acid--10% of the Carboxylic Groups Esterified--75% of the Carboxylic Groups Salified
- 1874 10 g (28.3 m. Eq.) of the tetrabutylammonium salt of **alginate** acid prepared from **alginate** acid obtained from *Laminaria* are solubilized in 400 ml of DMSO at 50°C. (D. 4.45 g (16 m. Eq.).
- 1875 Preparation of the Methyl Ester of **Alginate** Acid
- 1876 10 g (28.3 m. Eq.) of the tetrabutylammonium salt of **alginate** acid prepared from **alginate** acid obtained from *Laminaria* are solubilized in 400 ml of DMSO at 50°C. (D. 4.45 g (16 m. Eq.).
- 1877 Preparation of the Benzyl Ester of **Alginate** Acid
- 1878 10 g (28.3 m. Eq.) of the tetrabutylammonium salt of **alginate** acid prepared from **alginate** acid obtained from *Laminaria* are solubilized in 400 ml of DMSO at 50°C. (D. 4.45 g (16 m. Eq.).
- 1879 Preparation of the Ter-butyl Ester of **Alginate** Acid
- 1880 10 g (28.3 m. Eq.) of the tetrabutylammonium salt of **alginate** acid prepared from **alginate** acid obtained from *Laminaria* are solubilized in 400 ml of DMSO at 50°C. (D. 4.45 g (16 m. Eq.).
- 1881 Preparation of the Isopropyl Ester of **Alginate** Acid
- 1882 10 g (28.3 m. Eq.) of the tetrabutylammonium salt of **alginate** acid prepared from **alginate** acid obtained from *Laminaria* are solubilized in 400 ml of DMSO at 50°C. (D. 4.45 g (16 m. Eq.).
- 1883 Preparation of the Ethyl Ester of **Alginate** Acid
- 1884 10 g (28.3 m. Eq.) of the tetrabutylammonium salt of **alginate** acid prepared from **alginate** acid obtained from *Laminaria* are solubilized in 400 ml of DMSO at 50°C. (D. 4.45 g (16 m. Eq.).
- 1885 Preparation of the Amikacin Salt of **Alginate** Acid: Partially Esterified With Ethanol--75% of Carboxylic Groups Esterified With Ethanol--25% of Carboxylic Groups Salified With Amikacin
- 1886 10 g of a 75% ethyl ester of **alginate** acid and 10 g of salt at 25% (corresponding to 1 m. Eq. of a monomeric unit relative to the non-esterified carboxyl), are.
- 1887 Preparation of Erythromycin Salt of **Alginate** Acid: Partially Esterified With Ethanol--75% of Carboxylic Groups Esterified With Ethanol--25% of Carboxylic Groups Salified With Erythromycin
- 1888 10 g of a 75% ethyl ester of **alginate** acid and 10 g of salt at 25% (corresponding to 1 m. Eq. of a monomeric unit relative to the non-esterified carboxyl), are.
- 1889 Preparation of Streptomycin Salt of **Alginate** Acid: Partially Esterified With Ethanol--75% of Carboxylic Groups Esterified With Ethanol--25% of Carboxylic Groups Salified With Streptomycin
- 1890 10 g of a 75% ethyl ester of **alginate** acid and 10 g of salt at 25% (corresponding to 1 m. Eq. of a monomeric unit relative to the non-esterified carboxyl), are.
- 1891 Preparation of the Partial and Mixed Ethanol and Benzyl Ester of **Alginate** Acid--4% of Carboxylic Groups Esterified With Ethanol--30% of Carboxylic Groups Esterified With Benzyl Ester--4% of Salified Carboxylic Groups. Na
- 1892 10 g of the tetrabutylammonium salt of **alginate** acid prepared from *Laminaria* hyperborea are solubilized in 400 ml of DMSO at 50°C. (D. 4.45 g (16 m. Eq.).
- 1893 Preparation of the Partial Fluorocitric Ester of **Alginate** Acid--10% of Esterified Carboxylic Groups--75% of Carboxylic Groups Salified With Fluorocitric Groups. Na
- 1894 10 g of the tetrabutylammonium salt of **alginate** acid

[illegible]

Alginic Acid-- β -D of Esterified Carboxylic Groups-- β -D of Sulfated Carboxylic Groups-- β -D

1970 Preparation of the Mixed Ethanol and Propylene Glycol Esters of **Alginate** of **Alginate** Acid--2* of Carboxylic Groups Esterified With Ethanol--2* of Carboxylic Groups Esterified With Propylene Glycol

1911 Preparation of the Partial and Mixed Esters of the Carboxylic Acids

4.14 g of the tetrabutylammonium salt of **alginate** prepared from *Marinopyxis pyrifera* corresponding to 1 monomeric unit are solubilized in 100 ml of

187. One object of the present invention is the pharmaceutical preparations containing the 1) more **alginic** acid esters as described above or medicaments resulting from the association of said ester with a pharmacologically active substance as described above, that is medicaments in which the **alginic** ester acts as a vehicle for the active substance.

exploited. In this way, for example, the dosage of **salginate** ester with cortisone may be derived from its content in this same steroid and from its usual dosage in the . . .

DETAILED description of the invention is given in the following examples, that is with a physiologically tolerable pH. Adjustment of the pH, for example in the above mentioned salts of the **alginic** acid esters with a basic active substance, may be effected by suitably regulating the quantities of polysaccharide, its salts and of the basic substance itself. In this way, for example, if the pH is adjusted to

(10) In the present articles according to the invention the **alginic** esters and their salts are mixed with the excipients and used in this field and are for example those already listed for the pharmaceutical preparations. Above all, are used esters, in particular for topical use in which the **alginic** esters or their salts may constitute the active ingredient or in combination with the addition of other therapeutically active substances, such as for example steroids, for example prednisolone, or antibiotics, in accordance with the principles previously reported. In these preparations the **alginic** esters may be associated with a therapeutically active agent, such as prednisolone, or with an ester with an aliphatic chain of 1 to 10 carbon atoms, such as with **aliphatic** alcohols, for example 1-butanol.

these already present the effect is due to the presence of alginic acid in the preparation of the pharmaceutical composition, and not to the presence of free alginic acid or its salts.

- (81) . . . for example, disinfectant substances, such as alkalis, disinfectants, regeneration or antiwound substances, or disinfectant substances, especially perfumes. In this case the **alginic** ester itself may serve as the active ingredient, or the free alginic acid with these same properties, for example in the case of **aliphatic** alcohols or terpenes which is in the case of polyols, or in the case of vegetable substance for instance with the ester of . . .
- (82) partial ester of **alginic** acid with butyl alcohol, . . .
- (83) partial ester of **alginic** acid with glycerol, . . .
- (84) Formulation containing a partial ester of **alginic** acid with ethyl alcohol, of which 10 gr. contains . . .
- (85) partial ester of **alginic** acid with ethyl alcohol, . . .
- (86) Medical Articles Containing the **Alginic** Esters
- (87) The important application of the present invention is in the sanitary and surgical articles already described, the method of their manufacture and their use. The invention therefore includes all the articles similar to those already in the market and all **alginic** acid and containing an **alginic** ester or the salts in place of the free acid or one of its salts, for example insects.

- (88) Completely new surgical and sanitary articles according to the present invention are represented by the esters of **alginic** acid regenerated as such from appropriate organic alcohols and capable of being made into sheet and thread form, thus in films, sheets and threads for use in **surgery**, as skin auxiliaries and substitutes in cases of serious damage to the skin, such as for example following burns, or as suture threads in surgical operations. The invention includes in particular the preparation of a procedure for such articles consisting in the formation of a solution of **alginic** ester or of one of its salts in an appropriate organic solvent, for example a ketone, or ester or an appropriate solvent such as an amide of a carboxylic acid, especially a dialkylamide or of an **aliphatic** acid with between 1 and 8 carbon atoms and deriving from alkyl groups with between 1 and 8 carbon atoms.

- (89) . . . by contact with another organic or aqueous solvent, capable of being mixed with the first solvent and in which the **alginic** ester is not soluble, especially a lower **aliphatic** alcohol, for example ethyl alcohol (wet spinning), or should be solvent with a fairly low boiling point have been used to prepare solutions of **alginic** derivative, in removing such solvent under conditions with a current of gas, and especially under heated nitrogen (dry spinning).

- (90) The threads obtained with the **alginic** acid ester may be used for the preparation of gauzes to be used for the treatment of wounds and in **surgery**. These gauzes have the exceptional advantage of biodegradability in the organism, made possible by the naturally existing enzymes. These enzymes divide the ester into **alginic** acid and the corresponding alcohol, when the **alginic** ester deriving from a therapeutically acceptable alcohol is used, such as ethyl alcohol.

- (91) These gauzes and also the aforesaid threads may also be left inside the organism after **surgery**, being then slowly absorbed after the previously mentioned process of degradation.

- (92) In the preparation of the aforesaid sanitary and surgical articles, it is convenient to add plasticizing substances in order to improve their mechanical characteristics, as in the case of . . .

- (93) of great importance also is the preparation of gauzes and threads with **alginic** esters, solving the problems previously mentioned with their use, which are now very limited, for the same reason as above.

- (94) A further application of the new esters in the field of **medicine** and **surgery** involves the preparation of a wide variety of suture threads such as plates, films, laminas, etc. . . . of a proteinaceous nature, with the exception of unpleasant reactions, such as inflammation or other pathological reactions of **alginic** esters, this preparation of suture threads is . . .

the applications in the medical-surgical field of the ester esters according to the present invention, since they are stable, being expandable material, especially in the form of sponges, and the medication of **wounds** in various types of lesions.

183. The following preparations exemplify the medical uses according to the invention containing the **alginic** esters.
184. Preparation of Bile-Using Esters of **Alginic** Acid.
185. A solution is prepared in dimethylsulfoxide of 1 g. of **alginic** acid with a concentration of 1% in 10 ml. of solvent.
186. Preparation of Thio-Using Esters of **Alginic** Acid.
187. A solution is prepared in dimethylsulfoxide of 1 g. of **alginic** acid with a concentration of 1% in 10 ml. of solvent. This solution is pressed by means of a pump into a mold.
188. Preparation of a Spongy Material Made With **Alginic** Acid.
189. 1 g. of benzyl ester of **alginic** acid in which all the carboxylic groups are esterified, contained for example as described in Example 1, are dissolved in 10 ml. of solvent.
190. Preparation of a Spongy Material Made With **Alginic** Acid Esters.
191. In the manner described in Example 18, it is possible to prepare spongy materials with other **alginic** acid esters. In the case of dimethylsulfoxide it is possible to use, if desired, any other solvent capable of dissolving the ester.
192. Similar compounds, that is, compounds which are in suspension or solution of the solvent used to dissolve **alginic** acid in such a way as to form a gas, such as carbon dioxide, which has the effect of producing a spongy material.

200. What is claimed is:

1. A total, substantially water-insoluble ester of **alginic** acid in which the esterifying alcohol is selected from the group consisting of methyl alcohol, diethyl alcohol, triethyl alcohol, n-butyl alcohol, isobutyl alcohol, dodecyl alcohol, and lauryl alcohol.
2. An **alginic** acid ester according to claim 1, wherein said steroidal alcohol is selected from the group consisting of premarone, hexamethane, cortisone, . . .
3. A method of ophthalmologic treatment which comprises administering, to the corneal surface, an ophthalmologically effective amount of a total ester of **alginic** acid in which the esterifying alcohol moiety is derived from an **aliphatic** alcohol with a maximum of 4 carbon atoms.
4. A process for the preparation of total ester of **alginic** acid which comprises the following steps: a. dissolving a primary ammonium salt of **alginic** acid in an organic solvent, reacting the solubilized salt of **alginic** acid with an esterification agent which is gradually added to the dissolved primary ammonium salt, and b. . .
5. A process according to claim 4, wherein said primary ammonium salt is a lower tetraalkyl ammonium salt of **alginic** acid.
6. A process according to claim 5, wherein said primary ammonium salt is tetrabutylammonium salt of **alginic** acid.
7. A process according to claim 5, which, in step b, comprises adding an organic solvent to precipitate out said ester of **alginic** acid.
8. A process according to claim 5, which further comprises washing and drying said ester of **alginic** acid.
9. A process according to claim 5, wherein said primary ammonium salt of **alginic** acid is prepared by passing an **alginic** acid through a quaternary ammonium ion exchange resin, and recovering said quaternary ammonium salt of **alginic** acid.

10. A radical that an **aliphatic** ACH comprises by the group A and an H group is an optionally substituted **aliphatic** alcohol with a maximum of 34 carbon atoms; or b) an optionally substituted **aliphatic** alcohol with only one benzene ring and in which the **aliphatic** chain has a maximum of 4 carbon atoms; or c) an optionally substituted cycloaliphatic alcohol which is mon- or polycyclic with a maximum of 34 carbon atoms; and d) an optionally substituted **aliphatic**

104,000

168 ANSWER 1 OF 35 USEFUL
 AN 1006104438 USEFUL
 TI Aqueous viscoelastic surfactant solutions for the cleaning of hair and skin
 IN Balzer, Dieter, Haltern, Germany, Federal Republic of
 PA Huls Aktiengesellschaft, Marl, Germany, Federal Republic of Germany
 FI 10 06 08 10 10 10 10
 AI 10 10 06 10 10 10 10 10 10 10
 AB Continuation of Ser. No. US 1005-48172, filed 10 10 10 10 10 10 10 10 10 10
 EXAM Primary Examiner: Gupta, Yogendra; Assistant Examiner: Varde, John R.
 LRE Oblon, Spivak, McClelland, Maier & Neustadt, P.C.
 CLM Number of Claims: 11
 ETL Exemplary Claim: 1
 DRAW 1 Drawing Figures; 1 Drawing Pages
 INVT 11
 CAS INDEXING IS AVAILABLE FOR THIS PATENT.
 AB Aqueous, viscoelastic surfactant solutions for the cleaning of hair and skin which contain:

- (A) from 4 to 25% by weight of an anionic surfactant;
- (B) from 0 to 10% by weight of a betainic surfactant;
- (C) from 0 to 20% by weight of a nonionic surfactant;
- (D) from 0 to 6% by weight of an electrolyte;
- (E) from 0 to 5% by weight of a water-soluble polymer; and
- (F) from 0 to 5% by weight of a further constituent in which the sum of the amounts of (A), (B), and (C) is at least 10% by weight and the sum of the amounts of (C), (D), and (E) is between 2 and 10% by weight, in each case based on the total weight of the aqueous solution, and having a shear modulus, $G_{sub.0}$, between 50 and 500 Pa at temperatures between 0 and 40.degree. C. and a pH of from 4 to 8, and in which the conditions for the identity of the storage modulus, G' , and the loss modulus, G'' , are in the angular frequency range ω from 0.1 and 60 rad.multidot.s.sup.-1, exhibit optimum flow behavior for the intended applications.

CAS INDEXING IS AVAILABLE FOR THIS PATENT.

= 1 kwic

171 ANSWER 1 OF 45 USEFUL
 TI Preparation of lactams from **aliphatic** .alpha.,.omega.-dinitriles
 AB A process for the preparation of five-membered or six-membered ring lactams from **aliphatic** .alpha.,.omega.-dinitriles has been developed. In the process an **aliphatic** .alpha.,.omega.-dinitrile is first converted to an ammonium salt of an .omega.-nitrilecarboxylic acid in aqueous solution using a catalyst having an **aliphatic** nitrilase (EC 3.5.5.3) activity, and a combination of nitrile hydratase (EC 4.3.1.4) and nitrile hydratase (EC 3.5.1.4) activities. The ammonium salt of the corresponding lactam is then formed by hydration in aqueous solution, without isolation of the intermediate .omega.-nitrilecarboxylic acid or .omega.-aminocarboxylic acid. When the **aliphatic** .alpha.,.omega.-dinitrile is also unsymmetrically substituted at the .alpha.-carbon atom, the nitrile hydratase produces the .omega.-nitrilecarboxylic acid ammonium salt resulting from hydrolysis.

SUM This invention relates to a process for the preparation of five-membered
 SEARCHED BY SUSAN HANLEY 104,000

A six-membered ring lactams from **aliphatic** α,ω -dinitriles by a combination of chemical and enzymatic techniques. More particularly, an **aliphatic** α,ω -dinitrile is first converted to the ammonium salt of an ω -nitrilecarboxylic acid in aqueous solution using a catalyst having an **aliphatic** nitrilase EC 4.5.5.7 activity, and a combination of nitrile hydratase EC 4.2.1.24 and amidase EC 3.5.1.4 activities. The ammonium salt of the corresponding lactam is then hydrolyzed in aqueous solution, without isolation of the intermediate ω -nitrilecarboxylic acid or ω -aminoacetic acid. When the **aliphatic** α,ω -dinitrile is also unsymmetrical, substituted at the α -carbon atom, the nitrilase produces the ω -nitrilecarboxylic acid ammonium salt rather than hydrolysis.

SUMM An additional advantage of the enzyme-catalyzed hydrolysis of nitriles over chemical hydrolysis is that, for the hydrolysis of a variety of **aliphatic** or aromatic dinitriles, the enzyme-catalyzed reaction can be highly regioselective, where only one of the two nitrile groups is hydrolyzed.

SUMM The corresponding carboxylic acid ammonium salts have been known for many years, but it is only recently that the use of **aliphatic** nitrilases have been reported. Kobayashi et al. (Enzym. Biochem., 1991, vol. 46, 5587-5590; J. Bacteriology, (1990), vol. 131, 4807-4818) have described an **aliphatic** nitrilase isolated from *Brevibacterium* rhodochrous K12 which catalyzed the hydrolysis of **aliphatic** nitriles to the corresponding carboxylic acid ammonium salts; several **aliphatic** α,ω -dinitriles were also hydrolyzed, and glutaronitrile was converted to 4-cyanobutyric acid ammonium salt with 100% molar conversion using resting cells as catalyst. A nitrilase from *Corynebacterium* testosterone has been isolated which can convert a range of **aliphatic** α,ω -dinitriles to either the corresponding ω -nitrilecarboxylic acid ammonium salt or the corresponding carboxylic acid ammonium salt (Canadian patent application CA 711,616).

SUMM Knowles (Biotechnology Lett., (1994), vol. 16, 41-46) have reported the use of suspensions of *Rhodococcus ruber* NCIMB 11116 having an **aliphatic** nitrilase activity for the hydrolysis of several 1-methylalkylnitriles. Complete conversion of (1-)-2-methylbutyronitrile to 2-methylbutyric acid ammonium salt was obtained, while.

SUMM A combination of two enzymes, nitrile hydratase (EC 4.2.1.24) and amidase, can be also be used to convert **aliphatic** nitriles to the corresponding carboxylic acid ammonium salts in aqueous solution. Here the **aliphatic** nitrile is initially converted to an amide by the nitrile hydratase and then the amide is subsequently converted by the.

SUMM The intermediate formation of 5-cyanovaleic acid using *Brevibacterium* sp. R312 (nitrile hydratase and amidase activity). A. Kerridge et al. (Biorg. Medicinal Chem., (1994), vol. 2, 447-455) report the use of *Brevibacterium* sp. R312 (nitrile hydratase and amidase activity) to hydrolyze prochiral α,ω -dinitrile ammonium salts. European Patent 178,106 B1 (Mar. 31, 1993) discloses selective transformation of one of the cyano groups of an **aliphatic** dinitrile to the corresponding carboxylic acid, ester, ester or thioester using the mononitrilase activity (defined as either nitrilase or amidase).

SUMM No prior art has been found which describes the cyclization of ammonium salts of **aliphatic** ω -nitrilecarboxylic acids in aqueous solution to directly produce the corresponding lactams. In closely related art, U.S. Pat. No. 4,329,493 describes.

SUMM The various acid ammonium salts. However, no prior art has been found which describes the cyclization of ammonium salts of **aliphatic** ω -aminoacetic acids under the reaction conditions of the present invention (i.e., in an aqueous solution containing an excess of added ammonium hydroxide) to produce the corresponding lactams. In closely related art, the cyclization of **aliphatic** ω -aminoacetic acids (but not their ammonium salts) to the corresponding lactams under a variety of reaction conditions has been reported. The reaction is carried out at 10 to 40 degree C. The synthesis of five-, six- and seven-membered ring lactams by cyclodehydration of **aliphatic** ω -aminoacetic acids in

[illegible][illegible]

10000 In addition to producing lactams from aliphatic
10001 alipha , omega -dinitriles, N-methyl-lactams are
10002 prepared by the
10003 substitution of methylamine for ammonia in the
10004 synthesis of the amine.
10005

NOTE: Two microorganisms have been isolated for use as a bacterial catalyst for the conversion of **aliphatic** alpha, beta-unsaturated ketones to the corresponding gamma-nitrilecarboxylic acids and/or oxo acids. *EW 56746* and *Comamonas testosteroni* 5-100-4. ATCC 56744.

Aliphatic alpha,omega-Dinitrile Hydrolysis Reactions

An aqueous solution containing the ammonium salt of an **aliphatic** ω -nitrilecarboxylic acid is prepared by mixing the corresponding **aliphatic** α,ω -nitrile with an aqueous suspension of the appropriate enzyme catalyst as identified in part A above. Whole microbial cells may be used as catalyst without any pretreatment. Alternatively, they may be immobilized in a polymer matrix (e.g., **alginate** beads), polyacrylamide gel (PAG) particles) or on an insoluble solid support (e.g., **celite**) to facilitate recovery and reuse.

Some of the **aliphatic** alpha.,omega.-dinitriles such as starting material in the present invention are only moderately water soluble. Their solubility is also dependent on . . .

SUMMARY: The final concentration of **aliphatic** ω -nitrile- α -carboxylic acid ammonium salt in the product mixture at time of conversion of the α -**aliphatic** ω -nitrile may range from 0.10 M to the solubility limit of the **aliphatic** ω -nitrile- α -carboxylic acid ammonium salt. Typically, the concentration of the ω -nitrile- α -carboxylic acid ammonium salt ranged from 0.10 M to 3.0 M.

EXAMPLE 1 Catalytic hydrogenation is a preferred method for preparing an **aliphatic** amine from an **aliphatic** nitrile. In the present invention, the ω -**aliphatic**-aminocarboxylic Acid produced during the hydrogenation cyclizes to the corresponding five- or six- or six-membered ring. . . . ω -**aliphatic**-nitrilecarboxylic Acid is prepared by neutralization and filtration of the aperiodic product. . . . The ω -**aliphatic**-nitrilecarboxylic Acid is first mixed with 100 parts of ammonia hydroxide and water to produce a solution which is then heated from one to four . . .

1870 In the following examples, which serve to further illustrate the invention and not to limit it, the % recovery of **aliphatic** α,ω -dinitriles and the % yields of the four lysis products formed during the microbial hydrolysis reactions were based on the initial . . .

10-44,1

10-44,1 ANSWER 1 OF 35 USEPATFULL

AB 10-44,1 10-44,1 USEPATFULL

11 A protective cosmetic particulate gel delivery system and method of preparing complex gel particles

12 Leblond, Pascal, Jean-Pierre Tillet, France

13 Ding, Li, Jean-Pierre Tillet, France

14 Paragel Products S.A., France, non-U.S. applicant

15 US 5,861,000 10-44,1 10-44,1

16 US 10-44,1-10-44,1 10-44,1 10-44,1

17 Utility

18 EXAMINER: Primary Examiner: Bawa, Raj

19 LEXIS: Bandal and Morfey

20 CLASS: Number of Claims: 14

21 ECL: Exemplary Claim: 1

22 DRAW: 5 Drawing Figures; 1 Drawing Page 30

23 INVENT 10-44,1

24 INDEXING IS AVAILABLE FOR THIS PATENT.

AB A protective cosmetic particulate gel delivery system and method of applying applied active agent employs an agar gel and a restraining polymer to retain the active agent in the gel. The particles have an average particle diameter of at least 0.05 mm while the restraining polymer has a molecular weight of at least 50,000 daltons and retention groups to bind the active agent. The restraining polymer can be selected from the group consisting of polyquaternium 24, laurdimonium hydroxyethylcellulose, cecidimonium hydroxyethylcellulose, steardimonium hydroxyethylcellulose, quaternary ammonium substituted water-soluble polysaccharides, allyl quaternary celluloses and polyamides having or provided with retention groups to retain the active agent. The gel particles of the invention are manually crushable on the skin to increase the surface area of the gel particle matrix and expose the restraining polymer to the skin or other body surface for release of the active agent. The delivery system can be incorporated in multiphase cosmetic formulations such as gels, creams and lotions.

25 INDEXING IS AVAILABLE FOR THIS PATENT.

10-44,1

10-44,1 ANSWER 1 OF 45 USEPATFULL

AB A process for the preparation of five-membered or six-membered ring lactams from **aliphatic** α,ω -dinitriles has been developed. In the process an **aliphatic** α,ω -dinitrile is first converted to an ammonium salt of an ω -nitrilecarboxylic acid in aqueous solution using a catalyst having an **aliphatic** nitrilase (EC 3.5.5.7) activity, and a combination of nitrile hydratase (EC 4.2.1.24) and nitrile hydratase (EC 3.5.1.4) activities. The ammonium salt of the ω -nitrilecarboxylic acid is then hydrogenated in aqueous solution, without isolation of the intermediate ω -nitrilecarboxylic acid or ω -nitrilecarboxylic acid. When the **aliphatic** α,ω -dinitrile is also unsymmetrically substituted at the α -carbon atom, the nitrilase produces the ω -nitrilecarboxylic acid ammonium salt resulting from hydrolysis.

10-44,1 This invention relates to a process for the preparation of five-membered or six-membered ring lactams from **aliphatic** α,ω -dinitriles by a combination of chemical and chemical techniques. More particularly, an **aliphatic** α,ω -dinitrile is first converted to an ammonium salt of an ω -nitrilecarboxylic acid in aqueous solution using a catalyst having an **aliphatic** nitrilase (EC 3.5.5.7) activity, and a combination of nitrile hydratase (EC 4.2.1.24) and nitrile hydratase (EC 3.5.1.4) activities. The ammonium salt of the ω -nitrilecarboxylic acid is then hydrogenated in aqueous solution, without isolation of the intermediate

SEARCHED BY SUSAN HANLEY 10-44,1

10-44,1

omega-nitrile-carboxylic acid or omega-amino-carboxylic acid. When the **aliphatic** alpha,omega-dinitrile is acid hydrolyzed, the hydrolysis product is substituted with the alpha-amino group, the nitrile group is converted to the omega-nitrile-carboxylic acid and ammonium salt and the nitrile is hydrolyzed.

SUMMARY: An additional advantage of the enzyme-catalyzed hydrolysis of nitriles over chemical hydrolysis is that, for the hydrolysis of a variety of **aliphatic** aromatic nitriles, the enzyme-catalyzed reaction can be highly regiospecific, where only one of the two nitrile groups is hydrolyzed.

SUMMARY: Corresponding carboxylic acid ammonium salts have not been known for many years, but it is only recently that the use of **aliphatic** nitrilases have been reported. Horiyama et al., J. Biochem., 1961, vol. 48, 887-889; J. Bacteriology, 1961, vol. 121, 1-7-4-11, have described an **aliphatic** nitrilase isolated from *Brevibacterium* which catalyzed the hydrolysis of **aliphatic** nitriles to the corresponding carboxylic acid ammonium salts; several **aliphatic** alpha,omega-dinitriles were also hydrolyzed, and glutar nitrile was converted to 4-cyanobutyric acid ammonium salt with 100% conversion using resting cells as catalyst. Nitrilase from *Streptococcus lactococcus* has been isolated which can convert a range of **aliphatic** alpha,omega-dinitriles to either the corresponding carboxylic acid ammonium salt or the corresponding carboxylic acid ammonium salt. Canadian patent application 744,111, 1961.

SUMMARY: Knowles (Biotechnology Lett., (1964), vol. 16, 41-46) have reported the use of suspensions of *Brevibacterium* in a 0.1M NaHCO₃ solution having an **aliphatic** nitrilase activity for the hydrolysis of several 2-methylalkylnitriles. Complete conversion of 2-methylbutylnitrile to 2-methylbutyric acid ammonium salt was obtained, while.

SUMMARY: A combination of two enzymes, nitrile hydratase and amidase, can be used to convert **aliphatic** nitriles to the corresponding carboxylic acid ammonium salts in a single reaction. Here the **aliphatic** nitrile is initially converted to an amide by the nitrile hydratase and then the amide is subsequently converted by the amidase.

SUMMARY: The intermediate formation of 5-cyanopentanoic acid using *Brevibacterium* sp. R312 (nitrile hydratase and amidase activity). A. Herdige et al. (Bierg. Medicinal Chem., (1964), vol. 1, 447-455, report the use of *Brevibacterium* sp. R312 (nitrile hydratase and amidase activity) to hydrolyze prochiral. European Patent 106,106 B 1 (Mar. 31, 1963, discloses selective transformation of one of the cyano groups of an **aliphatic** dinitrile to the corresponding carboxylic acid, amide, ester or lactam using the mononitrilase activity (defined as either nitrilase or nitrile hydratase).

SUMMARY: No prior art has been found which describes the cyclization of ammonium salts of **aliphatic** omega-nitrilecarboxylic acids in aqueous solution to directly produce the corresponding lactams. In closely related art, U.S. Pat. No. 4,329,498 describes.

SUMMARY: No prior art has been found which describes the cyclization of ammonium salts of **aliphatic** omega-amino-carboxylic acids under the reaction conditions of the present invention (i.e., in aqueous solution containing an excess of added ammonium hydroxide to produce the corresponding lactams. In closely related art, the cyclization of

aliphatic omega-amino-carboxylic acids (but not the ammonium salts) to the corresponding lactams under a variety of reaction conditions has been reported. . . . steam at from 100 degree C. to 400 degree C. The synthesis of five-, six- and seven-membered ring lactams by cyclodehydration of **aliphatic** omega-amino-carboxylic acids in alkaline or acidic gel in toluene, and with continuous removal of the water produced during the reaction, has.

SUMMARY: No prior art has been found which describes the cyclization of ammonium salts of **aliphatic** omega-nitrilecarboxylic acids in aqueous solution containing methylamine to directly produce the corresponding N-methyl lactams. In closely related art, 1,5-dimethyl-2-pyrrolidone was.

SUMMARY: Mixtures of products which are not easily separated. A significant advance would be a process for the preparation of an **aliphatic** alpha,omega-dinitrile to the corresponding lactam.

1. N-nitrilase in aqueous solution, in high yields and with high selectivity, with little expense.
2. A process for the preparation of five-membered lactams from **aliphatic** α,ω -dinitriles, having the steps:
- a. reacting an **aliphatic** α,ω -dinitrile with an aqueous reaction mixture with an enzyme catalyst, characterized by either
 - a. **aliphatic** nitrilase activity, or
 - b. a combination of nitrile hydratase and amidase activities, whereby the **aliphatic** α,ω -dinitrile is converted to the corresponding ω -nitrilecarboxylic acid ammonium salt;
 - b. using a whole cell catalyst to select for a relatively high nitrilase activity or nitrile hydratase activity capable of following the conversion of **aliphatic** α,ω -dinitrile to the corresponding ω -nitrilecarboxylic acid ammonium salt. The whole cell catalyst to be treated is characterized by two factors:
 - a. purified enzymes, or
 - b. purified enzymes on a support.
 - c. Micro-organisms which are characterized by an **aliphatic** nitrilase activity, or which in the process are *Acidovorax facilis* 7L-PF-15 (ATCC 5874), *Acidovorax facilis* 7L-PF-17 (ATCC 58748), and *Acidovorax* sp.
3. A process to prepare lactams from **aliphatic** α,ω -dinitriles in high yields has been developed which includes a combination of enzymatic and chemical reactions. In cases where the:
- a. Conversion of an **aliphatic** α,ω -dinitrile to the corresponding ω -nitrilecarboxylic acid ammonium salt in high yield and with high regioselectivity.
 - b. The first step of this process is the conversion of an **aliphatic** α,ω -dinitrile to the corresponding ω -nitrilecarboxylic acid ammonium salt, using an enzyme catalyst. The enzyme catalyst has either a nitrilase activity, or a combination of two enzyme activities, nitrile hydratase (NHase) and amidase, where the **aliphatic** α,ω -dinitrile is initially converted to the corresponding ω -nitrilecarboxylic acid ammonium salt by the nitrile hydratase, and then the ω -nitrilecarboxylic acid ammonium salt is subsequently converted to the corresponding lactam.
4. A novel micro-organism *Acidovorax facilis* 72W (ATCC 5874) has been isolated from soil samples which had been exposed to **aliphatic** nitriles or dinitriles, and which could utilize 2-ethylsuccinonitrile as a nitrogen source. When used as a microbial whole-cell catalyst for:
- a. There are currently no non-enzymatic methods for the selective hydrolysis of only one nitrile group of an **aliphatic** dinitrile to either an amide group or a carboxylic acid group to complete conversion of the dinitrile. If such a:
 - b. *7L-PF-17* (ATCC 58748), do not require heat-treatment of the cells prior to use as catalyst for the hydrolysis of an **aliphatic** α,ω -dinitrile to the corresponding ω -nitrilecarboxylic acid ammonium salt of a ω -nitrilecarboxylic acid. A comparison of the yields of 4-CFA and 3-methylglutaric acid.
 - c. heat-treated *Acidovorax facilis* 72W (ATCC 5874) is used as a catalyst for the hydrolysis of aqueous solutions of the unsubstituted **aliphatic** α,ω -dinitriles succinonitrile (SCN, 1.25 M) or glutaronitrile (GLN, 1.5 M), the corresponding ω -nitrilecarboxylic acid ammonium salts 3-cyanopropionic acid (3-CFA) or 3-cyanoglutaric acid (3-CGA).
5. More than 30 different microbial cultures isolated from soil samples which had been exposed to **aliphatic** nitriles or dinitriles, and which could grow on various nitriles or amides as nitrogen source, were screened for high selectivity.
6. High yields by the direct hydrogenation of the ω -nitrilecarboxylic acid ammonium salt product mixture produced by the enzyme-catalyzed hydrolysis of **aliphatic** α,ω -dinitriles in aqueous solution. This method does not require the isolation of the ω -nitrilecarboxylic acid ammonium salt from the product.
7. After producing an aqueous product mixture containing the ammonium salt of a ω -nitrilecarboxylic acid from an **aliphatic** α,ω -dinitrile by using an enzyme catalyst, removal of the enzyme catalyst and reaction of the resulting ammonium salt.

- ... and **aliphatic** nitriles in the presence of the isolated catalyst was expected to produce a mixture containing an **aliphatic** ω -nitrilecarboxylic acid. (See, e.g., eq. 4: **STh** The use of an excess of ω -nitrile in the hydrogenation of a nitrile. . . .
- 0000 In addition to producing systems from **aliphatic** ω -nitriles, ω -nitriles, N-methyl- ω -nitriles, and the corresponding substitution of methylamine for ammonia in the synthesis of aqueous solutions of the ammonium salt. . . .
- 0000 Two microorganisms have been isolated for use as natural catalysts for the conversion of **aliphatic** ω -nitriles to the corresponding ω -nitrilecarboxylic acids: *Acidovorax* facilis 72W (ATCC 55746) and *Acidovorax* testostearii 8-MW-1 (ATCC 55747).
- 0000 **Aliphatic** ω -nitrilecarboxylic Acid Hydrolysis Reaction
- 0000 An aqueous solution containing the ammonium salt of an **aliphatic** ω -nitrilecarboxylic acid is prepared by mixing the corresponding **aliphatic** ω -nitrile with an aqueous suspension of the appropriate enzyme catalyst as identified in part A above. Whole microbial cells may be used as catalyst without any pretreatment. Alternatively, the cells may be immobilized in a polymer matrix (e.g., alginate-chitosan, polyacrylamide gel, IAS particles) or on an insoluble solid support (e.g., celite) to facilitate recovery and reuse. . . .
- 0000 Some of the **aliphatic** ω -nitrilecarboxylic acids starting material in the present invention are slightly water-soluble. Their solubility is also dependent on the . . .
- 0000 The final concentration of **aliphatic** ω -nitrilecarboxylic acid ammonium salt in the product mixture at complete conversion of the ω -nitrile may range from 0.001 M to the solubility limit of the **aliphatic** ω -nitrilecarboxylic acid ammonium salt. Typically, the concentration of the ω -nitrilecarboxylic acid ammonium salt ranged from 0.10 M to 1.0 M. . . .
- 0000 Catalytic hydrogenation is a preferred method for preparing an **aliphatic** amine from an **aliphatic** nitrile. In the present invention, the ω -aminocarboxylic acid produced during the hydrogenation cyclizes to the corresponding five- or six- or ω -membered ring. . . . ω -nitrilecarboxylic acid is prepared by centrifugation and filtration of the aqueous product mixture produced by the enzymatic hydrolysis of the corresponding **aliphatic** ω -nitrile. ω -nitrile is first mixed with concentrated ammonium hydroxide and water to produce a solution which is then . . .
- 0000 In the following examples, which serve to further illustrate the invention and not to limit it, the % recovery of **aliphatic** ω -nitriles and the % yields of the hydrolysis products formed during the microbial hydrolysis reaction are based on the initial. . . .
- 0000 What is claimed is:
1. Isolated microorganisms characterized by an **aliphatic** nitrilase activity and selected from the group consisting of *Acidovorax* facilis 72W (ATCC 55746), *Acidovorax* facilis 72-P-1 (ATCC 55747), *Acidovorax*. . . .

1000000 124 1

12. ANSWER 1 OF 6 HCAPLUS DAYLIGHT L. A.M.

13. 12-11-1997 HCAPLUS

14. 12-11-1997

15. Alkyl polyglycoside compositions and emulsions containing them with increased whiteness

16. Belcher, Catherine; Amalio, Mantel; Lopez, Antonio Miguel; Nelly, Milne, Alain

17. Societe d'Exploitation de Produits pour les Industries Chimiques, Fr.

18. 1997, Tokyo Tokyo, Jpn., 6 pp.

19. INCIEN: CHYXAP

20. Patent

21. Japanese

22. INT 1

PATENT NO.	INVENTOR	DATE	APPLICATION NO.	DATE
JP 1-16411	AL	1-16411	JP 1993-16411	1993-01-17
FR 17468	AL	1-16411	FR 1994-11611	1994-01-17
EP 16411	AL	1-16411	EP 1993-40111	1993-09-16

R: AD, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, NL, NO, SE, MD, PT, IE, SI, LT, LV, FI, RO

1997 FR 199-11611 1997-01-17

23. MARPAT 16411:141697

24. The comps., useful as emulsifiers for cosmetic emulsions, etc., comprise 5-6% (a) alkyl polyglycoside mixts. contg. (a) 40-50% by wt. of R1O(G1)k (R1 = C16-18 linear or branched **aliph.** group; G1 = **saccharide** residue; k = 1-5) and R2O(G2)l (R2, G2, and l have the same definition as R1, G1, and l, resp.) and 5-7% by wt. of R3O(G3)m (R3 = C20-22 linear or branched **aliph.** group; G3 = **saccharide** residue; m = 1-5) and R4O(G4)n (R4, G4, and n have the same definition as R3, G3, and m, resp.) and 40-55% by wt. of R5OH (R5 = C14-22 linear or branched **aliph.** group, preferably **aliph.** R1-R4). Also claimed are emulsions contg. (a) 5-6% by wt. of (a) phase, (b) phase, (c) phase, and the comps. as main emulsifiers. A mixture contg. 45.5% C16 and C18 alcs. and 14.5% C20 and C22 alcs. was treated with glucose at alc./glucose mol ratio 6:1 in the presence of acid at 100 degree. for 5 h to give alkyl polyglycoside mixt. The mixt. was used for emulsification of isobutyl isinonate, sweet almond oil, and isobutyl acetate to give stable white emulsions.

[illegible]

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01-  ALL RIGHTS RESERVED.  COPYRIGHT © 1995
02-  BY THE UNIVERSITY OF MICHIGAN
03-  10770-44785
04-  Translated as a computer program containing data, compiled
05-  Taketa, Minoru; Murakawa, Toshiyuki; Taketata, Masahiro; Taketa, Toshiyuki
06-  Fall 1994, Japan
07-  1994, Nikai Tokkyo. Kogyo, 7 pp.
08-  CODEN: JPHXXAF
09-  Patent
10-  Japanese
11-  Fall 1994

```

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
1	JP 624160	AL	1997.016	JP 1996-28697	1996.8.6
2	PARENT 1,711,041-85				

AB The compn., with good moisture retention and good solubility, comprise
aliph. and a sp. multivalent alcoh. in saccharides and
salts or derivs. of $\text{R1COOCH(R2)COOP(OH)} \cdot \text{R1CO} = \text{C6-22}$ and n up;
 $\text{R2} = \text{Me, H; p} = 1-50$. Thus, a compn. was prepd. fr. a mixt. of hard
tall w- and soft oil mixt. 3, 95% EtOH 10, glycerol 7, sugar 10 and
BHTA 1.5% and balanced water.

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36	37	38	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54	55	56	57	58	59	60	61	62	63	64	65	66	67	68	69	70	71	72	73	74	75	76	77	78	79	80	81	82	83	84	85	86	87	88	89	90	91	92	93	94	95	96	97	98	99	100
1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36	37	38	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54	55	56	57	58	59	60	61	62	63	64	65	66	67	68	69	70	71	72	73	74	75	76	77	78	79	80	81	82	83	84	85	86	87	88	89	90	91	92	93	94	95	96	97	98	99	100

[illegible]

	PATENT N.°	FIND	DATE	APPLICATION N.°	AGE
11	WO 96/464	A1	1996/11/10	WO 1998/02464	1998/01/29
	WO 96/464	EP, ES, FR, GB, DE, FI, HU, JP, KR, NL, PL, PT, SE, SF, UA			
	WO 96/464	BE, CH, CZ, DK, EG, ES, GB, GR, IE, IL, IT, MX, NL, PT, SE			
	US 546719	A	1996/11/10	US 1998-02137	1998/01/29
	EP 0441496	A1	1998/11/30	EP 1998-04496	1998/01/29
	EP 46101	A1	1998/01/10	EP 1998-01101	1998/01/29
	EP 46101	DE, ES, FR, GB, IT			
	EP 46101	IL	1998/01/10	EP 1998-01101	1998/01/29
1991	US 1991-02138		1991/01/29		
	WO 1993-02464		1993/01/29		

AB In the title process, a slurry of a hydrous saccharide source (e.g., hexose monohydrate) in a first portion of an aliph. sol. (exp. 22-22) is added to a second portion of the slurry obtained at an elevated temp. and reduced pressure to form a mixt. with a reduced water content, an acid catalyst is added, and the aliph. sol. is reacted with the saccharide source to form the glycoside.

1974, 1975, 1976

1. POLYMER OF 2, 3-DIACETYL-4-HYDROXY-5-NOR

2. 1, 2, 3, 4, 5-PENTACETYL-6-NOR

3. 1, 2, 3, 4, 5-PENTACETYL-6-NOR

4. 1, 2, 3, 4, 5-PENTACETYL-6-NOR

5. 1, 2, 3, 4, 5-PENTACETYL-6-NOR

6. 1, 2, 3, 4, 5-PENTACETYL-6-NOR

7. 1, 2, 3, 4, 5-PENTACETYL-6-NOR

8. 1, 2, 3, 4, 5-PENTACETYL-6-NOR

9. 1, 2, 3, 4, 5-PENTACETYL-6-NOR

10. 1, 2, 3, 4, 5-PENTACETYL-6-NOR

11. 1, 2, 3, 4, 5-PENTACETYL-6-NOR

12. 1, 2, 3, 4, 5-PENTACETYL-6-NOR

PATENT NO.	FILED	DATE	APPLICATION NO.	FILED
13	13	1976-11-1	13	1976-11-1
14	14	1976-11-1	14	1976-11-1
15	15	1976-11-1	15	1976-11-1
16	16	1976-11-1	16	1976-11-1
17	17	1976-11-1	17	1976-11-1
18	18	1976-11-1	18	1976-11-1
19	19	1976-11-1	19	1976-11-1
20	20	1976-11-1	20	1976-11-1
21	21	1976-11-1	21	1976-11-1
22	22	1976-11-1	22	1976-11-1
23	23	1976-11-1	23	1976-11-1
24	24	1976-11-1	24	1976-11-1
25	25	1976-11-1	25	1976-11-1
26	26	1976-11-1	26	1976-11-1
27	27	1976-11-1	27	1976-11-1
28	28	1976-11-1	28	1976-11-1
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30	30	1976-11-1	30	1976-11-1
31	31	1976-11-1	31	1976-11-1
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33	33	1976-11-1	33	1976-11-1
34	34	1976-11-1	34	1976-11-1
35	35	1976-11-1	35	1976-11-1
36	36	1976-11-1	36	1976-11-1
37	37	1976-11-1	37	1976-11-1
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39	39	1976-11-1	39	1976-11-1
40	40	1976-11-1	40	1976-11-1
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43	43	1976-11-1	43	1976-11-1
44	44	1976-11-1	44	1976-11-1
45	45	1976-11-1	45	1976-11-1
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47	47	1976-11-1	47	1976-11-1
48	48	1976-11-1	48	1976-11-1
49	49	1976-11-1	49	1976-11-1
50	50	1976-11-1	50	1976-11-1
51	51	1976-11-1	51	1976-11-1
52	52	1976-11-1	52	1976-11-1
53	53	1976-11-1	53	1976-11-1
54	54	1976-11-1	54	1976-11-1
55	55	1976-11-1	55	1976-11-1
56	56	1976-11-1	56	1976-11-1
57	57	1976-11-1	57	1976-11-1
58	58	1976-11-1	58	1976-11-1
59	59	1976-11-1	59	1976-11-1
60	60	1976-11-1	60	1976-11-1
61	61	1976-11-1	61	1976-11-1
62	62	1976-11-1	62	1976-11-1
63	63	1976-11-1	63	1976-11-1
64	64	1976-11-1	64	1976-11-1
65	65	1976-11-1	65	1976-11-1
66	66	1976-11-1	66	1976-11-1
67	67	1976-11-1	67	1976-11-1
68	68	1976-11-1	68	1976-11-1
69	69	1976-11-1	69	1976-11-1
70	70	1976-11-1	70	1976-11-1
71	71	1976-11-1	71	1976-11-1
72	72	1976-11-1	72	1976-11-1
73	73	1976-11-1	73	1976-11-1
74	74	1976-11-1	74	1976-11-1
75	75	1976-11-1	75	1976-11-1
76	76	1976-11-1	76	1976-11-1
77	77	1976-11-1	77	1976-11-1
78	78	1976-11-1	78	1976-11-1
79	79	1976-11-1	79	1976-11-1
80	80	1976-11-1	80	1976-11-1
81	81	1976-11-1	81	1976-11-1
82	82	1976-11-1	82	1976-11-1
83	83	1976-11-1	83	1976-11-1
84	84	1976-11-1	84	1976-11-1
85	85	1976-11-1	85	1976-11-1
86	86	1976-11-1	86	1976-11-1
87	87	1976-11-1	87	1976-11-1
88	88	1976-11-1	88	1976-11-1
89	89	1976-11-1	89	1976-11-1
90	90	1976-11-1	90	1976-11-1
91	91	1976-11-1	91	1976-11-1
92	92	1976-11-1	92	1976-11-1
93	93	1976-11-1	93	1976-11-1
94	94	1976-11-1	94	1976-11-1
95	95	1976-11-1	95	1976-11-1
96	96	1976-11-1	96	1976-11-1
97	97	1976-11-1	97	1976-11-1
98	98	1976-11-1	98	1976-11-1
99	99	1976-11-1	99	1976-11-1
100	100	1976-11-1	100	1976-11-1

13. 1, 2, 3, 4, 5-PENTACETYL-6-NOR

14. 1, 2, 3, 4, 5-PENTACETYL-6-NOR

15. Polysaccharide matrices are esterified with polyunsaturated aliph. acids which are prostaglandin precursors in vivo. The esters can be administered orally, parenterally, or topically for inhibition or induction of abortion. For example, 1, 2, 3, 4, 5-pentacetyl-6-nor was treated with imidazole [1, 2, 3, 4, 5] in the presence of N,N-dicyclohexylcarbodiimide to give 1, 2, 3, 4, 5-pentacetyl-6-nor and imidazole [1, 2, 3, 4, 5]. Dextran [1, 2, 3, 4, 5] was esterified with the latter compound to give a polymer [1, 2, 3, 4, 5] of dextran 22.5% dithio-1, 2, 3, 4, 5-pentacetyl-6-nor groups.

[illegible]

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RM      1114-04-4  HOARLUS
TN      1-2-14-10000000, N,N,N-Triethyl-, fluoride 001.  OF INDEX NAME

```

[illegible]

9005-38-3, Sodium alginate 9005-38-3D, Sodium
alginate, acryloyl derivs.
SL: PAB Properties
hydrographically assocg. alginate derivs. surface reaction
properties of their mixed eq. solns. with applicable related
substances

ALL INFORMATION CONTAINED HEREIN IS UNCLASSIFIED
DATE 06-18-2009 BY 60322 UCBAW/SAT

◆ ◆ ◆ *Journal of Management Education* 32(1) 1-12 ◆ ◆ ◆

[illegible]

*** CONFIDENTIAL ***

[illegible]

Table 1. *Continued*

10. *Journal of the American Medical Association*, 1990; 263: 1033-1037.

01 ANSWER L F - HANAPUS COPYRIGHT L AM
02 L F L F - HANAPUS
03 L F L F -
04 Non-identifiable wound dressings containing
05 polysaccharide derivatives
06 Maudsland, Philippe; Dellacherie, Edith; Hubert,
07 Patrick; Houzelle, Marie Christine; Pelletier,
08 Sophie
09 Les Laboratoires Biotinier, Fr.
10 Dep. Roger Thady, Paris, 6 pp.
11 GEN: PHYLAE

Figure 1 is a schematic representation of the experimental design. It is divided into two main sections: 'Pretest' and 'Main Experiment'. The 'Pretest' section includes a 'Pretest' box with a 'Pretest' label and a 'Pretest' box with a 'Pretest' label. The 'Main Experiment' section includes a 'Main Experiment' box with a 'Main Experiment' label and a 'Main Experiment' box with a 'Main Experiment' label.

1997, 1998, 1999, 2000, 2001, 2002, 2003, 2004, 2005, 2006, 2007, 2008, 2009, 2010, 2011, 2012, 2013, 2014, 2015, 2016, 2017, 2018, 2019, 2020, 2021, 2022, 2023, 2024, 2025, 2026, 2027, 2028, 2029, 2030, 2031, 2032, 2033, 2034, 2035, 2036, 2037, 2038, 2039, 2040, 2041, 2042, 2043, 2044, 2045, 2046, 2047, 2048, 2049, 2050, 2051, 2052, 2053, 2054, 2055, 2056, 2057, 2058, 2059, 2060, 2061, 2062, 2063, 2064, 2065, 2066, 2067, 2068, 2069, 2070, 2071, 2072, 2073, 2074, 2075, 2076, 2077, 2078, 2079, 2080, 2081, 2082, 2083, 2084, 2085, 2086, 2087, 2088, 2089, 2090, 2091, 2092, 2093, 2094, 2095, 2096, 2097, 2098, 2099, 2100, 2101, 2102, 2103, 2104, 2105, 2106, 2107, 2108, 2109, 2110, 2111, 2112, 2113, 2114, 2115, 2116, 2117, 2118, 2119, 2120, 2121, 2122, 2123, 2124, 2125, 2126, 2127, 2128, 2129, 2130, 2131, 2132, 2133, 2134, 2135, 2136, 2137, 2138, 2139, 2140, 2141, 2142, 2143, 2144, 2145, 2146, 2147, 2148, 2149, 2150, 2151, 2152, 2153, 2154, 2155, 2156, 2157, 2158, 2159, 2160, 2161, 2162, 2163, 2164, 2165, 2166, 2167, 2168, 2169, 2170, 2171, 2172, 2173, 2174, 2175, 2176, 2177, 2178, 2179, 2180, 2181, 2182, 2183, 2184, 2185, 2186, 2187, 2188, 2189, 2190, 2191, 2192, 2193, 2194, 2195, 2196, 2197, 2198, 2199, 2200, 2201, 2202, 2203, 2204, 2205, 2206, 2207, 2208, 2209, 2210, 2211, 2212, 2213, 2214, 2215, 2216, 2217, 2218, 2219, 2220, 2221, 2222, 2223, 2224, 2225, 2226, 2227, 2228, 2229, 2230, 2231, 2232, 2233, 2234, 2235, 2236, 2237, 2238, 2239, 2240, 2241, 2242, 2243, 2244, 2245, 2246, 2247, 2248, 2249, 2250, 2251, 2252, 2253, 2254, 2255, 2256, 2257, 2258, 2259, 2260, 2261, 2262, 2263, 2264, 2265, 2266, 2267, 2268, 2269, 2270, 2271, 2272, 2273, 2274, 2275, 2276, 2277, 2278, 2279, 2280, 2281, 2282, 2283, 2284, 2285, 2286, 2287, 2288, 2289, 2290, 2291, 2292, 2293, 2294, 2295, 2296, 2297, 2298, 2299, 2300, 2301, 2302, 2303, 2304, 2305, 2306, 2307, 2308, 2309, 2310, 2311, 2312, 2313, 2314, 2315, 2316, 2317, 2318, 2319, 2320, 2321, 2322, 2323, 2324, 2325, 2326, 2327, 2328, 2329, 2330, 2331, 2332, 2333, 2334, 2335, 2336, 2337, 2338, 2339, 2340, 2341, 2342, 2343, 2344, 2345, 2346, 2347, 2348, 2349, 2350, 2351, 2352, 2353, 2354, 2355, 2356, 2357, 2358, 2359, 2360, 2361, 2362, 2363, 2364, 2365, 2366, 2367, 2368, 2369, 2370, 2371, 2372, 2373, 2374, 2375, 2376, 2377, 2378, 2379, 2380, 2381, 2382, 2383, 2384, 2385, 2386, 2387, 2388, 2389, 2390, 2391, 2392, 2393, 2394, 2395, 2396, 2397, 2398, 2399, 2400, 2401, 2402, 2403, 2404, 2405, 2406, 2407, 2408, 2409, 2410, 2411, 2412, 2413, 2414, 2415, 2416, 2417, 2418, 2419, 2420, 2421, 2422, 2423, 2424, 2425, 2426, 2427, 2428, 2429, 2430, 2431, 2432, 2433, 2434, 2435, 2436, 2437, 2438, 2439, 2440, 2441, 2442, 2443, 2444, 2445, 2446, 2447, 2448, 2449, 2450, 2451, 2452, 2453, 2454, 2455, 2456, 2457, 2458, 2459, 2460, 2461, 2462, 2463, 2464, 2465, 2466, 2467, 2468, 2469, 2470, 2471, 2472, 2473, 2474, 2475, 2476, 2477, 2478, 2479, 2480, 2481, 2482, 2483, 2484, 2485, 2486, 2487, 2488, 2489, 2490, 2491, 2492, 2493, 2494, 2495, 2496, 2497, 2498, 2499, 2500, 2501, 2502, 2503, 2504, 2505, 2506, 2507, 2508, 2509, 2510, 2511, 2512, 2513, 2514, 2515, 2516, 2517, 2518, 2519, 2520, 2521, 2522, 2523, 2524, 2525, 2526, 2527, 2528, 2529, 2530, 2531, 2532, 2533, 2534, 2535, 2536, 2537, 2538, 2539, 2540, 2541, 2542, 2543, 2544, 2545, 2546, 2547, 2548, 2549, 2550, 2551, 2552, 2553, 2554, 2555, 2556, 2557, 2558, 2559, 2560, 2561, 2562, 2563, 2564, 2565, 2566, 2567, 2568, 2569, 2570, 2571, 2572, 2573, 2574, 2575, 2576, 2577, 2578, 2579, 2580, 2581, 2582, 2583, 2584, 2585, 2586, 2587, 2588, 2589, 2590, 2591, 2592, 2593, 2594, 2595, 2596, 2597, 2598, 2599, 2600, 2601, 2602, 2603, 2604, 2605, 2606, 2607, 2608, 2609, 2610, 2611, 2612, 2613, 2614, 2615, 2616, 2617, 2618, 2619, 2620, 2621, 2622, 2623, 2624, 2625, 2626, 2627, 2628, 2629, 2630, 2631, 2632, 2633, 2634, 2635, 2636, 2637, 2638, 2639, 2640, 2641, 2642, 2643, 2644, 2645, 2646, 2647, 2648, 2649, 2650, 2651, 2652, 2653, 2654, 2655, 2656, 2657, 2658, 2659, 2660, 2661, 2662, 2663, 2664, 2665, 2666, 2667, 2668, 2669, 2670, 2671, 2672, 2673, 2674, 2675, 2676, 2677, 2678, 26

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
(1)	JP 57111646	A1	01/01/2011	JP 2009-183007	08/27/2009
	FR 2841677	A1	10/06/2014	FR 28397488	10/06/2014
	RU 2642416	A1	01/06/2014	RU 2009-40416	10/06/2009

AB The invention provides a gelolytic-reversible wound dressing comprising a polysaccharide, esp. alginate, having aliphatic chains.

9005-32-7D, Aluminic acid, salts, reaction products, aliph. alkyl carbox., and aliph. amines
 RL: THU (Therapeutic use); BIOL (Biological study); TBC (Uses: self); reversible wound dressings cont.
 polysaccharide derivs. having aliph. side chains

10 0-0-0 HOURS

*** STRUCTURE DIAGRAM IS NOT AVAILABLE ***

1005-32-7D, Alginate

1005-32-7D, Alginate
 RN 1005-32-7D HCAPLUS
 UN 1005-32-7D
 IT Structural characterization of pharmaceutical fibers of alginate salts by
 confocal Raman spectral imaging
 AU Chappé, Louis; Dupontier, Philippe; Maingault, Philippe;
 Laroche, Pierre
 LW Lab. Chim. Anal., Fac. Pharm., Univ. de Tours, Tours, Fr.
 AB Polym. SPIE-Int. Soc. Opt. Eng., 1999, 3648-Biomedical Applications of
 Raman Spectroscopy, 47-54
 CLEEN: 961023; ISSN: 1077-766X
 AB SPIE-The International Society for Optical Engineering
 IT Journal
 LA English
 AB Alginates are natl. polymers extrd. from brown algae and are well known for
 their gelling and viscosifying properties and biodegradability. Calcium
 alginate gels are used for spinning fibers which are particularly
 interesting as wound dressings since they exhibit
 antithrombotic and healing effects. Alginates are copolymers of
 alpha-D-glucuronate (G) and beta-D-mannuronate (M). The M/G ratio and
 the monomer sequence are determinant for their physico-chem. properties
 but are difficult to control since they are variably depending on algae
 family, region of origin and season of their harvest. For rapid and
 non-destructive structural characterization of some fibers of alginates
 we used confocal Raman spectral imaging with a micron-scale spatial
 resolution. The characteristic Raman features were analyzed and correlated
 with the M/G ratio. For the fibers with a detd. av. M/G ratio, the
 spectral images appeared rather homogeneous. On the other hand, the Raman
 relative intensities were found to be dependent on the fiber orientation of
 the fiber and laser polarization. We concluded that, with a micron-scale
 resolution, the fiber samples are well homogeneous and the polymer chains are
 mainly ordered with domination of parallel stacking of the M-M blocks.
 The approach is actually being applied to study the alginate fibers at the
 tissular level.
 IT 9005-32-7D, Alginate acid, salts 9005-35-0, Calcium
 alginate 56687-62-8D, beta-D-Mannuronic acid, salts
 copolymers 56688-68-7D, alpha-L-Galacturonic acid, salts
 copolymers
 AB: ANT Analyte; PEE Physical, engineering or mechanical process; PEE
 Properties; THU Therapeutic use; ANST Analytical study; BIOL
 Biological study; PPOC (Process); USES (Uses)
 IT Structural characterization of pharmaceutical fibers of alginate salts
 by confocal Raman spectral imaging
 RN 1005-32-7D HCAPLUS
 UN Alginate acid (ECI, DCI) (CA INDEX NAME)

*** STRUCTURE DIAGRAM IS NOT AVAILABLE ***

RN 1005-35-0 HCAPLUS
 UN Alginate acid, calcium salt (DCI) (CA INDEX NAME)

*** STRUCTURE DIAGRAM IS NOT AVAILABLE ***

RN 1005-62-8 HCAPLUS
 UN beta-D-Mannuronic acid (DCI) (CA INDEX NAME)

1005-68-7D, Alginate

RN 1005-68-7D HCAPLUS
 UN alpha-L-Galacturonic acid (DCI) (CA INDEX NAME)

1005-68-7D, Alginate

RECEIVED
 REAR WET BY JUDAN HANLEY 1004-1-1

[illegible]

1. *Journal of the American Medical Association*, 1977; 237: 1001-1002.

4. *Chlorophyll a*, *b*, and *total chlorophyll* = 100, 100, 100 respectively

1. The first group of people who are not in the labor force are those who are not in the labor force because they are not in the labor force.

1. *Journal of Polymer Science: Polymer Chemistry Edition*, **14**, 1055 (1976).

	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36	37	38	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54	55	56	57	58	59	60	61	62	63	64	65	66	67	68	69	70	71	72	73	74	75	76	77	78	79	80	81	82	83	84	85	86	87	88	89	90	91	92	93	94	95	96	97	98	99	100
1	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36	37	38	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54	55	56	57	58	59	60	61	62	63	64	65	66	67	68	69	70	71	72	73	74	75	76	77	78	79	80	81	82	83	84	85	86	87	88	89	90	91	92	93	94	95	96	97	98	99	100

Figure 1. The effect of the concentration of the *Agrobacterium* suspension on the transformation efficiency of *Agrobacterium* strains. The *Agrobacterium* strains were grown in the YEA medium for 24 h at 28 °C. The cell concentration of the strains was adjusted to 10⁸ cells/ml. The cell suspension was mixed with the plant tissue and the transformation efficiency was determined. The results were expressed as the mean ± SD of three independent experiments. The asterisks indicate the significant difference between the strains at the same concentration of the cell suspension.

[illegible]

1984-1985

- 11- ANSWER 4 3-4 BIOACTIVE POLYMERIZATION OF ALG
 12- 1984-1985 HCAPLUS
 13- 1984-1985
 14- **alginate-alginate** combination for the preparation of
 of materials investigation of the behavior in aqueous solution
 15- **Lechner, Sandra; Payan, Elisabeth; Lapicque, Francis; Le, Nathalie;**
Hubert, Patrick; Miller, Sylvaine; Netter, Patrick; Lapicque,
Francis
 16- 88-1-4, Faculté de Médecine, UMS 2561 CNRS, FR 2561, Laboratoire de
 Pharmacie, Physiopathologie et Pharmacologie Appliquées, UMR Nancy
 1, Vandœuvre les Nancy, 54505, Fr.
 17- **Biophys. Biophys. Acta** 1983, 142, 1, 1-8-1984
 ISSN: 0005-2728; ISSN: 0005-2728
 18- Elsevier Science B.V.
 19- English
 20- English
 21- English
 22- English
 23- With the aim of producing a biomaterial for surgical applications, the
alginate-hyaluronate system was investigated. Different techniques
 were used to assess the existence of polymer interactions in aqueous
 solution. **Alginate** was obtained from algae and **hyaluronate**
 was purified from rooster comb. Viscometry measures including the
 capillary technique or the Couette flow, together with π -A investigations,
 evidenced the moderate significance of interactions between the
polysaccharides in dil. solns. In addn., the case of a combined
 soln. and contr. **alginate** was approached by visl.
 measurements in the flow mode; the behavior of the polymer associations
 appeared as a compromise between those of individual
polysaccharides.
 24- **9005-38-3, Sodium alginate 9067-32-7, Sodium**
hyaluronate
 25- FFE (Properties); THU (Therapeutic use); BIOL (Biological study); USES
 (Uses)
 26- **hyaluronate-alginate** combination for prepn. of biomaterials
 27- 90-1-3 HCAPLUS
 28- **Alginic acid, sodium salt (ACI, DCI)** (CA INDEX NAME)
 *** STRUCTURE DIAGRAM IS NOT AVAILABLE ***
 29- 90-67-32-7 HCAPLUS
 30- **Hyaluronic acid, sodium salt (DCI)** (CA INDEX NAME)
 *** STRUCTURE DIAGRAM IS NOT AVAILABLE ***
 31- 90-1-3
 32- 90-1-3
 33- **Edmeier, R; Pharm Res** 1989, V6, P413 HCAPLUS
 34- **Mezard, R; Biopolymers** 1970, V9, P399 HCAPLUS
 35- **Flangochat, A; Carbohydr Res** 1996, V184, P85 HCAPLUS
 36- **Garcia, A; J Control Release** 1996, V40, P129 HCAPLUS
 37- **Grant, S; FEBS Lett** 1973, V32, P105 HCAPLUS
 38- CITATIONS AVAILABLE IN THE RE FORMAT

1000-00-0000-0000

01- ANSWER: E- HANDELING: PHYSICIST: AAS
 02- 1000-00-0000-0000
 03- 1000-00-0000-0000
 04- The chemical study of aqueous solutions of hydrophobic poly-saccharide
 derivatives of propylene glycol alginate
 05- English, D.; Hubert, P.; Lape, A.; Houzelle, M.C.;
 06- Staph, A.; Maréchal, P.; Dellacherie, E.
 07- 1983, LEWIS-ENGLIS, NANCY, 14-11, 11.
 08- Rev. Inst. Fr. Lett. 1987, 11-11, 11-11-11
 09- INDEX: 11-11-11
 10- French
 11- English
 12- English
 13- We report our results in the physicochem. study in aqueous soln. of
 amphiphilic polysaccharide deriva. obtained by attachment of
 long chain alcohols (C₁₂H₂₅, n = 7, 10, 14) onto propylene glycol
 alginate, a partially-esterified deriv. of sodium alginate
 14- 111-86-4D, n-Octylamine, reaction products with propylene
 alginate 124-22-1D, n-Dodecylamine, reaction products
 with propylene alginate 2016-42-4D, n-Tetradecylamine,
 reaction products with propylene alginate 9005-37-2D,
 Propylene glycol alginate, reaction products with alcohols
 15- FEP: Physical, engineering or chemical process; 16- Properties;
 17- 111-86-4D
 18- chem. study of aq. solns. of hydrophobically-act. deriva. of
 propylene glycol alginate
 19- 111-86-4 HCAELUS
 20- 1-Octanamine (SCI) (CA INDEX NAME)

H₂O: CH₂-CH₂-Me

01- 114-11-1 HCAELUS
 02- 1-Dodecanamine (SCI) (CA INDEX NAME)

H₂O: CH₂-CH₂-Me

01- 116-41-4 HCAELUS
 02- 1-Tetradecanamine (SCI) (CA INDEX NAME)

H₂O: CH₂-CH₂-Me

01- 117-47-1 HCAELUS
 02- Aliphatic acid, ester with 1,1'-propanediol (SCI, SCI) (CA INDEX NAME)

CM 1

11- 11-31-1
 CMF Unspecified
 SCI RMS, MAN

*** STRUCTURE DIAGRAM IS NOT AVAILABLE ***

CM 1

11- 11-31-1
 CMF Unspecified

WIDE - 14, 15

B
WIDE - 14, 15

SEARCHED BY JUDITH HANLEY - 14, 15

14, 15

WIDE 12-164, 1-1

SEARCHED BY JUAN RAMIREZ 1-14-11

PAGE 11

1994-01-01 11:17

11- ANSWER 1 OF 1 HCAPLUS COPYRIGHT 2000 ACS
 AN 1-6170-77-6 HCAPLUS
 IN 1191:64:16
 TI Divalent coupling of a short polyether to sodium **alginate**: synthesis and characterization of the resulting copolymer derivative
 AU Jurek, Marie-Christine; Leclerc, Christine; Hubert, Patrick; Dellacherie, Edith
 AD Lab. Chim. Ind., Univ. Nancy 1, Vandœuvre les Nancy, France, FR
 JO Polym. Prepr. 1991, 32(4), 667-70
 COUNTRY: FRANCE; ISSN: 144-0017
 DT Journal
 LA English
 AB A divalent sodium **alginate**-polyoxyethylene model 1-6170-77-6 was prepared by reductive amination of aldehyde sodium **alginate**, in order to obtain a polymer with amphiphilic properties. Characterization of this series was carried out by NMR spectroscopy, viscosity measurements and low-angle laser light scattering. The data obtained suggested a limited capacity of the polymer structure to expand, resulting from intramolecular self-association and formation of hydrophobic domains.
 DT 9005-38-3DP, Sodium **alginate**, reaction products with
 alpha-amino, omega-tertiaryalkoxy tetraoxyethylene 86770-77-6DP,
 reaction products with sodium **alginate**
 RD: RSEP: Preparation
 AB amphiphilic, synthesis and limited expansion capability of
 AN 1-6170-77-6 HCAPLUS
 CN Alginic acid, sodium salt (-O⁻, OCl⁻) (CA INDEX NAME)
 *** STRUCTURE DIAGRAM IS NOT AVAILABLE ***
 AN 1-6170-77-6 HCAPLUS
 CN 1,5,7,11,14-Pentaoxahexadecan-16-amine, 1-phenyl- (CA INDEX NAME)

AB 1-A

H₂N-CH₂-CH₂-CH₂-CH₂-CH₂-O-CH₂-CH₂-O-CH₂-CH₂-O-CH₂-CH₂-O-CH₂-CH₂-O-

AB 1-B

-CH₂-Ph

1994-01-01 11:17

11- ANSWER 1 OF 1 HCAPLUS COPYRIGHT 2000 ACS
 AN 1-6170-77-6 HCAPLUS
 IN 1191:64:16
 TI FT-Raman spectroscopy and Raman multispectral imaging study of alginate wound dressings in vitro and in tissues
 AU Jurek, Marie-Christine; Carpentier, Philippe; Maingault, Philippe; Durrieu, Pierre
 AD Laboratoire de Chimie Analytique, Université de Tours, Tours, 37200, FR
 JO Spectrosc. Biol. Med.: New Dir., Eur. Conf., 9th, 1991, 17-51v.
 ED: J. A. Brice, J. A. Puppels, G. W. Ott, G. W. Ott; Publisher: Kluwer Academic Publishers, Dordrecht, Neth.
 COUNTRY: FRANCE
 DT Conference
 LA English
 AB The structural organization of the calcium alginate dressings were studied by Raman multispectral imaging with red laser excitation.
 RE: WT: 11
 RE

SEARCHED BY JUDAN HANLEY 1-1-95

1-1-95

= 1 018

FILE 'HOME' ENTERED AT 13:37:11 ON 26 SEP 2000

FILE 'HCAPLUS' ENTERED AT 13:37:14 ON 26 SEP 2000

L1 7 S MAINSULT P1 AU
 L2 134 S DELLACHERIE L1 AU
 L3 487 S HUBERT P1 AU
 L4 4 S NOVELLE M1 AU
 L5 341 S FELLETTIER S1 AU
 L6 1 S L1 AND L2 AND L3 AND L4 AND L5
 L7 971 S L1-5
 L8 18 S L7 AND ALGINAT?
 L9 106906 S ALIPHATIC
 L10 1 S L6 AND L9
 L11 101 S L9-SACCHARID?
 L12 1 S L11 AND L9
 L13 37813 S IPOLYSACCH?
 L14 6 S L9 AND L13
 L15 4 S L7 AND (WOUND?
 L16 9 S L14 OR L18
 SELECT RN L16 1-9

in ventor search

FILE 'REGISTRY' ENTERED AT 13:43:32 ON 26 SEP 2000

L17 20 S E1-2)

FILE 'HCAPLUS' ENTERED AT 13:43:45 ON 26 SEP 2000

L18 8 S L16 AND L17
 L19 1 S L16 NOT L18
 L20 17412 S PALGIN?
 L21 1537 S PALGIN? (L1THU/RL
 L22 154 S L21 AND (WOUND OR DRESSING)
 L23 6 S L22 AND L11
 L24 6 S L23 NOT L16
 L25 366738 S AMMONIUM? OR NH4
 L26 101 S L21 AND L25
 L27 33 S L13 AND L26
 L28 30 S HYDROWIGEL
 L29 3 S L29 AND L20
 L30 15671 S ALGINIC OR ALGINATE
 L31 3 S L28 AND L30
 L32 4 S L28 AND L16 AND L9
 L33 179 S GEL AND L16 AND L30
 L34 4 S L33 AND L9
 L35 692425 S CARBOXYL? OR ESTER?
 L36 1683 S L35 AND (L30 OR L30)
 L37 227 S L36 AND L16
 L38 69 S L37 AND L16
 L39 3 S L38 AND L9

8 cites w/ 20 cps displayed
 1 cite no cps

FILE 'STNGUIDE' ENTERED AT 14:00:07 ON 26 SEP 2000

FILE 'HCAPLUS' ENTERED AT 14:09:35 ON 26 SEP 2000

L40 453894 S GEL(MASOL OR PREVERS?
 L41 3 S L38 AND L40
 L42 69 S L38 NOT L16
 L43 12 S L42 AND FYM1999
 L44 57 S L42 NOT L43

57 cites

FILE 'MEDLINE, BIOSIS, USPATFULL' ENTERED AT 14:13:16 ON 26 SEP 2000

L45 41341 S PALGIN? OR ALGINAT? OR ALGINIC
 L46 513515 S NH4 OR AMMONIUM?
 L47 648546 S GEL
 L48 1186341 S SOL OR SOLUTION
 L49 30290 S L47 OR L48
 L50 57634 S RHEOL
 L51 44445 S POLYSACCHARID?
 L52 114503 S LONG-CHAIN? OR ALIPHATIC?
 L53 21741 S L45 AND L51 OR ALPHYL

SEARCHED BY SUSAN HANLEY 9-24-00

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134 14-48 S L63 AND L46
 135 1048 S L64 AND L49 OR L51
 136 1088 S L45 AND L55
 137 617 S L61 F ALKYL OR L61
 138 1470 S L48 F ALKYL OR L61
 139 8-41 S L67 OR L68
 140 437 S L69 F L46
 141 180 S L60 F L45 OR L49
 142 7 S L45 AND L61
 143 71 DUE REM L61 DUPLICATES REMOVED
 144 38 S L63 AND MEDICINE OR SURGEY OR WOUND S L61 S L63
 145 7 S L63 AND MACROMOLECULE
 146 1 S L61 AND MACROMOLECULE
 147 18 S L61 AND L49 OR L51
 148 35 S L66 OR L67 *cites 1 1/2 only*
 149 11 S L51 AND L68
 150 64 S L61 AND L63
 151 19 S L51 AND L64
 152 45 S L69 OR L71 *selected citations*